The Reactivity of Hydrogen Enriched Turbulent Flames

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Abstract

The use of hydrogen enriched fuel blends, e.g. syngas, offers great potential in the decarbonisation of gas turbine technologies by substitution and expansion of the lean operating limit. Studies assessing explosion risks or laminar flame properties of such fuels are common. However, there is a lack of experimental data that quantifies the impact of hydrogen addition on turbulent flame parameters including burning velocities and scalar fluxes. Such properties are here determined for aerodynamically stabilised flames in a back-to-burnt opposed jet configuration featuring fractal grid generated multi-scale turbulence $(Re_t = 314 \pm 19)$ using binary $\rm H_2$ / CH₄ and $\rm H_2$ / CO fuel blends. The binary $\rm H_2$ / CH₄ fuel blend is varied from $\alpha = X_{H2}/(X_{H2} + X_F) = 0.0, 0.2$ and $0.4 - 1.0$, in steps on 0.1, and the binary H₂ / CO fuel blend from $\alpha = 0.3 - 1.0$ also in steps of 0.1. The equivalence ratio is adjusted between the mixture specific lower limit of local flame extinction and the upper limit of flashback. The flames are characterised using PIV measurements combined with a flame front detection algorithm. The study quantifies the impact of hydrogen enrichment on (i) turbulent burning velocity (S_T) , (ii) turbulent transport and (iii) the rate of strain acting on flame fronts. Scaling relations (iv) that correlate S_T with laminar flame properties are evaluated and (v) flow field data that permits validation of computational models is provided. It is shown that $CH₄$ results in a stronger inhibiting effect on the reaction chemistry of $H₂$ compared to CO, that turbulent transport and burning velocities are strongly correlated with the rate of compressive strain and that scaling relationships can provide reasonable agreement with experiments.

Keywords: Hydrogen, Turbulent Premixed Combustion, Syngas, Turbulent Burning Velocity, Scalar Transport, Rate of Strain

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1. Introduction

The confluence of climate change, environmental protection and dimin- ishing fossil fuel resources have promoted the development of low carbon footprint and clean energy technologies [1]. State-of-the-art gas turbines operate under fuel lean conditions optimised for natural gas [2] and single digit nitrogen oxide emissions have been achieved. The use of hydrogen en- riched fuel blends, e.g. syngas, offers great potential in the decarbonisation of related technologies by substitution and expansion of the lean operating limit [3]. Decarbonisation by means of hydrogen substitution is most ef- fective if hydrogen is produced from renewable energy sources. However, the variability of syngas compositions can lead to fuel flexibility problems for manufacturers [4] and the increased hydrogen concentration to safety concerns [5]. The syngas mixture reactivity is strongly dependent upon hy- drogen content with direct implications on flame propagation speeds, ex- plosion over-pressures, auto-ignition and turbulence–chemistry interactions. The latter can lead to differences in flame surface area [6] and combustion instabilities [1, 4]. Li et al. [5] measured flame speeds and over-pressures generated in an obstructed flame tube for a wide range of binary and ternary $_{19}$ H₂, CO and CH₄ mixtures, with methane showing a stronger inhibiting effect on the mixture reactivity. Scaling based on the amount of air required to fully oxidise the mixture correlated the fuel composition impact on explosion 22 over-pressures $[5]$ and turbulent flow fields $[7]$. Simatos *et al.* $[8]$ investigated ²³ the effect of H₂ content in lean ($\Phi = 0.80$) binary CH₄ and CO fuel blends on auto-ignition in turbulent shear layers and the stronger inhibiting effect of $CH₄$ prevailed.

