The Reactivity of Hydrogen Enriched Turbulent Flames

F. Hampp^{a,b}, K. H. H. Goh^a, R. P. Lindstedt^{a,*}

^aDepartment of Mechanical Engineering, Imperial College, London, SW7 2AZ, UK ^bGerman Aerospace Center (DLR), Institute of Combustion Technology, Stuttgart 70569,

Germany

Abstract

The use of hydrogen enriched fuel blends, e.g. syngas, offers great potential in the decarbonisation of gas turbine technologies by substitution and expansion of the lean operating limit. Studies assessing explosion risks or laminar flame properties of such fuels are common. However, there is a lack of experimental data that quantifies the impact of hydrogen addition on turbulent flame parameters including burning velocities and scalar fluxes. Such properties are here determined for aerodynamically stabilised flames in a back-to-burnt opposed jet configuration featuring fractal grid generated multi-scale turbulence ($Re_t = 314 \pm 19$) using binary H_2 / CH₄ and H_2 / CO fuel blends. The binary H_2 / CH₄ fuel blend is varied from $\alpha = X_{H2}/(X_{H2} + X_F) = 0.0, 0.2$ and 0.4 - 1.0, in steps on 0.1, and the binary H₂ / CO fuel blend from $\alpha = 0.3 - 1.0$ also in steps of 0.1. The equivalence ratio is adjusted between the mixture specific lower limit of local flame extinction and the upper limit of flashback. The flames are characterised using PIV measurements combined with a flame front detection algorithm. The study quantifies the impact of hydrogen enrichment on (i) turbulent burning velocity (S_T) , (ii) turbulent transport and (iii) the rate of strain acting on flame fronts. Scaling relations (iv) that correlate S_T with laminar flame properties are evaluated and (v) flow field data that permits validation of computational models is provided. It is shown that CH_4 results in a stronger inhibiting effect on the reaction chemistry of H_2 compared to CO, that turbulent transport and burning velocities are strongly correlated with the rate of compressive strain and that scaling relationships can provide reasonable agreement with experiments.

Keywords: Hydrogen, Turbulent Premixed Combustion, Syngas, Turbulent Burning Velocity, Scalar Transport, Rate of Strain

^{*}Corresponding author. Tel: +44 20 7594 7039; Email: p.lindstedt@imperial.ac.uk

1 1. Introduction

The confluence of climate change, environmental protection and dimin-2 ishing fossil fuel resources have promoted the development of low carbon 3 footprint and clean energy technologies [1]. State-of-the-art gas turbines 4 operate under fuel lean conditions optimised for natural gas [2] and single 5 digit nitrogen oxide emissions have been achieved. The use of hydrogen en-6 riched fuel blends, e.g. syngas, offers great potential in the decarbonisation 7 of related technologies by substitution and expansion of the lean operating 8 limit [3]. Decarbonisation by means of hydrogen substitution is most ef-9 fective if hydrogen is produced from renewable energy sources. However, 10 the variability of syngas compositions can lead to fuel flexibility problems 11 for manufacturers [4] and the increased hydrogen concentration to safety 12 concerns [5]. The syngas mixture reactivity is strongly dependent upon hy-13 drogen content with direct implications on flame propagation speeds, ex-14 plosion over-pressures, auto-ignition and turbulence-chemistry interactions. 15 The latter can lead to differences in flame surface area [6] and combustion 16 instabilities [1, 4]. Li et al. [5] measured flame speeds and over-pressures 17 generated in an obstructed flame tube for a wide range of binary and ternary 18 H_2 , CO and CH_4 mixtures, with methane showing a stronger inhibiting effect 19 on the mixture reactivity. Scaling based on the amount of air required to 20 fully oxidise the mixture correlated the fuel composition impact on explosion 21 over-pressures [5] and turbulent flow fields [7]. Simatos et al. [8] investigated 22 the effect of H₂ content in lean ($\Phi = 0.80$) binary CH₄ and CO fuel blends 23 on auto-ignition in turbulent shear layers and the stronger inhibiting effect 24 of CH_4 prevailed. 25

The turbulent burning velocity (S_T) is a key parameter required, for ex-26 ample, to evaluate the strength of explosions [5] and the flashback propensity 27 in gas turbine engines [9]. Investigations seeking to correlate S_T to the tur-28 bulence intensity are quite common [10]. Wang et al. [11] investigated the 29 effect of hydrogen content on S_T and explored scaling relations under gas 30 turbine relevant conditions, and found that H_2 addition alters the correla-31 tion due to preferential diffusive-thermal effects [2]. Scaling relations for the 32 turbulent consumption speed of H_2 /CO mixtures can include flame stretch 33 and pressure effects as shown by Venkateswaran et al. [12] for a wide range of 34 conditions. Shy et al. [13] evaluated the effect of turbulent Reynolds (Re_t) , 35 Damköhler (Da) and Karlovitz (Ka) numbers for lean syngas flames. Zhang 36 et al. [14] evaluated a Kobayashi type turbulent burning velocity correlation 37 for mixtures including H_2 / CO blends with reasonable agreement. Daniele 38 et al. [15] showed a linear correlation of S_T with equivalence ratio and inlet 39 bulk velocity for various syngas related mixtures. The impact of hydrogen 40 blending on methane was found to be particularly prominent under ultra-lean 41 conditions. The Lewis number effect on S_T was also found to be significant 42 in binary H_2 / CH_4 and propane mixtures [16] and the effect of pressure was 43 investigated by Liu et al. [17]. Multiple turbulent burning velocity defini-44 tions (e.g. Bray [18]) have been advanced and Driscoll [19] has shown that 45 the definitions of Shepherd and Cheng [20] and Lawn and Schefer [21] are 46 not equivalent. Yet, the S_T definitions of Bray [18], Lawn and Schefer [21] 47 and a variant proposed by Driscoll [19] agree within $\pm 20\%$ for some pre-48 mixed twin opposed jet flames [22]. However, a broader understanding of the 49 impact of mixture reactivity on turbulent flame propagation is lacking. An 50

additional difficulty is that theories for premixed flames with a $Da \simeq 1$ are not well developed in contrast to the corrugated flamelet regime ($Da \gg 1$). The current study provides an extensive data set that establishes the impact of fuel reactivity on S_T at $Da \simeq 1$ to support the removal of such limitations.

