

1                   Effects of Oxygen-Containing Salts on the  
2                   Detection of Organic Biomarkers on Mars and in  
3                   Terrestrial Analogue Soils

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17   **Running title:** Effects of Oxyanion Salts on Biomarker Detection

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19   **Index Terms and Key Words**

20   Mars, organics, pyrolysis-GC-MS, Atacama, perchlorate, detection

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23    **Abstract**

24    The detection of chlorinated hydrocarbons by Curiosity on Mars has been  
25    attributed to the presence of unidentified indigenous organic matter. Similarly,  
26    oxychlorines on Earth have been proposed to be responsible for the apparent lack  
27    of organics in the Atacama Desert. The presence of perchlorate ( $\text{ClO}_4^-$ ) poses a  
28    unique challenge to the measurement of organic matter due to the oxidizing power  
29    of oxychlorines during commonly used pyrolysis-gas chromatography-mass  
30    spectrometry (py-GC-MS) methods. Here, we show that perchlorates and other  
31    oxyanion salts inhibit the detection of organic compounds, but that removing  
32    these problematic species prior to pyrolysis by using an optimal sample extraction  
33    duration and suitable ratios of water to sample mass enables analysis. We have  
34    characterized leached and unleached samples containing perchlorates from the  
35    Atacama Desert and have found that after leaching, the py-GC-MS  
36    chromatograms of the dried mineral residues show identifiable biomarkers  
37    associated with indigenous cyanobacteria. Samples which were pyrolyzed  
38    without leaching showed no detectable organic matter other than background  
39    siloxane and very weak or no trace of detectable polychlorinated benzenes. Dried  
40    sample residues remaining after leaching, the mineral matrix and water-insoluble  
41    organic matter, showed a strong organic response in all cases when analyzed by  
42    py-GC-MS. These residues are most likely the product of the pyrolysis of water  
43    insoluble organics originally present in the samples. In addition, our results imply  
44    that previous soil analyses which contained high levels of oxyanions and  
45    concluded that organics were either not present, or at extremely low levels, should  
46    be re-examined.

## 47    **1. Introduction**

48        Organic compounds should be present on Mars, if for no other reason than  
49    ongoing chondritic infall (Flynn 1996). Several were detected during the first  
50    attempt in the mid-1970s by the two Viking Landers, but they were all attributed  
51    to terrestrial contamination (Biemann *et al.* 1977). Subsequently, the martian  
52    surface was viewed for several decades as containing one or more unidentified  
53    oxidants that destroyed organics (Zent and McKay 1994). In 2008, the Phoenix  
54    Mars lander analyzed the martian soil at its northern latitude landing site. The  
55    *Thermal and Evolved Gas Analyzer* (TEGA), a thermal desorption system feeding  
56    a mass spectrometer detected only CO<sub>2</sub> released from carbonates and no organic  
57    molecules (Boynton *et al.* 2009). The *Wet Chemistry Laboratory* (WCL)  
58    (Kounaves *et al.* 2009) performed the first wet chemical analyses of three martian  
59    soil samples, one from the surface and two from 5 cm depth. For all samples, the  
60    soil/water mixture was found to be slightly alkaline (pH  $\approx$  7.7), with an average  
61    electrical conductivity  $\approx$  1.4 mS/cm, and containing several salt species, Ca<sup>2+</sup>,  
62    Mg<sup>2+</sup>, Na<sup>+</sup>, K<sup>+</sup>, Cl<sup>-</sup>, and SO<sub>4</sub><sup>2-</sup> (Kounaves *et al.* 2010a; 2010b). However, the most  
63    significant finding was that the dominant soluble form of chlorine at the Phoenix  
64    site was perchlorate (ClO<sub>4</sub><sup>-</sup>), at a concentration of  $\sim$  0.6 wt% ClO<sub>4</sub><sup>-</sup> in the soil  
65    (Hecht *et al.* 2009; Kounaves *et al.* 2010a) and most likely present as a 60%  
66    Ca(ClO<sub>4</sub>)<sub>2</sub>, 40% Mg(ClO<sub>4</sub>)<sub>2</sub> mixture (Kounaves *et al.* 2014b). Evidence for ClO<sub>4</sub><sup>-</sup>  
67    at similar levels was subsequently also reported in the Rocknest aeolian deposit

68 by the *Sample Analysis at Mars* (SAM) instrument on the *Mars Science*  
69 *Laboratory* (MSL) rover Curiosity (Glavin *et al.* 2013; Leshin *et al.* 2013) and in  
70 the martian meteorite EETA 79001 (Kounaves *et al.* 2014a).

71 In addition to the  $\text{ClO}_4^-$ , the SAM instrument has also detected chloro-,  
72 dichloro-, and trichloro-methane ( $\text{CH}_3\text{Cl}_x$ ,  $x=1-3$ ) (Ming *et al.* 2014),  
73 chlorobenzene ( $\text{C}_6\text{H}_5\text{Cl}$ ) and  $\text{C}_2$ - $\text{C}_4$  dichloroalkanes (Freissinet *et al.* 2015; Glavin  
74 *et al.* 2013), and more recently naphthalene which was suggested to be indicative  
75 of complex macromolecular organic matter (Eigenbrode *et al.* 2018).

76 Interestingly, the Viking lander's pyrolysis-gas chromatograph-mass  
77 spectrometer (py-GC-MS) had also detected traces of  $\text{CH}_3\text{Cl}$  and  $\text{CH}_2\text{Cl}_2$ . This  
78 detection had led to the recent suggestion that this may have been a result of  $\text{ClO}_4^-$   
79 in the soil chlorinating either terrestrial organic contaminants or, if present,  
80 indigenous organics (Navarro-González *et al.* 2010).

81 Laboratory experiments under simulated martian conditions have shown that  
82 oxychlorines (including  $\text{ClO}_4^-$  and  $\text{ClO}_3^-$ ) can be produced on Cl-bearing mineral  
83 surfaces via UV driven oxidation (Carrier and Kounaves 2015). This supports the  
84 hypothesis that not only  $\text{ClO}_4^-$ , but also  $\text{ClO}_3^-$ , are very likely ubiquitous on Mars.  
85 Recently,  $\text{ClO}_4^-$  was detected and quantified in regolith and rock samples from the  
86 Moon, and two chondrite meteorites (Jackson *et al.* 2015). Although present at  
87 orders of magnitude less than Mars, these findings suggest its ubiquitous presence  
88 in the solar system. Thus, analyses of organic compounds on Mars and elsewhere

89 in the solar system must consider the possible presence of oxychlorines,  
90 especially  $\text{ClO}_3^-$  and  $\text{ClO}_4^-$ .

