

Supplementary Material for Charge Transport Phenomena in Heterojunction Photocatalysts: the WO₃/TiO₂ System as an Archetypical Model

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TABLE S1: Parameters used in numerical computations [1–5]

Parameter	Value
TiO ₂ Bandgap Energy (eV)	3.2
TiO ₂ Electron Affinity (eV)	5.1
TiO ₂ Dielectric Constant	41.4
TiO ₂ Electron Effective Mass	10
TiO ₂ Hole Effective Mass	0.8
TiO ₂ Bulk Electron Concentration (cm ⁻³)	10 ¹⁸
TiO ₂ Absorption Co-efficient, α (cm ⁻¹)	9.9 × 10 ³
WO ₃ Bandgap Energy (eV)	2.74
WO ₃ Electron Affinity (eV)	4.91
WO ₃ Dielectric Constant	50
WO ₃ Electron Effective Mass	2.4
WO ₃ Hole Effective Mass	2.4
WO ₃ Bulk Electron Concentration (cm ⁻³)	10 ¹⁹
WO ₃ Absorption Co-efficient, α (cm ⁻¹)	1.02 × 10 ⁵
Wavelength (nm)	365
UVA Intensity (mW cm ⁻²)	3.15
β_n	1
β_p	1
Saturation Velocity (cm s ⁻¹), $V_{sat} = \frac{2.4 \times 10^2}{1 + 0.8 \exp\left[\frac{T}{600}\right]}$	1.03 × 10 ²

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- [1] C. Sotelo-Vazquez, R. Quesada-Cabrera, M. Ling, D. O. Scanlon, A. Kafizas, P. K. Thakur, T.-L. Lee, A. Taylor, G. W. Watson, R. G. Palgrave, et al., *Advanced Functional Materials* **27**, 1605413 (2017).
- [2] A. Kafizas, X. Wang, S. R. Pendlebury, P. Barnes, M. Ling, C. Sotelo-Vazquez, R. Quesada-Cabrera, C. Li, I. P. Parkin, and J. R. Durrant, *The Journal of Physical Chemistry A* **120**, 715 (2016).
- [3] D. O. Scanlon, C. W. Dunnill, J. Buckeridge, S. A. Shevlin, A. J. Logsdail, S. M. Woodley, C. R. A. Catlow, M. J. Powell, R. G. Palgrave, I. P. Parkin, et al., *Nature Materials* **12**, 798 (2013).
- [4] B. Enright and D. Fitzmaurice, *Journal of Physical Chemistry* **100**, 1027 (1996).
- [5] D. Vasileska, S. M. Goodnick, and G. Klimeck, *Computational electronics: semiclassical and quantum device modelling and simulation* (UK: Taylor and Francis, 2010), 1st ed.