The opposed jet configuration provides a canonical geometry with dis-55 tinct advantages for the assessment of fuel effects [23]. The geometry fea-56 tures aerothermochemistry related flame stabilisation [24], essentially adia-57 batic conditions [25], comparatively well developed turbulence [26], excellent 58 optical access [27] and accurate control of boundary conditions [23]. The tur-59 bulent to bulk strain ratio can be substantially increased without bulk flow 60 instabilities [23] via the use of cross fractal grids (CFGs) [26]. The back-to-61 burnt (BTB) configuration further allows the stabilisation of flames at low 62 Damköhler numbers and permits investigations of combustion regime transi-63 tions [24, 28]. In particular, self-sustained flames detach from the stagnation 64 plane and become independent of the opposing burnt gas [29], while low Da65 combustion is dominated by interactions between the two streams [30, 31]. 66

The current study is using Particle Image Velocimetry (PIV) and a flame front detection algorithm [23, 29] to quantify the impact of H₂ enrichment on CH₄ and CO fuel blends by determining the impact on (i) turbulent burning velocities (S_T) , (ii) turbulent transport and (iii) the rate of strain acting on flame fronts. Scaling relations (iv) that correlate S_T with laminar flame properties are evaluated and (v) flow field data for validation of computational fluid mechanics (CFD) based models for risk assessment is provided.

74 2. Experimental configuration

The present opposed jet configuration was originally developed by Geyer 75 et al. [27] and has been used by Geipel et al. [26], Goh et al. [22, 25] and 76 Hampp and Lindstedt [24, 29, 30, 32]. A schematic is provided in Fig. 1. 77 The burner consists of two identical nozzles with an inner diameter D =78 30 mm, nozzle length of L = 50 mm and nozzle separation of one D. The 79 burner is operated in a back-to-burnt (BTB) configuration to accommodate 80 the wide range of mixture reactivities. The configuration entails considerable 81 advantages in the current context: (i) cases can be compared on a basis of 82 nearly identical flow conditions [29], (ii) all fuel gases are combusted (a safety 83 requirement for H_2 and CO containing mixtures) [5] as global extinction is 84 prevented, (iii) the transition from self-sustained flame propagation to ther-85 mally supported burning can be assessed [24] and (iv) self-sustained flames 86 detach from the stagnation plane with combustion dynamics related to the 87 intrinsic aerothermochemistry. A purpose written LabView interface is used 88 to control the Bronkhorst mass flow controllers with a maximum uncertainty 89 of $\pm 0.5\%$ of full scale. All gases are supplied at a pressure of 4.0 bar(g) with 90 the upper nozzle air seeded with $\sim 3 \ \mu m \ Al_2O_3$ particles [25]. 91

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93 2.1. Flow conditions

In order to determine the impact of fuel reactivity on turbulent burning velocities and scalar transport, the binary H₂ / CH₄ and H₂ / CO fuel blends are varied from $\alpha = X_{H2}/X_F = 1.0, 0.9, 0.8, 0.7, 0.6, 0.5, 0.4, 0.2$ and 0 and 1.0, 0.9, 0.8, 0.7, 0.6, 0.5, 0.4 and 0.3, respectively, where X is the fuel mole



Figure 1: Schematic of the back-to-burnt opposed jet burner configuration. CFG – cross fractal grid, DSI – density segregation iso-contour, SP – stagnation plane, (x,y) – coordinate system, HCP – hot combustion products, PP – perforated plate.

fraction in the blend and F the blending component (i.e. CH_4 or CO). The 98 mixture compositions are listed in the supplementary material. The premixed 99 fuel / air mixtures are injected through the upper nozzle with a constant bulk 100 velocity of $U_b = 9.0 \text{ ms}^{-1}$ and a reactant temperature of $T_r = 298 \text{ K}$. A cross 101 fractal grid (CFG) is used to provide a well developed multi-scale turbulent 102 flow with enhanced intensity [22]. The flow is maintained constant with an 103 integral length scale of turbulence of $L_I = 3.9 \pm 0.2 \text{ mm}$ [30]. Velocity fluc-104 tuations at the nozzle exit of $u_{rms} = 1.4 \pm 0.1 \text{ ms}^{-1}$ are measured using PIV. 105 The turbulent Reynolds number $(Re_t = L_I \cdot u_{rms} / \nu_r = 314 \pm 19)$ is modestly 106 affected by fuel composition changes, primarily due to the kinematic viscos-107 ity of the reactants (ν_r). The turbulent kinetic energy dissipation rate in the 108 reactants is determined as $\epsilon_r = u_{rms}^3 L_I^{-1} \simeq 700 \text{ m}^2 \text{s}^{-3}$. 109

The lower nozzle hot combustion products (HCP) are generated from lean $(\Phi = 0.60)$ premixed 50% H₂ / 50% CH₄ flames stabilised on a perforated plate. The HCP are in close-to thermochemical equilibrium with a nozzle exit temperature of 1640 \pm 7.1 K measured using a 50 μ m R-type thermocouple. The HCP composition, including the residual oxygen concentration, does not exert a strong impact on the self-sustained flames [28, 30, 31] of primary interest here.

117 2.2. Diagnostic setup

The flow field is measured by PIV using a double pulse Nd:YAG laser. 118 The laser sheet thickness is estimated to ~ 0.3 mm using burn marks with 119 an optimum time delay between pulses of $\Delta t = 25 \ \mu s$. The interrogation 120 region of 40 \times 30 mm is resolved by 1376 \times 1040 pixels using a CCD-121 camera. A 100 mm Nikon lens is equipped with a 99% neutral density and a 122 3 nm bandpass filter centred at 532 nm to minimise noise. The PIV vectors 123 are calculated via a multi-pass cross-correlation with decreasing window size 124 resulting in a spatial vector resolution of 0.90 mm and vector spacing of 125 0.45 mm. For each set of conditions, 1000 double frame images are recorded 126 to assure statistically independent data. The relaxation time (τ_p) of the 127 $3~\mu\mathrm{m}~\mathrm{Al_2O_3}$ seeding particles is estimated to $\tau_p=107~\mu\mathrm{s}$ which is similar to 128 the smallest PIV timescale of 102 μ s. The Stokes number $(St = \tau_p/\tau_\eta)$ is 129 determined to $St \simeq 0.75$ based on a Kolmogorov timescale of $\tau_{\eta} \simeq 156 \ \mu s$. 130 Out-of-plane, out-of-pattern, beam steering and uncertainties due to thermal 131 gradients are negligible [24, 29]. 132

133 3. Data analysis

The following section introduces the utilised scaling relations, evaluated turbulent flame properties and uncertainty analysis.