91       Although perchlorate is stable under ambient terrestrial and martian  
92 conditions, it becomes a powerful oxidant when heated to temperatures higher  
93 than about 150°C. Onset of  $\text{O}_2$  evolution from oxychlorine decomposition in the  
94 John Klein mudstone in Gale Crater was  $\sim 150^\circ\text{C}$  (Ming *et al.* 2014). Thus, the  
95 pyrolysis heating used prior to GC-MS (or any other analytical combination using  
96 similarly high temperatures) could destroy or transform a large fraction of the  
97 organics via combustion by oxychlorines and prevent direct identification of  
98 indigenous organic molecules, including potential biomarkers. One exception to  
99 this would be organics trapped inside minerals or refractory materials stable at  
100 temperatures above 600°C that may not be affected by the presence of  
101 oxychlorine compounds that decompose at lower temperatures. For analyses in  
102 the temperature range of 150-600°C, this process would have thwarted the  
103 detection of organics during the thermal-based analyses on Viking and Phoenix,  
104 and thus the previous interpretation of these results as an absence of organics is  
105 not definitive. The MSL-SAM instrument also detected chlorinated  
106 hydrocarbons, specifically chlorobenzene and  $\text{C}_2$  to  $\text{C}_4$  dichloroalkanes.  
107 However, it is not clear whether these may have been the reaction products of  
108 martian chlorine and terrestrial contaminants, or indigenous organic molecules  
109 either already chlorinated or thermally decomposed in the presence of these

110 oxychlorine compounds (Freissinet *et al.* 2015). It has also been shown that even  
111 though organics trapped in magnesium sulfate may undergo some oxidation and  
112 sulfuration during py-GC-MS, they also protect organics from oxidation by  
113 calcium perchlorate (François *et al.* 2016).

114       Terrestrial-based analyses of organics prior to the recognition of the effects of  
115 perchlorate on organics may have also been misinterpreted. For example, in the  
116 core of the Atacama Desert (Chile), one of the driest places on Earth, organic  
117 carbon content and biomass fall to the lowest levels found anywhere on Earth (ca.  
118  $10^3$  to  $10^5$  cells/g of soil), concurrent with an increase in mean residence time of  
119 soil organic carbon to tens of thousands of years (Ewing *et al.* 2008). At the same  
120 time, salt crusts formed by deliquescence provide a minimalistic habitat for  
121 survival under extremely dry conditions, colonized by communities composed of  
122 extremely resistant *Chroococcidiopsis* morphospecies of cyanobacteria and  
123 associated heterotrophic bacteria (Wierzchos *et al.* 2012).

124       The Atacama Desert represents a reasonable martian analogue because, as  
125 may also be the case for Mars, these salt rich crusts and soils are the only  
126 remaining habitat available for life (Davila and Schulze-Makuch 2016). Thus,  
127 when searching for biomarkers on Mars one objective should be to analyze the  
128 evaporitic, chloride-bearing rocks that may have provided (or still provide) a  
129 suitable habitat enabling photosynthetic activity and adaptation to a harsh UV and  
130 hyper-arid environment (Vitek *et al.* 2010). Since these regions on the martian

131 surface will most likely also contain high levels of perchlorates (Clark and  
132 Kounaves 2016) and sulfates (Gaillard *et al.* 2013), the methodology to detect  
133 organic compounds on Mars has to be adjusted for their presence, and existing  
134 analyses may need reconsideration in the same way as the Atacama results, taking  
135 the hydrologic and atmospheric history of Mars into account.

136         It is now clear that the use of py-GC-MS in the presence of oxychlorines  
137 poses an enormous challenge, and requires the development of methodologies that  
138 can mitigate the effect of oxychlorines and other oxyanions during pyrolysis.  
139 Recent suggestions have included the use of sacrificial molecules (Kenig *et al.*  
140 2016), supercritical-CO<sub>2</sub> extraction (McCaig *et al.* 2016), laser desorption-mass  
141 spectrometry, which is currently part of *Mars Organic Molecule Analyser*  
142 (MOMA) on the ESA/ExoMars 2020 mission (Li *et al.* 2015), a polymeric anion  
143 exchange resin (von Kiparski *et al.* 2013), and interpretation via gases produced  
144 during whole rock pyrolysis (Sephton *et al.* 2014). More comprehensive  
145 development is possible for future missions, as currently these techniques have  
146 either not been successfully demonstrated or introduce high risk for planetary  
147 missions.

148         Here, we demonstrate the utility of removing these problematic species prior  
149 to pyrolysis by using an optimal sample extraction duration and suitable ratios of  
150 water to sample mass. We have characterized leached and unleached samples  
151 containing perchlorates and sulfates from the Atacama Desert. We have found

152 that after leaching, the py-GC-MS chromatograms of the dried mineral residues  
153 show identifiable biomarkers associated with cyanobacteria present in this  
154 environment (Biller *et al.* 2015).

## 155 **2. Site and Sample Description**

### 156 *2.1 Site description*

157 The samples used in this study were collected from the Antofagasta Region  
158 of the Atacama Desert (Chile) and which have already been extensively  
159 characterized in terms of soil habitability, biomolecules indicative of potentially  
160 active cells, situ replication rates of genomes, and microbial community patterns  
161 specific to soil parameters and depths (Schulze-Makuch *et al.* 2018). The  
162 Atacama's arid to hyperarid environment renders it inhospitable, not only to  
163 macroscopic primary producers, but also limits the few microbial photosynthetic  
164 communities found there (Warren-Rhodes *et al.* 2006), mainly specialized  
165 endolithic cyanobacteria in salt outcrops (Wierzchos *et al.* 2006) and subsurface  
166 microbial communities (Fletcher *et al.* 2011). More importantly, in addition to  
167 widespread areas covered with evaporitic halite, these arid soils contain high  
168 levels of oxyanions, especially perchlorates, sulfates, and nitrates (Perez-Fodich *et*  
169 *al.* 2014). The Atacama Desert is considered to be a good analogue site for Mars'  
170 environment during its late Hesperian and early Amazonian epochs (Fairen *et al.*  
171 2010).

172 The soil and rock samples were collected from three sites, all approximately  
173 55-85 km east of Antofagasta: (1) Red Sands (RS: 24° 6.041' S, 70° 7.733' W); (2)  
174 Lomas Bayas (LB: 23° 23.596' S, 69° 36.237' W); and (3) Yungay (YU: 24°  
175 5.296' S, 69° 59.673' W). The LB and YU sites are considered to be part of the  
176 hyperarid core of the Atacama Desert, while RS is part of the arid region. A map  
177 and images of the sampling sites are available as supplementary information to  
178 Schulze-Makuch *et al.* 2018.

## 179 2.2 Sample collection descriptions

180 At the RS and LB sites the top layer of desert pavement rocks was removed  
181 and a surface sample was taken from 0-5 cm depth. At the YU site, samples were  
182 collected from two faces of the pit. One set of samples were collected every 5 cm  
183 to a depth of 100 cm to be used for the leaching ratio analysis. A subset of these  
184 high resolution samples collected at depths of 25, 60, 75, and 85 cm were for the  
185 efficiency analyses. The other set of samples from an adjacent face were  
186 collected at depths of 0-5, 20-30, 50, and 100 cm to be used for the organic  
187 compound analyses. The RS, LB, and YU (50 cm depth) samples were selected  
188 to represent "salt poor", "sulfate rich", and "oxyanion/salt rich" environments,  
189 respectively, all appearing to contain low levels of organics when studied using  
190 standard py-GC-MS of raw samples, consistent with previous work in this area  
191 (Ewing *et al.* 2008).

192 A rock sample was collected from the surface ~ 2 km from the YU site. Its  
193 exposed surfaces contained partially visible grayish-green streaks. A sample of  
194 the relatively pure endolithic cyanobacteria was physically removed from a layer  
195 on the rock for analysis. Previous analyses have shown these layers to be  
196 composed of endolithic photosynthetic bacteria belonging to the genus  
197 *Chroococcidiopsis*, an extremely resistant morphospecies of cyanobacteria  
198 (Wierzchos *et al.* 2006). The cyanobacteria sample was analyzed based on the  
199 assumption that organics in the other Atacama soil samples may have significant  
200 contributions from the growth and death of such endolithic organisms.