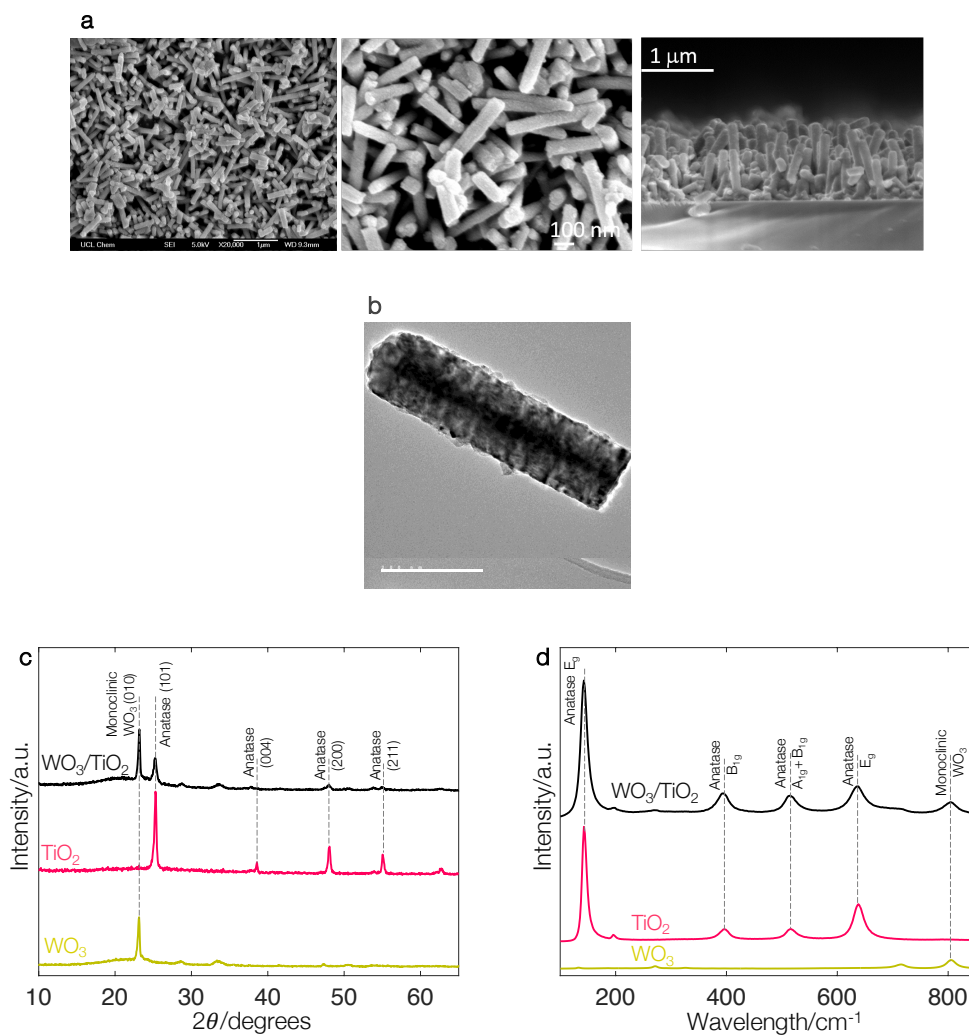


FIG. S1: (a) Scanning electron microscopy (SEM) image and (b) Transmission electron microscopy (TEM), (c) X-ray diffraction patterns of the WT heterojunction film, WO_3 nanorods and anatase TiO_2 ($\lambda = 1.54 \text{ \AA}$), (d) Raman spectroscopy analysis showing the presence of pure anatase TiO_2 and monoclinic WO_3 phases. [1]

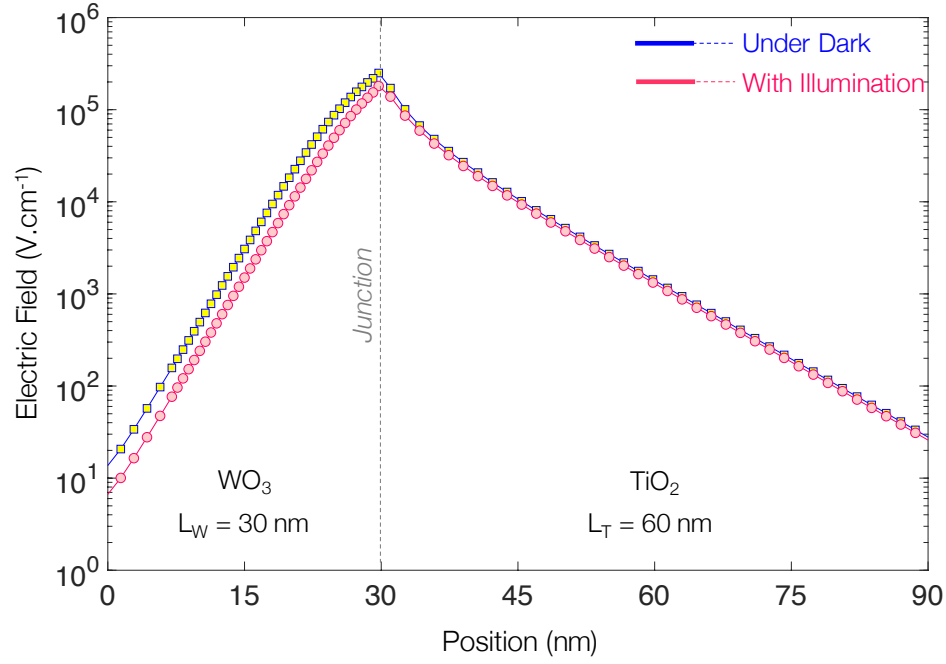


FIG. S2: Calculated electric fields under dark (marked in blue) and UV illumination (marked in red) for the WT junction with $L_W = 30$ nm and $L_T = 60$ nm.