136 3.1. Reaction progress and scalar transport

A conventional PIV particle based density segregation technique is used to 137 detect the instantaneous flame front [23, 33]. The accuracy of the algorithm 138 was evaluated by Hampp and Lindstedt [24] and found to deviate less than 139 two Kolmogorov length scales from the 600 K iso-contour (e.g. Schlieren 140 contour). The turbulent flame brush and the reaction progress variable (\bar{c}) 141 are determined from the statistical location of the instantaneous flame front 142 (c_i) , while the measured turbulent burning velocity $(S_{T,l})$ is determined based 143 on the leading edge ($\bar{c} = 0.02$) [21]. The scalar flux is evaluated as $\overline{c'u'} =$ 144 $\overline{c} \cdot (1 - \overline{c}) \cdot \overline{U_s}$, where the slip velocity is defined as the difference between 145 product and reactant fluid velocities, i.e. $\overline{U_s} = \overline{U_p} - \overline{U_r}$ [34]. 146

147 3.2. Conditional rate of strain

The normal rate of strain conditioned on the instantaneous flame front 148 is calculated using the methodology of Hampp and Lindstedt [29, 30]. The 149 instantaneous planar strain rate tensor $(e_{ij} = 0.5(\partial u_i/\partial x_j + \partial u_j/\partial x_i))$ is 150 calculated from the PIV data. The normal $(a_n = f_{11})$ strain component is 151 determined by $f = \mathbf{R}(\Theta) \cdot e$, where **R** is the rotation matrix and Θ is the 152 angle between the iso-contour normal and the theoretical stagnation point 153 streamline (SPS). The analysis includes the stagnation point movement in 154 the radial limit $\pm 0.5 L_I$ away from the SPS [26]. Correlations with mean 155 values of the conditional normal compressive strain $(\overline{a_n})$ are discussed below. 156

$_{157}$ 3.3. Scaling relations for mixtures with high H_2 content

The wide range of H_2 concentrations results in significant differences in the mixture reactivity as well as reactant and burning properties. Li *et al.* [5, 7] presented a scaling factor for fuel lean blends of H₂ with CH₄ and CO defined based on the amount of air required to fully oxidise each fuel component (0 $\leq \beta \leq 1$). The β factor improved the scaling of explosion over-pressures and related flow velocities compared to $\alpha (= X_{H2}/(X_{H2} + X_F))$ for a wide range of binary and ternary H₂ / CH₄ / CO mixtures. The scaling of auto-ignition in a turbulent shear layer was also improved [8] and β (see Eq. (1), where $(X_k/X_A)_{st}$ is the stoichiometric fuel air ratio) is used here.

$$\beta = \left(\frac{X_{H2}}{(X_{H2}/X_A)_{st}}\right) / \left(\frac{X_{H2}}{(X_{H2}/X_A)_{st}} + \frac{X_F}{(X_F/X_A)_{st}}\right)$$
(1)

167 3.4. Turbulent burning velocity scaling relationships

Classical theories for turbulent combustion resulting in eddy breakup based models for the reaction rate source term (e.g. [35]) only provide a scaling of the turbulent burning velocity based on the (isotropic) velocity fluctuations (e.g. u') as shown via Kolmogorov, Petrovskii and Piskunov (KPP) [36] or eigenvalue [37] analyses. The velocity fluctuations are kept constant in the current study to isolate the impact of the mixture reactivity. The classical form for expressing the turbulent burning velocity is:

$$\frac{S_T}{S_L} = 1 + c \left(\frac{u_l'}{S_L}\right)^m \text{ with } 0.5 \le m \le 1$$
(2)

Examples of correlations (e.g. [10, 11, 13, 14]) for S_T from Zhang *et al.* [14] ($S_T^{\#}$), Bradley *et al.* [38] (S_T^{\ddagger}) and Peters [39] (S_T^{\dagger}) are given in Eq. (3),

$$\frac{S_T^{\#}}{S_L} = 2.92 \left(\frac{u_l'}{S_L}\right)^{0.55} \\
\frac{S_T^{\ddagger}}{S_L} = \frac{0.88}{(KLe_r)^{0.3}} \left(\frac{u_l'}{S_L}\right) \text{ where } K = 0.157 \left(\frac{u_l'}{S_L}\right)^2 Re_t^{-1/2} \\
\frac{S_T^{\ddagger}}{S_L} = 1 - \frac{\psi L_I}{4\delta_f} + \left[\left(\frac{\psi L_I}{4\delta_f}\right)^2 + \psi \frac{u_l' L_I}{S_L \delta_f}\right]^{\frac{1}{2}}$$
(3)

In Eq. (3), $\psi = 0.78$, u'_l is the axial velocity fluctuation at the leading edge, δ_f the laminar flame thickness, K the Karlovitz stretch factor and Le_r the Lewis number (ratio of Schmidt and Prandtl numbers) of the reactants. The expression for Le_r is given in Eq. (4) and assumes that the deficient reactant is the dominant fuel component for the current fuel lean mixtures.

$$Le_r = \frac{X_{H_2}Le_{H_2} + X_F Le_F}{X_{H_2} + X_F}$$
(4)

A fractal flame front analysis has been used to show that the rate of 182 reaction depends on the ratio of the laminar burning (S_L) and Kolmogorov 183 $(V_{\kappa} = (\nu_r \epsilon_r)^{1/4})$ velocities [40]. Such forms are evidently more applicable 184 in the current context where fuel reactivity comes to the fore. The applied 185 KPP analysis [41] provides the estimate of the turbulent burning velocity 186 (S_T^*) shown in Eq. (5). The form includes the KPP limit eigenvalue ($\Lambda = 2$), 187 the customary turbulent eddy viscosity $C_{\mu} = 0.09$ and reaction rate $C_R \simeq 4$ 188 constants [42]. The Le_r correction by Aluri *et al.* [43] gives $C_R = 4.0/e^{Le_r-1}$. 189

$$\frac{S_T^*}{S_L} = 1 + \Lambda \cdot \sqrt{\frac{3}{2} C_R \frac{C_\mu}{Sc_t} \frac{S_L}{V_\kappa}} \cdot \left(\frac{u_l'}{S_L}\right)$$
(5)