### 201 **3. Methods**

202 The soil was sieved to remove pebbles < 2 mm and homogenized using a  
203 mortar and pestle to < 75  $\mu\text{m}$ . For each sample, 1.0 g of homogenized soil was  
204 leached with Nanopure® 18.2 M $\Omega$ -cm water for 1 hour with gentle stirring. The  
205 leachate was then removed and filtered using a 0.2  $\mu\text{m}$  PTFE filter. Samples  
206 prepared for  $\text{ClO}_4^-$  leaching ratio analysis were leached at a 1:5, 1:10, and 1:25  
207 (w/v) ratio. Based on the homogeneity of the leaching ratio analyses, the samples  
208 for the  $\text{ClO}_4^-$  leaching efficacy and the inorganic anion and cation analyses were  
209 only leached at a 1:10 (w/v) ratio and diluted to a conductivity of ~ 50  $\mu\text{S}/\text{cm}$ .  
210 The air dried residue from the 1:10 leached soil was then used for further analyses  
211 by py-GC-MS for organic content. The time between leaching and py-GC-MS  
212 was about 2 days. Samples were stored in clean vials in light-proof boxes in the

213 climate-controlled lab. It was assumed there would be contamination from the  
214 various plastic storage containers, but none was ever observed either because  
215 there wasn't sufficient transfer or the perchlorate destroyed all of it.

216 Each single leach was analyzed in triplicate for soluble inorganic content by  
217 ion chromatography (IC) using a Dionex ICS-2000 Reagent Free™ IC equipped  
218 with suppressed electrical conductivity detection. For separation and detection of  
219 perchlorate we used a 100  $\mu$ L injection volume, a Dionex AS16 Ionpac analytical  
220 column with 35 mM potassium hydroxide (KOH) eluent at a flow rate of 1.25  
221 mL/min, and a suppressor current of 100  $\mu$ S. For anions we used a 25  $\mu$ L  
222 injection volume, an AS18 column with 20 mM KOH at a flow rate of 1.0  
223 mL/min, and suppressor current of 75  $\mu$ S. For cations, we used a 25  $\mu$ L injection  
224 volume, a CS12 column with 23 mM methyl sulfonic acid at 1.0 mL/min, and  
225 suppressor current of 59  $\mu$ S.

### 226 *3.1 Organic compound analysis by py-GC-MS*

227 Organic compound analyses by py-GC-MS were performed by placing a ~10  
228 mg powdered sample in quartz pyrolysis tubes held in place by quartz wool. The  
229 pyrolysis tubes were placed inside the platinum coil of a CDS 5200 pyroprobe  
230 under helium and heated at a rate of 20°C ms<sup>-1</sup> to the target temperature of 650°C  
231 where it was held for 15 seconds. Although these conditions were different than  
232 those used with the MSL-SAM instrument on Mars (35°C/min up to 835°C under  
233 30 mbar He), the overall results should be similar. The pyrolysis unit was

234 coupled to an Agilent Technologies 6890 gas chromatograph using split injection  
235 at a 10:1 split ratio and an inlet temperature of 250°C. Separation was performed  
236 on a 30 m J&W Scientific DB-5MS Ultra Inert column. The GC-MS oven  
237 temperature program comprised of a start temperature of 40°C held for 2 minute,  
238 followed by a ramp of 7.5°C min<sup>-1</sup> to 310°C where the temperature was held for  
239 10 min. Helium column flow was 1.1 ml min<sup>-1</sup>. Post-separation compound  
240 identification took place using an Agilent 5973 inert Mass Selective Detector,  
241 which collected data over a scan range of  $m/z = 50$  to 550. Assignments are made  
242 by considering the elution order against published work and comparing mass  
243 spectra against a standard library (NIST08).

244 In addition to instrumental blanks, we also pyrolyzed procedural blanks from  
245 the ion chromatography consisting of an aliquot of clean quartz sand and an  
246 aliquot of quartz sand after it had been subjected to the leaching procedure.

## 247 **4. Results**

### 248 *3.1 Effects of leaching ratio on perchlorate concentration*

249 The ClO<sub>4</sub><sup>-</sup> ion salts are highly soluble and thus can easily be leached from  
250 most samples. The samples collected every 5 cm to a depth of 100 cm were used  
251 to understand the effects of the leaching ratio. The results, shown in **Figure 1**,  
252 indicate that there was no significant difference in the concentration of recovered

253  $\text{ClO}_4^-$  between the 1:5, 1:10, and 1:25 leaching ratios. Minor differences are  
254 likely due to sample composition rather than leaching ratio.

### 255 *3.2 Efficacy of perchlorate leaching*

256 To confirm the efficacy of the leaching method for the  $\text{ClO}_4^-$  in the samples, a  
257 set of leaching efficacy analyses were performed using a subset of representative  
258 YU pit samples collected at depths of 25, 60, 75, and 85 cm. Two samples from  
259 each depth were used to provide an indication for the homogeneity of the sample  
260 sizes used. The samples were leached at soil:water ratios of 1:10 and the  
261 concentration of  $\text{ClO}_4^-$  in each measured using IC. The air dried residues from the  
262 1:10 leach were leached a second time and the remaining concentrations  
263 measured. The results (**Figure 2** and **Table 1**) showed that 7-10% of the initial  
264 perchlorate remained in the samples after the first leaching. This corresponds to  
265 the volume of leachate that remained in the sample after filtration, and thus  
266 provided the basis for estimating the residual perchlorate remaining in the  
267 samples that were analyzed by py-GC-MS for organic compounds.

### 268 *3.3 IC analysis of all salts present*

269 The IC analyses of the leachate from the three soil samples showed  
270 similarities in the patterns of salts present, but with order-of-magnitude  
271 differences in concentrations. As can be seen in **Figure 3** and **Table 2**, the  
272 dominant anion in all samples is sulfate ( $\text{SO}_4^{2-}$ ), but is  $\sim 100$  times greater in both

273 the hyperarid LB and YU samples than the RS sample. The same is also true for  
274 the apparent  $\text{SO}_4^{2-}$  parent salts, with the LB and YU soils potentially containing ~  
275 100 times more  $\text{CaSO}_4$  than the RS soils. However, in the LB soil the  $\text{CaSO}_4$  also  
276 dominates all other salts by two orders of magnitude.

### 277 *3.4 Organic compound analysis by py-GC-MS*

278 Samples which were pyrolyzed without leaching showed no detectable  
279 organic matter other than background siloxane (**Figure 4a-d**) and very weak (YU)  
280 or no trace of detectable polychlorinated benzenes (**Figure 5a-b**). The significant  
281 excess in column bleed (siloxane) with the YU sample is likely due to the  
282 interaction of the pyrolyzed perchlorate with the column, rather than the column  
283 itself.