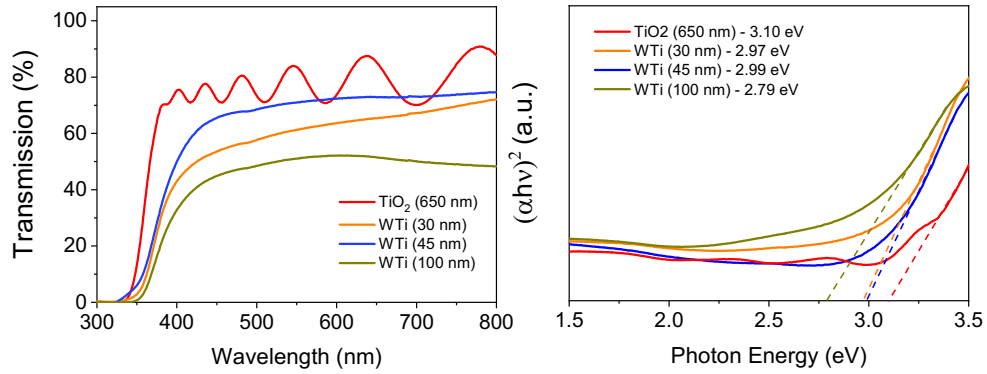


FIG. S3: (a) Absorbance spectra and (b) Tauc plot of WT heterojunctions with different TiO_2 thickness.

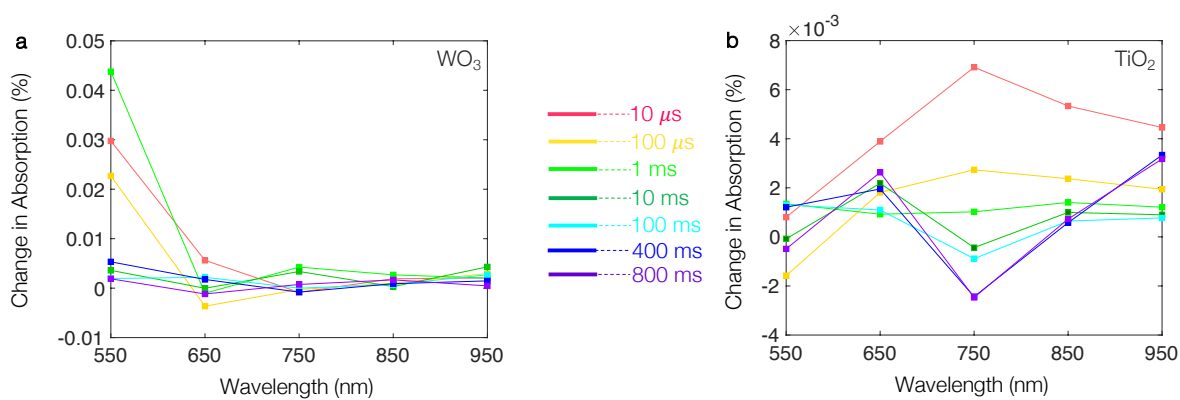


FIG. S4: Transient absorption spectra (probed from 550 to 950 nm) of WO₃ (a) and TiO₂ (b). Samples were excited with a 365 nm laser pulse (1.2 mJ cm^{-2} per pulse, 0.65 Hz pulse rate). ΔA is measured from 10 μ s \rightarrow 800 ms after the laser pulse excitation.

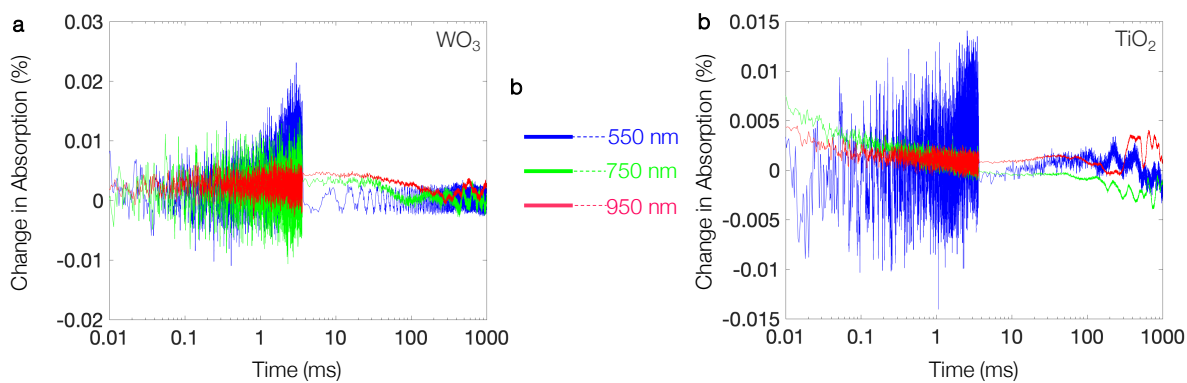


FIG. S5: Transient absorption decay kinetics probed at various wavelengths (550 nm \rightarrow 950 nm) of WO₃ (a) and TiO₂ (b). Samples were excited with a 365 nm laser pulse (1.2 mJ cm^{-2} per pulse, 0.65 Hz pulse rate).