¹⁹⁰ The turbulent Schmidt number (Sc_t) is also required and depends greatly on

the flow configuration. It is typically in the range $0.3 \leq Sc_t \leq 1$ [44] and 191 a common value $Sc_t = 0.7$ is used here. As discussed below, the gradient 192 diffusion hypothesis used to derive Eq. (5) is subject to significant uncertain-193 ties. Laizet and Vassilicos [45] have further shown that scalar transport can 194 be significantly enhanced for fractal grids due to the multiscale nature of the 195 generated turbulence. Nevertheless, Goh et al. [22, 23] have shown that the 196 above expression can provide reasonable agreement and it was further noted 197 that a modest reduction in Sc_t would bring improved agreement. 198

¹⁹⁹ 3.5. Laminar flame properties required for scaling

The laminar flame data, i.e. unstrained burning velocity $(S_{L,0})$, flame 200 thickness (δ_f) and adiabatic flame temperature (T_{ad}) , required for the eval-201 uation of the used scaling relationship, are obtained from freely propagating 202 flame calculations using detailed chemistry with differential diffusion effects 203 included [46]. Li et al. [5] validated the chemical mechanism for a wide range 204 of binary fuel mixtures. The inlet conditions of the laminar flame calcula-205 tions matched the experiment, i.e. $T_r = 298$ K, $P_0 = 1$ atm and species mole 206 fractions. The computational domain is resolved by 550 nodes featuring a 207 mesh size of $\sim 10 \ \mu m$. Reactant mixture properties such as kinematic vis-208 cosity (ν_r) , density (ρ_r) and Lewis number (Le_r) are also inferred from the 209 calculations. The laminar extinction strain (a_q) is determined by means of 210 counterflow flame calculations. The computational domain of the latter is 211 resolved by 340 nodes with a local resolution of $\sim 20 \ \mu m$ in the reaction zone. 212 The rate of strain is increased in steps of 100 s⁻¹ until the laminar flames 213 extinguished. The extinction strain is defined as the highest rate of strain 214 prior extinction for each mixture. The strained laminar burning velocity (S_L) 215

used for the theoretical scaling relations is determined at the bulk strain rate $(a_b = 600 \text{ s}^{-1})$ matching the experiment. This accounts for the differences in stretch effects on S_L for the wide range of mixture reactivity investigated and improves the predictability of the scaling relations [12]. The calculation results are summarised in Table 1. The data are further used to determine the Damköhler number $(Da = (L_I \cdot S_L)/(u_{rms} \cdot \delta_f))$ as listed in Table 2.

222 3.6. Uncertainty analysis and data rejection

The flames of interest in the current investigation are within or close-to 223 the thin reaction zone regime. Moreover, the heat release parameter is only 224 weakly dependent on the fuel blending fractions with similar adiabatic flame 225 temperature for a given Φ . Consequently, the effect of the fuel mixture com-226 position on the detected flame front and thus reaction progress variable is 227 negligible for a given stoichiometry. In the range from $\bar{c} = 0.5 \pm 0.1$, i.e. the 228 conditioning variable, the reactant and product fluid velocity changes are <229 2.5%. Consequently, a translation of the detected instantaneous iso-contour 230 within the laminar flame thickness has a negligible impact on the measured 231 scalar flux. A translation of the detected instantaneous iso-contour also pro-232 vides the largest uncertainties in the measured turbulent burning velocities 233 and the turbulence intensity required for the determination of $S_T^{\#}, S_T^{\ddagger}, S_T^{\dagger}$ 234 and S_T^* . Therefore, a sensitivity analysis of the leading edge position (x_l) on 235 S_T and u'_l is conducted in the range from $x_l \pm 0.25 L_I$. The range corresponds 236 to pure mixing dominated reaction progress that provides the lowest gradi-237 ent in the reaction progress variable and thus a conservative estimate. The 238 turbulent intensity is approximately constant in the proximity of the leading 239 edge with a variation below 3% for the above x_l variation. The effect of x_l 240

on the measured S_T^* is below 10%. Thus, a shift in the reaction progress variable iso-contour, e.g. due to uncertainties in detection, fuel composition and changes in equivalence ratio, has a modest impact on the determined turbulent burning velocities.

Flame propagation in the BTB configuration can be influenced or gov-245 erned by the counterflow hot combustion products for cases with insufficient 246 flame detachment from the stagnation plane [30]. This can result in ques-247 tionable S_T values as the reaction progress may become influenced by auto-248 ignition events. Accordingly, the mean velocity at the trailing edge $(\overline{u_t})$ of 249 the flame brush (i.e. at $\bar{c} = 0.95$) is evaluated and cases where $\overline{u_t}$ is negative 250 (the flow direction of the HCP) are removed. For such cases the thermal 251 support provided by the HCP can have a significant impact. 252

253 4. Results and discussion

254 4.1. Flow field statistics

An example of unconditional velocity statistics along the theoretical stag-255 nation point streamline is shown in Fig. 2 for the 60% H₂ / 40% CH₄ mix-256 ture at $\Phi = 0.50$ and 0.70 and 60% H₂ / 40% CO mixture at $\Phi = 0.45$. An 257 increase in mixture reactivity (e.g. H_2 content or Φ) results in a faster prop-258 agating flame that stabilises further upstream. An increase in equivalence 259 ratio further leads to a stronger dilatation. The latter results in an elevated 260 mean axial velocity (\overline{U}/U_b) and an earlier and more pronounced peak in 261 the axial velocity fluctuations $(\sqrt{u'u'}/U_b)$ [23]. The stronger inhibiting ef-262 fect of methane on the H₂ reaction chemistry compared to carbon monoxide, 263 that was also observed in turbulent explosions [5, 7] and auto-ignition re-264

lated flame stabilisation [8], is readily evident in the unconditional velocity statistics. This can be inferred from Fig. 2, where the unconditional velocity statistics of the 60% H₂ / 40% CO mixture at $\Phi = 0.45$ is distinctly closer to the 60% H₂ / 40% CH₄ mixture at $\Phi = 0.70$ than the $\Phi = 0.50$ flame.



Figure 2: Unconditional velocity statistics (\overline{U} mean axial velocity, u' axial fluctuation, v' radial fluctuation) along the stagnation point streamline, normalised by the bulk velocity (U_b), for the 60% H₂ / 40% CH₄ and 60% H₂ / 40% CO mixtures at different Φ .