284 Dried sample residues left over after leaching for IC analysis (i.e., the mineral  
285 matrix and water-insoluble organic material) were pyrolyzed and showed a strong  
286 organic response in all cases (**Figures 4e-h, 5c-d**). These residues are most likely  
287 the product of the pyrolysis of water insoluble organics originally present in the  
288 samples. For purposes of comparison, **Figure 4** also shows the results for a  
289 sample containing endolithic cyanobacteria from the Atacama Desert that was  
290 pyrolyzed after leaching. The unleached sample was not analyzed due to  
291 methodological difficulties. Alkanes, alkenes, furfural and benzonitrile were  
292 present in the cyanobacteria sample in similar distributions, although the signal  
293 was stronger as would be expected. No organics were detected from the

294 procedural blank from leaching and the quartz sand source material (data not  
295 shown). The total ionic current (TIC) chromatograms were analyzed for three  
296 categories of organic molecules relating to biomarkers, all of which were detected  
297 in the leached samples, none of which were detected in the sample prior to  
298 leaching. The extracted-ion chromatograms (XIC) showed: alkenes ( $m/z = 55$ )  
299 and alkanes ( $m/z = 57$ ) which are indicative of fragmentation of a biopolymer,  
300 possibly from the cell wall or polymerized fatty acids (**Figure 4b,f**); furfural ( $m/z$   
301  $= 96$ ) and aldehydes ( $m/z=105$ , not shown) are indicators of carbohydrates  
302 (**Figure 4c,g**); and benzonitrile ( $m/z = 103$ ), which is a product of protein  
303 breakdown (**Figure 4d,h**).

304 The extracted-ion chromatograms (XIC) (**Figure 5b,d**) show  $Cl_1$  to  $Cl_6$ -  
305 benzenes ( $m/z = 112, 146, 180, 216, 250, 284$ ) detected in the post-leaching  
306 sample residues. Since the perchlorates and other highly soluble salts were  
307 extracted prior to pyrolysis, these chlorobenzenes were most likely indigenous to  
308 the samples, as either atmospherically deposited polychlorinated benzene (PCB)  
309 contaminants or products formed during possible reactions of indigenous organics  
310 with intermediaries formed during UV production of perchlorate (Carrier and  
311 Kounaves 2015). These results are in contrast to the MSL results, where it was  
312 argued that chlorobenzenes were formed as a reaction product between  
313 perchlorates and benzene carboxylates during pyrolysis (Freissinet *et al.* 2015).

314 Biogenic organic matter is known to be composed predominantly of  
315 macromolecules which are insoluble in water (Killops and Killops 2005; Tissot  
316 and Welte 1984). Given the low amount of organic material in the original  
317 samples and the small uptake expected, the organic content of the leachate is  
318 likely to be below the detection limits of py-GC-MS. Leaching the samples a  
319 second time and analyzing the leachate with UV-fluorescence also showed no  
320 signal at the 1 ppm level using humic acid as a standard.

## 321 **5. Discussion**

322 Experiments which tested the efficacy of the leaching procedure showed that  
323 this method does not necessarily remove all of the inorganic ions from a sample  
324 (**Figures S1-S3**). However, for the samples shown in **Table 3** the presence of  
325 identifiable peaks in the py-GC-MS chromatogram of the post-leaching sample  
326 (**Figures 4 and 5**) indicates that the ionic species present no longer result in  
327 complete destruction of the organic content of the sample, resulting in meaningful  
328 py-GC-MS data from which conclusions may be drawn about the source of the  
329 organic material.

330 Recent work to establish the minimum mass ratio (MMR) of organic carbon  
331 to perchlorate for incomplete combustion during thermal decomposition indicates  
332 that there needs to be at least 3-6× as much organic carbon as  $\text{ClO}_4^-$  by weight or  
333 46-95 moles of carbon for each mole of  $\text{ClO}_4^-$  (Royle *et al.*, 2018). This generally  
334 agrees with the results of this study. The ratios for our samples determined using

335 IC for the oxyanions and TOC for the organic carbon were below this cutoff range  
336 **(Table 3)**. Post-leaching it would be expected that (with extraction efficiency  $\approx$   
337 7%) the LB and RS samples would yield detectable organic molecules but YU  
338 would not, since its ratio still falls below that necessary for detection, 18.5, 540,  
339 and 2.4, respectively. However, all three samples yielded detectable organics  
340 post-leaching. This may be due to sample heterogeneity, or to the fact that the  
341 natural sample contains a mixture of salts and other oxidizing materials which  
342 decreases the MMR for organic carbon.

343 **Figure 3** shows the challenges that will be faced measuring organic  
344 compounds in martian soil. In comparing the levels of salts between the Atacama  
345 and the Mars Phoenix sites, it is clear that even though most of the cations and  
346 anions in the martian soil are similar to those in the Yungay, there is one glaring  
347 difference. For Mars the concentration of  $\text{NO}_3^-$  is an order of magnitude lower,  
348 and the  $\text{ClO}_4^-$  and  $\text{ClO}_3^-$  are almost 3 orders of magnitude higher. With  $\text{ClO}_4^-$   
349 being the dominant species of the oxidizing anions on Mars, this implies that the  
350 level of organic compounds will need to be at least  $\sim 3$ -6 times that of the  
351 perchlorate, or  $\sim 1800$ -3600 ppm to allow for a typical py-GC-MS detection.

352 Despite there being no measurable  $\text{ClO}_4^-$  present at either the Red Sands or  
353 Lomas Bayas samples, no organics were detected before leaching. This shows  
354 that the other oxyanion salts present, (i.e., sulfates, nitrates, chlorates, phosphates)  
355 may also have oxidative effects and can obfuscate organic molecule detection

356 during thermal decomposition. While it would not normally be expected that the  
357 sulfate and phosphate salts would break down and contribute oxygen at the  
358 pyrolysis temperatures used, it is otherwise problematic to explain the non-  
359 detection of organic species in the unleached samples. When mixed with other  
360 phases the onset of thermal decomposition temperature of a number of sulfates  
361 has been shown to drop to well below the temperatures used in this study (Mu and  
362 Perlmutter, 1981; McAdam *et al.*, 2016; Sutter *et al.*, 2017). In these natural soil  
363 samples there is likely a mixture of phases, such as metal-oxides and carbon,  
364 which have this known catalytic effect on the sulfates (Mu and Perlmutter, 1981)  
365 and a similar effect may be expected on the other salts present. Thus, the amount  
366 of oxygen available for combustion of organic carbon increases, even in samples  
367 low in perchlorates. Other than perchlorates, only sulfates have so far been  
368 examined in any detail (Lewis *et al.* 2018; Lewis *et al.* 2015).

369       Considering these findings and the solubility of perchlorate salts in water,  
370 future missions to detect organic matter on Mars must aim for sites with evidence  
371 of water activity. Since  $\text{ClO}_4^-$  (and  $\text{NO}_3^-$ ) is highly soluble and thus a marker of  
372 liquid water, this may be an indication that the area may have been subjected to  
373 "recent" water flow. At the same time, ancient surface or near-surface water flow  
374 may have concentrated organic compounds downslope while simultaneously  
375 leaching away the perchlorate and other soluble oxyanion salts, thus efforts  
376 should focus on analyzing fluvial/lacustrine sediments and especially those sites