The corresponding conditional reactant and product fluid velocity statis-269 tics are depicted in Figs. 3 and 4, respectively. The advanced reaction onset 270 for the case with $\Phi = 0.70$ yields an elevated mean axial reactant veloc-271 ity $(\overline{U_r}/U_b)$ in the proximity of x/D = 0. As the turbulent flow field is 272 maintained constant, the conditional axial and radial reactant velocity fluc-273 tuations are very similar. The latter also highlights the good experimental 274 repeatability and accurate control of boundary conditions that is essential 275 for the comparisons with numerical investigations. The enhanced dilation 276 of the $\Phi = 0.70$ case leads to the higher mean axial product fluid velocity 277 and reduced fluctuations close to the reaction onset (i.e. $x/D \simeq 0$). The 278



Figure 3: Conditional reactant fluid velocity statistics (\overline{U} mean axial velocity, u' axial fluctuation, v' radial fluctuation) along the stagnation point streamline, normalised by the bulk velocity (U_b), for the 60% H₂ / 40% CH₄ and 60% H₂ / 40% CO mixtures at different equivalence ratios.

²⁷⁹ 60% H₂ / 40% CO at $\Phi = 0.45$ shows conditional reactant and product fluid ²⁸⁰ velocities and fluctuations similar to the $\Phi = 0.70$ mixture with CH₄. This ²⁸¹ can be attributed to the advanced reaction onset and flame anchoring in low ²⁸² strain regions where dilatation is more effective.

283 4.2. Turbulent burning velocity

The measured leading edge ($\bar{c} = 0.02$ iso-contour) [21] turbulent burning 284 velocity $(S_{T,l})$ is depicted for all H₂ / CH₄ mixtures in the top panel of Fig. 5 285 as a function of β . As the amount of CH₄ is increased, a higher equivalence 286 ratio is required to stabilise a self-propagating flame that detaches from the 287 stagnation plane. For example, the pure H_2 / air flame is stabilised at $\Phi =$ 288 0.35 (i.e. upper limit to avoid flashback). A CH₄ blending of 20% and 50% 289 results in a significantly decreased reactivity, a strong reduction of $S_{T,l}$ and 290 increased upper Φ limits of 0.50 and 0.80, respectively. Scaling of $S_{T,l}$ with 291



Figure 4: Conditional product fluid velocity statistics (\overline{U} mean axial velocity, u' axial fluctuation, v' radial fluctuation) along the stagnation point streamline, normalised by the bulk velocity (U_b), for the 60% H₂ / 40% CH₄ and 60% H₂ / 40% CO mixtures at different equivalence ratios.

the axial velocity fluctuations collapses the data to $S_{T,l}/\sqrt{u'_l u'_l} = 2.68 \pm 0.19$. 292 By omitting cases that are influenced by the HCP (i.e. grey data points in 293 Fig. 5) the predictive capability of the KPP type velocity in Eq. (5) over 294 the current range is reasonable with $S_{T,l}/S_T^* = 1.16 \pm 0.07$. Scaling the 295 data with the correlations from Peters [39], Bradley et al. [38] and Zhang et 296 al. [14] result in $S_{T,l}/S_T^{\dagger} = 2.83 \pm 0.65, S_{T,l}/S_T^{\dagger} = 1.64 \pm 0.11, S_{T,l}/S_T^{\#} =$ 297 1.95 ± 0.11 , respectively. The KPP derived expression also shows reasonable 298 agreement with $S_{T,l}/S_T^* = 1.24 \pm 0.09$ over the entire H₂ / CO data range. 299 Scaling by $\sqrt{u'_l u'_l}$ results in 2.66 \pm 0.20 and the correlations from Peters [39], 300 Bradley et al. [38] and Zhang et al. [14] result in $S_{T,l}/S_T^{\dagger} = 3.16 \pm 0.57$, 301 $S_{T,l}/S_T^{\ddagger} = 1.76 \pm 0.16, S_{T,l}/S_T^{\#} = 2.02 \pm 0.21$, respectively. 302

The level of disagreement obtained with some of the scalings is not surprising. The opposed jet configuration features an imposed pressure gradient



Figure 5: Leading edge turbulent burning velocity $(S_{T,l})$ for H_2 / CH_4 mixtures as a function of β . Top: Measured $S_{T,l}$; Bottom: $S_{T,l}/S_T^*$. The grey symbols indicate thermally supported cases.

and a complex transition between gradient and non-gradient transport. In 305 addition, it has been shown that pressure transport, dilation and scram-306 bling terms play a significant role [47] in the flamelet regime. Furthermore, 307 the observed trend is in agreement with the observations of Laizet and Vas-308 silicos [45] that scalar transport is significantly enhanced by fractal grids. 309 Accordingly, the systematic deviation $\simeq 21\%$ obtained with the KPP de-310 rived scaling expression obtained with $Sc_t = 0.7$ appears modest and can be 311 further reduced to $\simeq 10\%$ for $Sc_t = 0.5$. 312

The influence of CO blending on the mixture reactivity is strongly reduced compared to CH₄, in particular for lean cases as illustrated in Fig. 6. For example, a mixture with $\Phi = 0.35$ can accommodate a 30% CO substitution while maintaining a self-sustained flame. By contrast, a 20% CH₄ substitution results in thermally supported burning. The turbulent burning velocity reduces by ~ 0.02 ms⁻¹ per % of CO substitution and approximately twice this value for CH₄ substitution in the range $80 \leq H_2$ (%) ≤ 100 content.



Figure 6: Leading edge turbulent burning velocity $(S_{T,l})$ for H₂ / CO mixtures as a function of β . Top: Measured $S_{T,l}$; Bottom: $S_{T,l}/S_T^*$.

This strong difference is also evident at $\Phi = 0.50$, where $S_{T,l}$ reduces by 0.01 and 0.02 ms⁻¹ per percent CO or CH₄ substitution corresponding to a variation from $0.3 \leq \beta \leq 0.5$ for both blends. The strong inhibiting effect of CH₄ on the H₂ reaction chemistry compared to CO was also evident in laminar flames, explosion over pressures and fast turbulent deflagrations [5, 7] as well as auto-ignition in a turbulent shear layer [8].

326 4.3. Reaction progress conditions

The conditional reactant $(\overline{U_r})$, product $(\overline{U_p})$ and slip velocity $(\overline{U_s}/U_b)$ are 327 depicted along with the axial scalar flux $(\overline{c'u'})$ across the turbulent flame 328 brush in Fig. 7 for the 60% H₂ mixture at $\Phi = 0.50$ and 0.70 for CH₄ and 329 $\Phi = 0.45$ for CO. The $\Phi = 0.70$ mixture exhibits a higher $S_{T,l}$, an earlier 330 reaction onset and thus a higher reactant velocity throughout the entire flame 331 brush compared to the leaner mixtures. This leads to a higher product fluid 332 velocity, advanced dilatation and consequently a less negative slip velocity 333 and reduced gradient scalar flux. The reduced inhibiting effect of CO on the 334



Figure 7: Normalised slip velocity $(\overline{U_s})$ and scalar flux (\overline{cu}) as a function of \overline{c} for the mixture 60% H₂ / 40% CH₄ or CO. \Box CH₄ $\Phi = 0.50$; \circ CH₄ $\Phi = 0.70$; \times CO $\Phi = 0.45$.

H₂ reactivity leads to a reduced gradient scalar flux for the CO ($\Phi = 0.45$) compared to the CH₄ ($\Phi = 0.50$) mixture.

Tian and Lindstedt [47] showed that the product τS_L , where $\tau = (T_{ad} - T_{ad})$ 337 T_r/T_r is the heat release parameter, scales the pressure transport, dilatation 338 and scrambling terms in the flamelet regime of combustion, while Li *et al.* [5] 339 showed that a normalised scaling based on $(\tau+1)S_L$ can successfully correlate 340 syngas explosion data. The influence of τS_L on the scalar flux conditioned on 341 the $\bar{c} = 0.50$ iso-contour is shown in Fig. 8 for varying H₂ concentration and 342 Φ while maintaining the other parameters constant to quantify their impact. 343 The Φ variation is performed for the 60% H₂ fuel blends in the range 0.50 344 \leq Φ \leq 0.70 for CH_4 and 0.40 \leq Φ \leq 0.45 for CO due to the number of 345 available points. For a given lean mixture, an increase in Φ exhibits a two-346 fold effect that influences the scalar flux: (i) a higher burning velocity due 347 to the elevated mixture reactivity and (ii) an enhanced dilatation due to 348 the higher heat release parameter. For CH_4 / H_2 , $S_{T,l}$ increases from 4.7 to 349

 6.2 ms^{-1} and τS_L from 1.4 to 3.0 ms⁻¹ with Φ . For CO / H₂, $S_{T,l}$ increases 350 from 5.5 to 5.9 ms⁻¹ and τS_L from 1.5 to 2.0 ms⁻¹. The scalar flux in 351 the centre of the turbulent flame brush decreases accordingly (i.e. towards 352 counter-gradient transport) from -0.41 to -0.27 ms⁻¹ and -0.48 to -0.37 ms⁻¹ 353 with increasing Φ for CH₄ and CO blends, respectively. The H₂ variation for 354 the CH₄ and CO blends is performed at $\Phi = 0.80$ and 0.35, representing the 355 highest and lowest heat release limits. The addition of H_2 to CH_4 / (CO) 356 yields a pronounced / (modest) increase of τS_L of 25 \times 10⁻³ ms⁻¹ / (8.7) 357 $\times 10^{-3} \text{ ms}^{-1}$) per percent H₂ substitution for the given Φ . Moreover, the 358 measured $S_{T,l}$ increases from 4.4 to 5.5 ms⁻¹ by moving from 100% CH₄ to the 359 50% H₂ / 50% CH₄ fuel blend at $\Phi = 0.80$. Similarly, $S_{T,l}$ increases from 4.8 360 to 5.5 ms^{-1} by increasing the H₂ content from 70% to 100% in the CO blend 361 at $\Phi = 0.35$. By contrast, $\overline{cu}|(\overline{c} = 0.50)$ decreases significantly with increasing 362 H_2 content for both fuel blends but at a different rate. The transition towards 363 counter-gradient transport is more rapid in CO mixtures where $\overline{cu}|(\overline{c}=0.50)$ 364 reduces by 0.57 ms⁻¹ per unit increase of τS_L (5.7 × 10⁻³ ms⁻¹ per % H₂ 365 increase) in CO blends compared to $0.30 \text{ ms}^{-1} (8.7 \times 10^{-3} \text{ ms}^{-1})$ in CH₄. The 366 trends are consistent for all conditions with the acceleration across the flame 367 front, as expected, stronger for mixtures with a higher $S_{T,l}$. With increasing 368 reactivity, $\overline{cu}|(\overline{c}=0.50)$ moves towards counter-gradient transport (i.e. less 369 negative values), yet the transition is suppressed at the current turbulence 370 levels. The stronger inhibiting effect of CH₄ prevails for $\overline{cu}|(\overline{c}=0.50)$. 371

The turbulent flow field (i.e. U_b , u_{rms} and Re_t) is maintained constant, yet the rate of the normal compressive strain increases towards the stagnation plane. Selected data are plotted in terms of $S_{T,l}$ as function of the mean



Figure 8: Scalar flux (\overline{cu}) conditioned on the $\overline{c} = 0.5$ iso-contour as a function of τS_L for selected data. Left: H₂/CH₄ with 0% – 50% H₂ at $\Phi = 0.80$; $0.50 \leq \Phi \leq 0.70$ at 60% H₂/40% CH₄; Right: H₂/CO with 70% – 100% H₂ at $\Phi = 0.35$; $0.40 \leq \Phi \leq 0.45$ at 60% H₂/40% CO.

- normal compressive strain conditioned on the instantaneous flame front $(\overline{a_n},$ 375 see Sec. 3.2) in Fig. 9 to delineate the effect of H_2 addition and Φ separately. 376 An increase in H₂ content from 0 to 50% in CH₄ with $\Phi = 0.80$ (2.0 $\leq \tau S_L$ 377 $(ms^{-1}) \leq 3.5$ results in a reduction of $\overline{a_n}$ acting on the flame front from -582 378 to -39.4 s^{-1} . This is attributed to the increasing detachment of the flame 379 front from the stagnation plane with $S_{T,l}$, i.e. from 4.4 to 5.5 ms⁻¹. The 380 corresponding scalar flux conditioned on the $\bar{c} = 0.50$ iso-contour reduces 381 from -0.53 to -0.09 ms⁻¹. 382
- An increase in Φ (0.50 to 0.70) of the 60% H₂ / CH₄ mixture results in 383 a similar increase of $S_{T,l}$ (4.7 to 6.2 ms⁻¹) and τS_L (1.4 to 3.0 ms⁻¹) and 384 a weaker reduction in $\overline{a_n}$ (-493 to -232 s⁻¹) as well as $\overline{cu}|(\overline{c} = 0.5)$ (-0.41 385 to -0.28 ms^{-1}). Hydrogen blending to CO (70 to $100\% \text{ H}_2$) at $\Phi = 0.35$ 386 causes an increase of $S_{T,l}$ from 4.8 to 5.5 ms⁻¹ and a reduced mean normal 387 compressive strain from -473 to -231 s⁻¹. The scalar flux reduces towards 388 counter-gradient transport from -0.47 to -0.34 ms⁻¹. The increase in Φ 389 leads to an increase in $S_{T,l}$ from 5.5 to 5.9 ms⁻¹, $\overline{a_n}$ is reduced from -422 to 390 -291 s^{-1}) and $\overline{cu}|(\overline{c} = 0.5)$ from -0.48 to -0.37 ms^{-1} . 391