377 where there is evidence for post-depositional aqueous activity such as the lower  
378 Murray mudstone (Yen *et al.*, 2017) . This is the one locality where complex,  
379 non-chlorinated organic molecules have been detected on Mars (Eigenbrode *et al.*,  
380 2018) and these sediments have also yielded some of the lowest concentrations of  
381 perchlorate yet measured in the Martian near surface sediments (Sutter *et al.*,  
382 2017). It is thus hypothesized that aqueous flow through the Murray mudstone  
383 has leached the soluble perchlorate out of the sediments, making the organic  
384 matter preserved within this unit more amenable to detection via the thermal  
385 decomposition technique used by SAM. This is in contrast to the other unit in  
386 which ‘indigenous’ organic molecules have been detected in on Mars by SAM,  
387 the Sheepbed mudstone. In the Sheepbed mudstone, only simple chlorinated  
388 organic molecules have been reported, suggesting oxychlorines are having a  
389 deleterious and obfuscatory effect on the detection of organic matter (Freissenet *et*  
390 *al.*, 2015). The Sheepbed mudstone appears to have a similar depositional setting  
391 to the Murray (at least superficially), however, it does not show the same level of  
392 evidence for extensive late-stage fluid flow and aqueous alteration as the Murray  
393 and consequently contains perchlorate levels approximately 10× higher. Areas of  
394 recent subsurface or surface water, such as above near-surface aquifers, near  
395 exposed water-ice and downslope of recurring slope lineae (if they are in fact  
396 deliquescing-brine-induced (Chevrier and Rivera-Valentin 2012; McEwen *et al.*  
397 2011, Dundas and McEwen 2015)), could also be attractive targets in the search

398 for organic molecules. However, these current or recent ‘hydrous’ areas are the  
399 most likely to be capable of supporting Earth-like microbial life and are thus  
400 designated ‘Special Regions’, presenting stringent planetary protection issues if  
401 they are to be explored (Kminek and Rummel 2015; Rummel *et al.* 2014).

402

## 403 **6. Conclusions**

404 Our findings demonstrate that by reducing the salt content of the sample with  
405 water, we counter the effects of pyrolyzing in the presence of oxychlorines and  
406 thus biologically derived organic material is readily identifiable. Since py-GC-  
407 MS is a standard analytical technique for detecting and identifying organic  
408 compounds and biomarkers on Mars and Earth, it is clear that its use in the  
409 presence of oxyanions, and especially perchlorates, will pose enormous  
410 challenges unless their oxidizing effects can be mitigated. We have shown that  
411 simple removal of perchlorate prior to pyrolysis using an aqueous extraction  
412 method allows for the identification of organic compounds reflective of biological  
413 activity. In addition, our results have shown that previous soil analyses which  
414 contained high levels of oxyanions and concluded that organics were either not  
415 present, or at extremely low levels, should be re-examined.

416

417

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425  
426 **Author Disclosure Statement**

427 No competing financial interests exist for any of the authors.

428  
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## 621 **Tables**

622

623 **Table 1.** Sample preparation data for leaching efficiency experiments. Initial  
 624 leaching volume was 10 mL. Recovered volume from Leach-1 was used to  
 625 calculate the expected concentration remaining in Leach-2.

| <i>Sample</i> | <i>Mass</i><br><i>(g ± 0.5 mg)</i> | <i>Recovered Volume</i><br><i>(mL ± 5 µL)</i> |                | <i>Avg. Concentration</i><br><i>(ppb ± 5%)</i> |                | <i>Leaching</i><br><i>Efficiency</i><br><i>%</i> |
|---------------|------------------------------------|-----------------------------------------------|----------------|------------------------------------------------|----------------|--------------------------------------------------|
|               |                                    | <i>Leach-1</i>                                | <i>Leach-2</i> | <i>Leach-1</i>                                 | <i>Leach-2</i> |                                                  |
| Y1-25         | 1.042                              | 9                                             | 9.4            | 26.3                                           | < 5            | 100%                                             |
| Y2-25         | 1.047                              | 9                                             | 9.2            | 21.2                                           | < 5            | 100%                                             |
| Y1-60         | 1.030                              | 9                                             | 9.2            | 176.4                                          | 12.8           | 93%                                              |
| Y2-60         | 1.026                              | 9                                             | 9.6            | 186.9                                          | 12.6           | 93%                                              |
| Y1-75         | 1.039                              | 9                                             | 9.4            | 258.9                                          | 16.4           | 94%                                              |
| Y2-75         | 1.047                              | 9                                             | 9.2            | 148.4                                          | 10.6           | 93%                                              |
| Y1-85         | 1.019                              | 9                                             | 9.2            | 423.3                                          | 21.9           | 95%                                              |
| Y2-85         | 1.029                              | 9                                             | 9.2            | 322.3                                          | 32.1           | 90%                                              |

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636 **Table 2.** Inorganic composition of the Atacama samples as determined by ion  
 637 chromatography. The RS and LB samples were taken from 0-5 cm depth and the  
 638 YU samples from 50 cm depth. These samples were selected to represent "salt  
 639 poor", "sulfate rich", and "oxyanion/salt rich" environments, respectively.  
 640 Average (n=3) concentration of detectable species in soil reported as mg/kg.  
 641 Addition data for the other Yungay depths is shown in **Table S1** in Supporting  
 642 Information.

| <i>Ion<br/>Species</i>        | <i>Sampling Site</i>                  |                                         |                                    |
|-------------------------------|---------------------------------------|-----------------------------------------|------------------------------------|
|                               | <i>Red Sands (RS)</i><br><i>mg/kg</i> | <i>Lomas Bayas (LB)</i><br><i>mg/kg</i> | <i>Yungay (YU)</i><br><i>mg/kg</i> |
| Na <sup>+</sup>               | 31.4±0.2                              | 82±14                                   | 4,937±115                          |
| K <sup>+</sup>                | 22.4±0.1                              | 98±4                                    | 785±14                             |
| Mg <sup>2+</sup>              | 3.04±0.08                             | 39.2±0.1                                | 1,831±4                            |
| Ca <sup>2+</sup>              | 11.2±0.8                              | 5,148±67                                | 3,521±128                          |
| Cl <sup>-</sup>               | 1.86±0.05                             | 12±6                                    | 3,357±4                            |
| SO <sub>4</sub> <sup>2-</sup> | 101±1                                 | 12,810±221                              | 13,981±20                          |
| NO <sub>3</sub> <sup>-</sup>  | 5.5±0.1                               | 63±30                                   | 5,681±70                           |
| PO <sub>4</sub> <sup>3-</sup> | 4.15±0.07                             | < 31                                    | < 8                                |
| ClO <sub>3</sub> <sup>-</sup> | < 0.37                                | < 18                                    | 63±2                               |
| ClO <sub>4</sub> <sup>-</sup> | < 0.03                                | < 0.03                                  | 4.6±0.1                            |
| Total Ions                    | 182                                   | 18,200                                  | 36,000                             |

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647 **Table 3:** Total organic carbon (TOC):Total Cl, N oxyanion (Ox)-salt ratios  
 648 calculated using ion chromatography and TOC data (Schulze-Makuch *et al.*  
 649 2018), before and after leaching. The RS and LB samples were taken from 0-5 cm  
 650 depth and the YU samples from 0-5, 20-30, 50, and 100 cm depths. Three of the  
 651 samples were selected to represent "salt poor" (RS), "sulfate rich" (LB), and  
 652 "oxyanion/salt rich" (YU 50 cm) environments, respectively.