Figure 9: Turbulent burning velocity $(S_{T,l})$ vs the mean normal compressive strain acting on the flame front $(\overline{a_n})$. Left: H₂/CH₄ with 0% \leq H₂ \leq 50% at $\Phi = 0.80$; 0.50 $\leq \Phi \leq 0.70$ at 40% CH₄; Right: H₂/CO with 70% \leq H₂ \leq 100% in CO at $\Phi = 0.35$; 0.40 $\leq \Phi \leq 0.45$ at 40% CO. The symbol colour blue to red indicates $\overline{cu}|(\overline{c} = 0.50)$, i.e. gradient towards counter-gradient transport.



Figure 10: Scalar flux conditioned at the centre of the turbulent flame brush $(\overline{cu}|(\overline{c}=0.5))$ as a function of mean normal compressive strain acting on the flame front $(\overline{a_n})$. Top left: H₂ / CH₄; Right: H₂ / CO.

The correlation between the mean normal compressive strain and scalar flux conditioned on ($\bar{c} = 0.50$) is depicted in Fig. 10 for all data points. The data are very well correlated with linear regression coefficients (R^2) of 0.93 and 0.95 for H_2 / CH_4 and H_2 / CO, respectively. Consequently, scalar transport is strongly influenced by the total rate of strain strain.

397 5. Conclusions

The present study has quantified the impact of the reactivity of hydro-398 gen enriched fuel blends, e.g. syngas, on turbulence-chemistry interactions 399 and scalar transport utilising a back-to-burnt opposed jet configuration for 400 aerodynamic flame stabilisation. The turbulent flow field $(Re_t = 314 \pm 19)$ 401 and the burnt gas state temperature $(T_{HCP} = 1640 \pm 7 \text{ K})$ are maintained 402 constant to delineate chemistry effects of binary H_2 / CH_4 and H_2 / CO fuel 403 blends. The $\rm H_2$ content is varied from pure $\rm CH_4$ to $100\%~\rm H_2$ and from 30 404 to 100% H₂ in CO blends. The equivalence ratio is adjusted between the 405 mixture specific lower limit of local flame extinction and the upper limit of 406 flashback. An increase in mixture reactivity results in a higher turbulent 407 burning velocity. Methane exhibits a stronger pronounced inhibiting effect 408 on the hydrogen chemistry compared to carbon monoxide with the β scaling 409 attenuating the differences. The KPP-type approach predicts the measured 410 $S_{T,l}$ within 21 \pm 9% for all self-sustained flames. A distinct transition to-411 wards counter-gradient transport with increasing $S_{T,l}$ and thermal expansion 412 ratio is established and influenced by the rate of compressive strain. Thus, 413 mixtures with a high $S_{T,l}$ detach further from the stagnation plane and an-414 chor in low compressive strain regions where dilatation is more effective and 415 lead to a transition towards counter-gradient transport. The transition is 416 suppressed by the high turbulence levels. This is supported by a correlation 417 coefficient $(R^2) > 0.93$ for the scalar flux conditioned on the centre of the 418 turbulent flame brush as a function of mean normal compressive strain $(\overline{a_n})$ 419 acting on the flame front. The correlation $S_{T,l} - \overline{a_n}$ also shows a reason-420 able correlation. The results provide guidance to engine manufacturers and 421

inform risk assessments as to the use of hydrogen enriched mixtures. The
wide range of conditions further presents an excellent challenge for turbulent
combustion models that aim to delineate the influence of fuel reactivity.

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Table 1: Data summary of H₂ / CH₄ and H₂/CO mixtures, where β is defined in Eq. (1), Φ is the equivalence ratio, ρ_r the density of the reactants (r), ν_r the kinematic viscosity, Le_r the Lewis number, $\tau = (T_{ad} - T_r)/T_r$ the heat release parameter, $S_{L,0}$ and S_L the unstrained and strained laminar burning velocity, δ_f the laminar flame thickness and a_q the laminar rate of strain at extinction.