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| <i>Sample Site</i> | <i>TOC<br/>(mg/kg)</i> | <i>Total AOx<br/>(mg/kg)</i> | <i>Ratio<br/>TOC:AOx</i> | <i>Total AOx<br/>post leach<br/>(mg/kg)</i> | <i>Post leach<br/>Ratio<br/>TOC:AOx</i> |
|--------------------|------------------------|------------------------------|--------------------------|---------------------------------------------|-----------------------------------------|
| Red Sands          | 600                    | 111                          | 5.40                     | 1.11                                        | 540                                     |
| Lomas Bayas        | 2380                   | 12,870                       | 0.18                     | 129                                         | 18.5                                    |
| Yungay 0-5 cm      | 700                    | 998                          | 0.70                     | 10                                          | 70                                      |
| Yungay 20-30 cm    | 610                    | 15,050                       | 0.04                     | 150                                         | 4.0                                     |
| Yungay 50 cm       | 480                    | 19,730                       | 0.02                     | 197                                         | 2.4                                     |
| Yungay 100 cm      | 450                    | 75,100                       | 0.01                     | 751                                         | 0.60                                    |

654 **Figure Legends:**

655 **FIG. 1.** Concentration of perchlorate in soil taken from 20 depths and analyzed  
656 using three different leaching ratios 1:5, 1:10, and 1:25 all leached for 1 hr. The %  
657 difference does not vary significantly with leaching ratio and minor differences in  
658 concentration are likely due to differences in sample composition rather than  
659 leaching ratio. The samples taken at 25, 60, 75, and 85 cm (stared), were then  
660 used for further leaching efficacy testing.

661 **FIG. 2.** Leaching efficiency for  $\text{ClO}_4^-$  for eight soil samples. Two samples were  
662 taken from each of four depths, selected to cover a broad range of concentrations.  
663 Detailed data used in this figure is provided in **Table 1**. Solid orange bars  
664 represent the concentration of  $\text{ClO}_4^-$  after the first leaching and green hatched bars  
665 after the second leaching. Error bars are the sum of the propagated uncertainty  
666 from the volume and concentration of the recovered leach and the standard  
667 deviation of three measurements. Dashed line denotes the limit of quantification  
668 (LOQ) for the  $\text{ClO}_4^-$  (2.5 ppb). Additional data for the other Yungay depths is  
669 shown in **Figure S1** in Supporting Information.

670

671 **FIG. 3.** Average (n=3) concentration of anions and cations in the Atacama RS,  
 672 LB, and YU, soil samples as determined by ion chromatography, compared with  
 673 those measured by the Phoenix-WCL and MSL-SAM instruments. The RS and  
 674 LB samples were taken from 0-5 cm depth and the YU samples from 50 cm  
 675 depth. Based on these results we have designated the RS sample as "*salt poor*" (~  
 676 180 ppm total salts), the LB sample as "*sulfate rich*" (~ 13,000 ppm  $\text{SO}_4^{=}$ ), and  
 677 the YU sample as "*oxychlorine rich*" (~ 68 ppm total  $\text{ClO}_4^- + \text{ClO}_3^-$ ). However,  
 678 YU is also clearly "*salt rich*" (36,000 ppm total salts) and also "*oxyanion rich*"  
 679 with ~ 14,000 ppm  $\text{SO}_4^{=}$  and 6,000 ppm  $\text{NO}_3^-$ . For Mars, the concentrations for  
 680 all ions, except  $\text{NO}_3^-$  and  $\text{ClO}_3^-$ , are from the Phoenix-WCL (Kounaves *et al.*  
 681 2010a; 2010b). The data for  $\text{NO}_3^-$  and  $\text{ClO}_3^-$  are estimates from both the Phoenix-  
 682 WCL, the Mars EETA79001 meteorite (Kounaves *et al.* 2014a), and the MSL-  
 683 SAM (Stern *et al.* 2017; 2015).

684 **FIG 4.** Total ion current (TIC) chromatograms showing qualitative data for  
 685 unleached (a-d) and leached (e-h) 50 cm depth soil and rock cyanobacteria  
 686 samples. The TIC of the unleached samples contains only analytical artifacts  
 687 (siloxanes; indicated with \*) where the TIC of the leached samples contain  
 688 numerous peaks. Extracted-ion chromatograms (XIC) included (b, f) alkanes and  
 689 alkenes ( $m/z = 57$  and  $55$ , respectively; alkanes are indicated by carbon number,  
 690 e.g.  $\text{C}_{17}$ ), which are a common component of biopolymers, (c, g) furfural ( $m/z =$

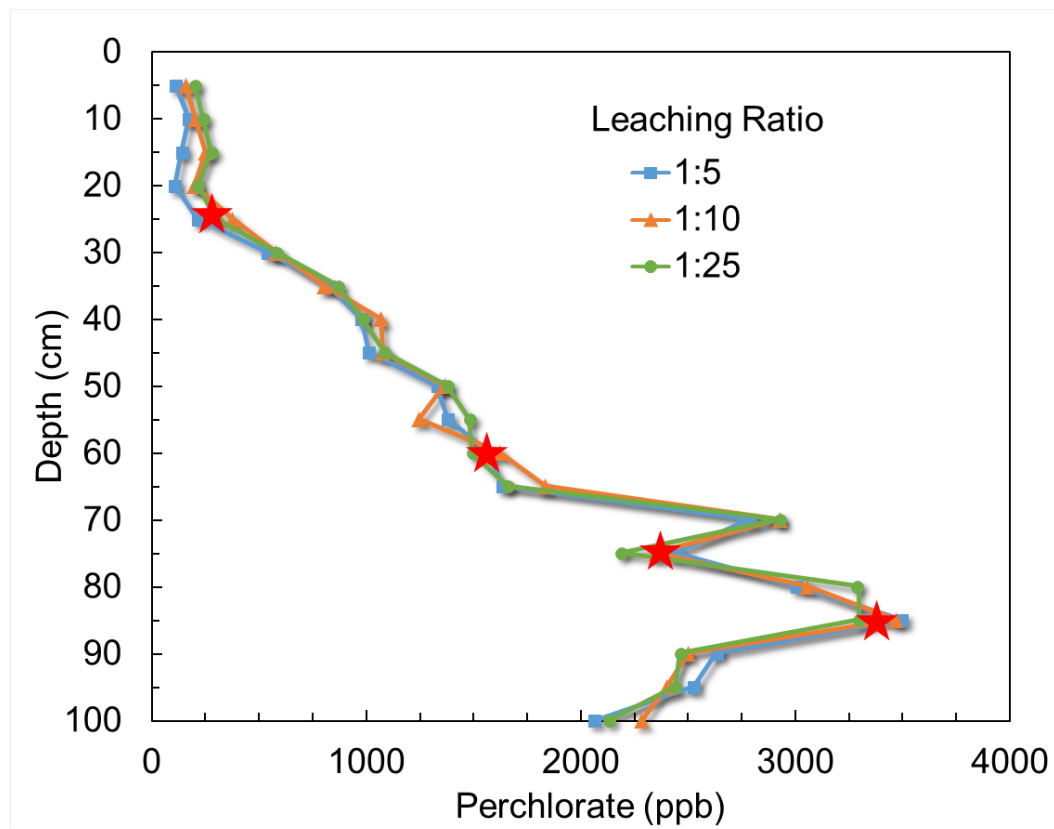
691 96), which represent carbohydrates, and (d, h) benzonitriles ( $u/z = 103$ ), which are  
692 a product of protein breakdown, where N=tetradecene. The unleached  
693 cyanobacteria sample was not amenable to analysis.