Mixture	β	Φ	$ ho_r$	$\nu_r \times 10^{-5}$	Le_r	au	$S_{L,0}$	S_L	δ_f	a_q
${\rm H}_2$ / ${\rm CH}_4$	[-]	[-]	$[kg/m^3]$	$[m^2/s]$	[-]	[-]	[m/s]	[m/s]	[mm]	$[s^{-1}]$
$\overline{100\%/0\%}$	1.0	0.35	1.04	1.78	0.374	3.33	0.076	0.44	1.19	3600
90%/10%	0.69	0.35	1.08	1.71	0.438	3.22	0.032	0.33	2.59	2000
80%/20%	0.50	0.35	1.10	1.68	0.500	3.13	0.016	0.26	4.42	800
80%/20%	0.50	0.50	1.07	1.72	0.527	4.10	0.164	0.52	0.703	4000
70%/30%	0.37	0.50	1.09	1.68	0.585	4.04	0.121	0.39	0.821	2400
70%/30%	0.37	0.60	1.07	1.70	0.600	4.55	0.240	0.55	0.515	4300
60%/40%	0.27	0.50	1.10	1.66	0.641	3.99	0.096	0.35	0.971	1400
60%/40%	0.27	0.60	1.09	1.67	0.654	4.52	0.194	0.46	0.588	2900
60%/40%	0.27	0.70	1.08	1.69	0.666	4.93	0.308	0.61	0.436	4300
50%/50%	0.20	0.60	1.11	1.65	0.707	4.49	0.164	0.39	0.651	1900
50%/50%	0.20	0.70	1.09	1.66	0.717	4.91	0.263	0.52	0.480	3100
50%/50%	0.20	0.80	1.08	1.67	0.727	5.24	0.364	0.66	0.393	4000
40%/60%	0.14	0.70	1.11	1.64	0.768	4.89	0.231	0.46	0.517	2300
40%/60%	0.14	0.80	1.10	1.65	0.775	5.22	0.322	0.57	0.420	3000
20%/80%	0.06	0.80	1.12	1.61	0.869	5.19	0.267	0.46	0.468	1900
$0\% \ /100\%$	0.00	0.80	1.14	1.58	0.962	5.09	0.228	0.39	0.511	1300
$\overline{\mathrm{H}_2 \ / \ \mathrm{CO}}$										
$\overline{100\%/0\%}$	1.0	0.35	1.04	1.78	0.374	3.34	0.077	0.44	1.19	3600
90%/10%	0.90	0.35	1.05	1.75	0.467	3.37	0.072	0.41	1.25	3500
80%/20%	0.80	0.35	1.07	1.73	0.554	3.40	0.069	0.38	1.31	2700
80%/20%	0.80	0.40	1.05	1.75	0.567	3.76	0.142	0.47	0.786	4200
70%/30%	0.70	0.35	1.08	1.71	0.635	3.44	0.065	0.34	1.37	1900
70%/30%	0.70	0.40	1.07	1.73	0.648	3.80	0.130	0.43	0.839	3200
60%/40%	0.60	0.40	1.08	1.70	0.722	3.83	0.120	0.39	0.895	2300
60%/40%	0.60	0.45	1.07	1.72	0.734	4.15	0.191	0.48	0.672	3500
50%/50%	0.50	0.40	1.10	1.68	0.788	3.87	0.110	0.35	0.955	1600
50%/50%	0.50	0.45	1.09	1.69	0.800	4.19	0.173	0.43	0.724	2500
50%/50%	0.50	0.50	1.08	1.70	0.811	4.48	0.246	0.52	0.596	3500
40%/60%	0.40	0.50	1.10	1.67	0.866	4.51	0.219	0.46	0.646	2500
40%/60%	0.40	0.60	1.09	1.69	0.883	4.97	0.366	0.63	0.508	4200
30%/70%	0.30	0.50	1.12	1.65	0.911	4.54	0.193	0.39	0.708	1700
30%/70%	0.30	0.60	1.11	1.66	0.924	5.00	0.319	0.56	0.556	2900

Table 2: Result summary for H_2 / CH_4 and H_2 / CO fuel mixtures, where Φ_{UN} is the upper nozzle equivalence ratio, Re_t the turbulent Reynolds number, Da the Damköhler number, u'_r the axial reactant velocity fluctuation at the nozzle exit, u'_l the axial velocity fluctuation at the leading edge ($\bar{c} = 0.02$ iso-contour), \hat{u}' the maximum axial velocity fluctuation in the flame brush, $\overline{u_t}$ the mean velocity at the trailing edge ($\bar{c} = 0.95$ iso-contour), $S_{T,l}$ the measured turbulent burning velocity at the leading edge ($\bar{c} = 0.02$ iso-contour) and S_T^* is the KPP-type turbulent burning velocity).

Mixture	Φ_{UN}	Re_t	Da	u'_r	u'_l	\widehat{u}'	$\overline{u_t}$	$S_{T,l}$	S_T^*
H_2 / CH_4	[-]	[-]	[-]	[m/s]	[m/s]	[m/s]	[m/s]	[m/s]	[m/s]
100%/0%	0.35	307	1.03	1.60	1.87	1.96	0.846	5.51	4.65
90%/10%	0.35	294	0.384	1.47	1.88	1.96	0.154	4.64	3.90
80%/20%	0.35	339	0.157	1.68	2.01	2.19	-0.0880	4.76	3.55
80%/20%	0.50	314	2.09	1.61	1.98	1.99	0.669	5.20	5.03
70%/30%	0.50	299	1.44	1.48	1.87	1.97	0.169	4.66	3.97
70%/30%	0.60	302	3.13	1.55	1.94	1.98	1.11	5.74	4.95
60%/40%	0.50	300	1.11	1.47	1.91	2.04	0.0056	4.71	3.74
60%/40%	0.60	320	2.22	1.63	1.96	1.98	0.338	5.06	4.41
60%/40%	0.70	334	3.74	1.71	2.07	2.13	0.545	6.22	5.37
50%/50%	0.60	315	1.76	1.49	1.93	2.01	-0.0114	4.56	3.89
50%/50%	0.70	310	3.20	1.59	2.01	2.04	0.585	5.25	4.69
50%/50%	0.80	302	5.11	1.48	1.98	1.98	1.58	5.47	5.27
40%/60%	0.70	321	2.56	1.58	2.00	2.02	0.124	5.52	4.28
40%/60%	0.80	317	3.96	1.52	1.93	2.00	0.722	5.28	4.66
20%/80%	0.80	327	2.83	1.56	2.03	2.09	0.164	5.04	4.16
0% / 100%	0.80	324	2.27	1.51	2.03	2.17	-0.0145	4.41	3.64
H_2 / CO									
100%/0%	0.35	307	1.03	1.60	1.87	1.96	0.846	5.51	4.85
90%/10%	0.35	289	0.985	1.48	1.93	2.01	0.413	5.53	4.42
80%/20%	0.35	306	0.833	1.54	1.93	1.96	0.266	5.23	4.08
80%/20%	0.40	303	1.70	1.52	1.93	1.97	0.713	5.43	4.56
70%/30%	0.35	307	0.719	1.50	1.96	1.99	0.0462	4.81	3.77
70%/30%	0.40	314	1.44	1.51	1.98	2.06	0.435	5.48	4.29
60%/40%	0.40	323	1.20	1.56	1.99	2.10	0.193	5.48	3.96
60%/40%	0.45	313	2.03	1.54	1.99	2.08	0.544	5.92	4.41
50%/50%	0.40	381	0.876	1.95	2.23	2.24	0.0958	5.03	4.02
50%/50%	0.50	358	2.16	1.88	2.11	2.13	1.67	5.07	4.68
40%/60%	0.50	347	1.85	1.81	1.97	2.08	1.56	5.27	4.04
40%/60%	0.60	338	3.29	1.75	2.03	2.04	3.33	5.64	4.89
30%/70%	0.50	342	1.48	1.72	2.03	2.05	0.465	4.88	3.71
30%/70%	0.60	332	2.76	1.66	1.89	1.97	1.47	5.08	4.24