694 **FIG. 5.** TIC and XIC highlighting the detection of chlorobenzenes in the (a, b)  
695 unleached and (c, d) leached samples from the Atacama Desert.  $\text{Cl}_1\text{-Cl}_6$ -benzenes  
696 were detected in the leached samples using  $m/z=112, 146, 180, 216, 250$ , and  $284$ .  
697 (\* = column bleed (siloxanes, contaminants from sampling))

698

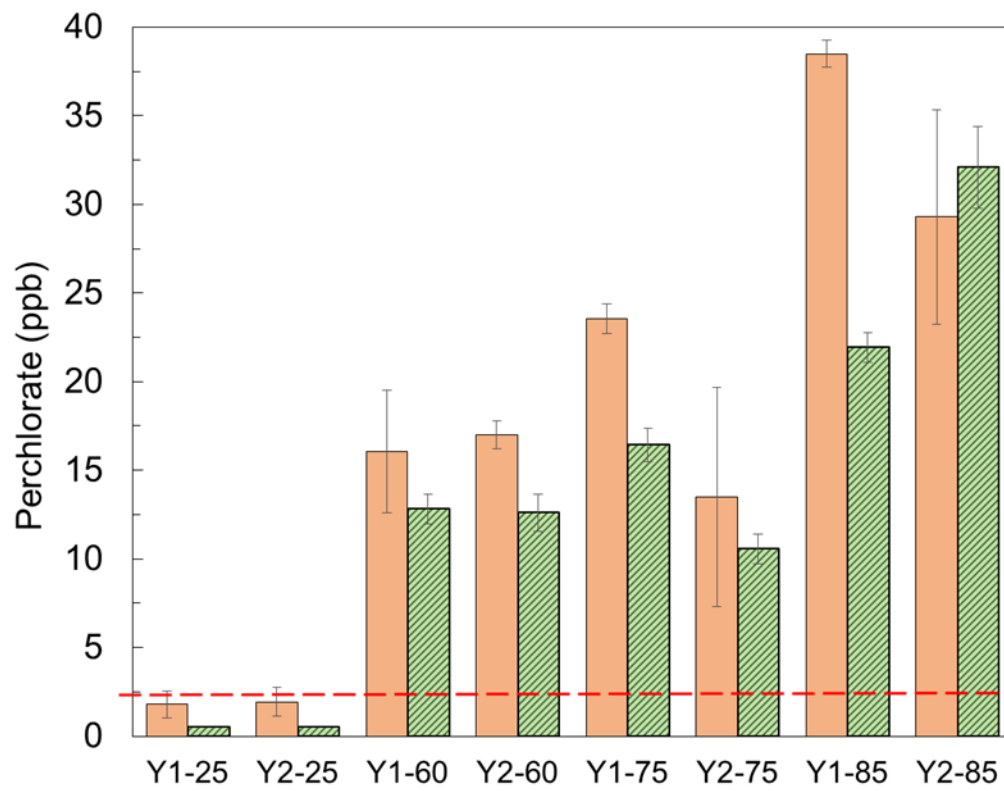
699 **Figures**

700 **FIG. 1.**



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709 **FIG. 2.**



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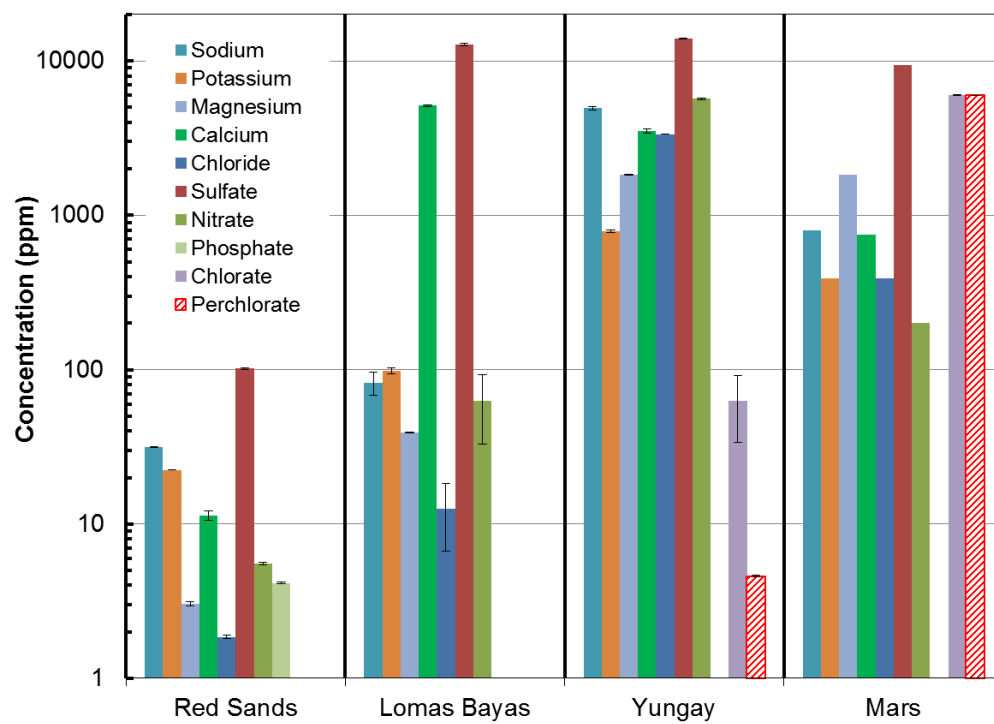
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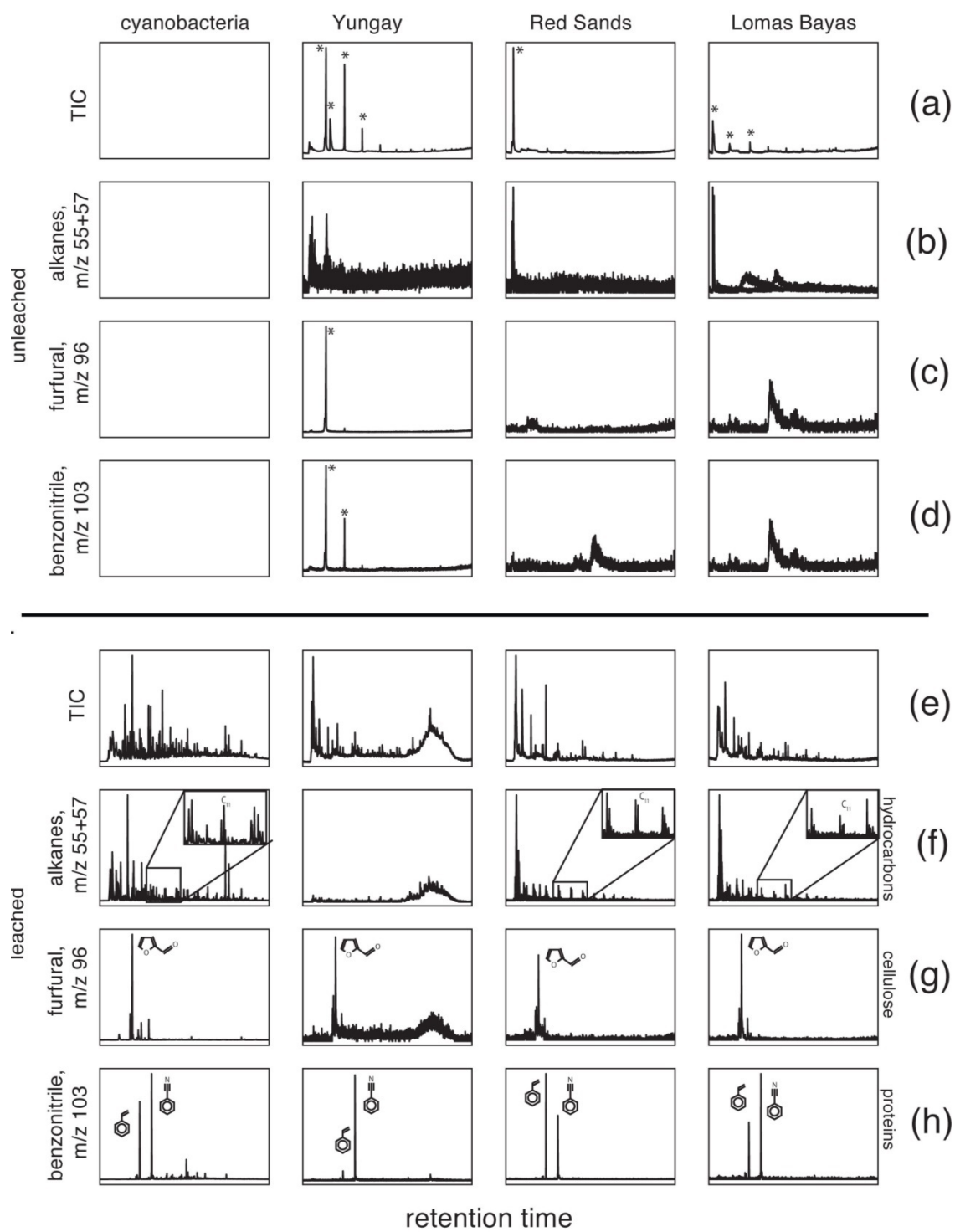
718

719 **FIG. 3.**



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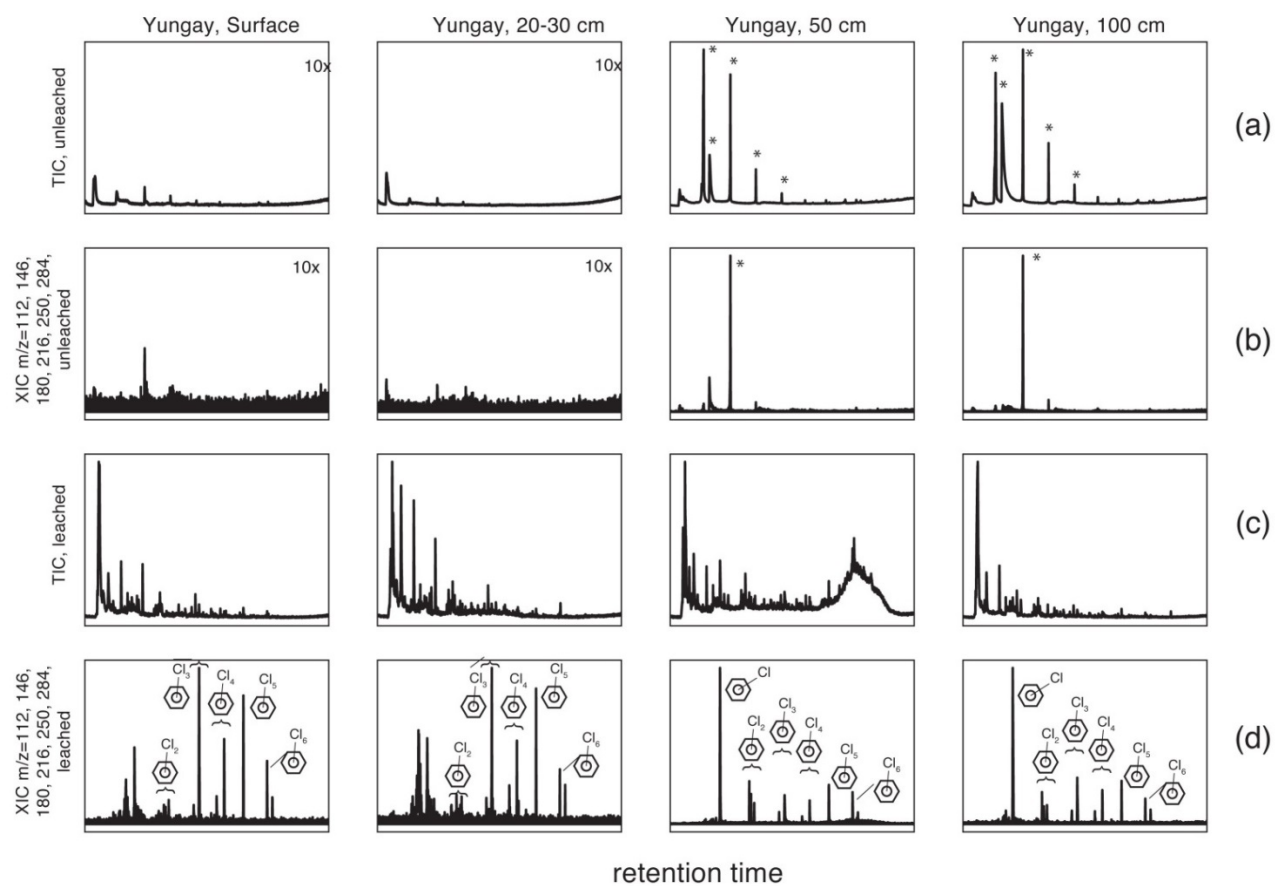
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724 **FIG. 5.**

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## 727 APPENDIX A - Additional Supporting Information

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729

730 **Table S1.** Inorganic composition of the Atacama samples as determined by ion  
731 chromatography for the YU samples for 0-5, 10-30 and 100cm depths. Average  
732 (n=3) concentration of detectable species in soil reported as mg/kg.

733

| <i>Ion<br/>Species</i>        | <i>Yungay (YU) Sampling Depth</i> |                           |                         |
|-------------------------------|-----------------------------------|---------------------------|-------------------------|
|                               | <i>0-5 cm<br/>mg/kg</i>           | <i>20-30 cm<br/>mg/kg</i> | <i>100 cm<br/>mg/kg</i> |
| Na <sup>+</sup>               | 50.5                              | 63.1                      | 25,549                  |
| K <sup>+</sup>                | 1,437                             | ?                         | 835                     |
| Mg <sup>2+</sup>              | 46.9                              | 62.3                      | 1,457                   |
| Ca <sup>2+</sup>              | 471                               | 6,040                     | 4,500                   |
| Cl <sup>-</sup>               | 69.2                              | 70.2                      | 3,133                   |
| SO <sub>4</sub> <sup>2-</sup> | 830                               | 14,980                    | 66,365                  |
| NO <sub>3</sub> <sup>-</sup>  | 88.4                              | 68.6                      | 5,622                   |
| PO <sub>4</sub> <sup>3-</sup> | 81.3                              | < 3                       | < 3                     |
| ClO <sub>3</sub> <sup>-</sup> | < 0.3                             | < 0.3                     | 90.0                    |
| ClO <sub>4</sub> <sup>-</sup> | 0.039                             | 0.114                     | 5.20                    |
| Total Ions                    | 3,074                             | 21,284                    | 107,556                 |

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740 **Figure S1.** Extraction efficiency for chloride, sulfate, and nitrate.