²⁶ The turbulent burning velocity (S_T) is a key parameter required, for ex-²⁷ ample, to evaluate the strength of explosions [5] and the flashback propensity ²⁸ in gas turbine engines [9]. Investigations seeking to correlate S_T to the tur-29 bulence intensity are quite common [10]. Wang et al. [11] investigated the 30 effect of hydrogen content on S_T and explored scaling relations under gas 31 turbine relevant conditions, and found that H_2 addition alters the correla-³² tion due to preferential diffusive-thermal effects [2]. Scaling relations for the 33 turbulent consumption speed of H_2 /CO mixtures can include flame stretch 34 and pressure effects as shown by Venkateswaran *et al.* [12] for a wide range of 35 conditions. Shy *et al.* [13] evaluated the effect of turbulent Reynolds (Re_t) , ³⁶ Damköhler (*Da*) and Karlovitz (*Ka*) numbers for lean syngas flames. Zhang ³⁷ et al. [14] evaluated a Kobayashi type turbulent burning velocity correlation 38 for mixtures including H_2 / CO blends with reasonable agreement. Daniele 39 et al. [15] showed a linear correlation of S_T with equivalence ratio and inlet ⁴⁰ bulk velocity for various syngas related mixtures. The impact of hydrogen ⁴¹ blending on methane was found to be particularly prominent under ultra-lean 42 conditions. The Lewis number effect on S_T was also found to be significant 43 in binary H_2 / CH_4 and propane mixtures [16] and the effect of pressure was 44 investigated by Liu *et al.* [17]. Multiple turbulent burning velocity defini-⁴⁵ tions (e.g. Bray [18]) have been advanced and Driscoll [19] has shown that ⁴⁶ the definitions of Shepherd and Cheng [20] and Lawn and Schefer [21] are 47 not equivalent. Yet, the S_T definitions of Bray [18], Lawn and Schefer [21] ⁴⁸ and a variant proposed by Driscoll [19] agree within $\pm 20\%$ for some pre-⁴⁹ mixed twin opposed jet flames [22]. However, a broader understanding of the ⁵⁰ impact of mixture reactivity on turbulent flame propagation is lacking. An

51 additional difficulty is that theories for premixed flames with a $Da \simeq 1$ are 52 not well developed in contrast to the corrugated flamelet regime $(Da \gg 1)$. The current study provides an extensive data set that establishes the impact 54 of fuel reactivity on S_T at $Da \simeq 1$ to support the removal of such limitations.

 The opposed jet configuration provides a canonical geometry with dis- tinct advantages for the assessment of fuel effects [23]. The geometry fea- tures aerothermochemistry related flame stabilisation [24], essentially adia- batic conditions [25], comparatively well developed turbulence [26], excellent optical access [27] and accurate control of boundary conditions [23]. The tur- bulent to bulk strain ratio can be substantially increased without bulk flow $\epsilon_{\rm i}$ instabilities [23] via the use of cross fractal grids (CFGs) [26]. The back-to- burnt (BTB) configuration further allows the stabilisation of flames at low ω Damköhler numbers and permits investigations of combustion regime transi- $\frac{64}{64}$ tions [24, 28]. In particular, self-sustained flames detach from the stagnation ϵ ₆₅ plane and become independent of the opposing burnt gas [29], while low Da combustion is dominated by interactions between the two streams [30, 31].

 The current study is using Particle Image Velocimetry (PIV) and a flame 68 front detection algorithm [23, 29] to quantify the impact of H_2 enrichment on CH₄ and CO fuel blends by determining the impact on (i) turbulent burning τ_0 velocities (S_T) , (ii) turbulent transport and (iii) the rate of strain acting on flame fronts. Scaling relations (iv) that correlate S_T with laminar flame prop- erties are evaluated and (v) flow field data for validation of computational fluid mechanics (CFD) based models for risk assessment is provided.

⁷⁴ 2. Experimental configuration

⁷⁵ The present opposed jet configuration was originally developed by Geyer τ ⁶ *et al.* [27] and has been used by Geipel *et al.* [26], Goh *et al.* [22, 25] and π Hampp and Lindstedt [24, 29, 30, 32]. A schematic is provided in Fig. 1. ⁷⁸ The burner consists of two identical nozzles with an inner diameter $D =$ $\frac{79}{20}$ mm, nozzle length of $L = 50$ mm and nozzle separation of one D. The ⁸⁰ burner is operated in a back-to-burnt (BTB) configuration to accommodate ⁸¹ the wide range of mixture reactivities. The configuration entails considerable ⁸² advantages in the current context: (i) cases can be compared on a basis of ⁸³ nearly identical flow conditions [29], (ii) all fuel gases are combusted (a safety $\frac{84}{4}$ requirement for H₂ and CO containing mixtures) [5] as global extinction is ⁸⁵ prevented, (iii) the transition from self-sustained flame propagation to ther-⁸⁶ mally supported burning can be assessed [24] and (iv) self-sustained flames ⁸⁷ detach from the stagnation plane with combustion dynamics related to the ⁸⁸ intrinsic aerothermochemistry. A purpose written LabView interface is used ⁸⁹ to control the Bronkhorst mass flow controllers with a maximum uncertainty ⁹⁰ of \pm 0.5% of full scale. All gases are supplied at a pressure of 4.0 bar(g) with 91 the upper nozzle air seeded with $\sim 3 \mu m$ Al₂O₃ particles [25].

92

⁹³ 2.1. Flow conditions

⁹⁴ In order to determine the impact of fuel reactivity on turbulent burning 95 velocities and scalar transport, the binary H_2 / CH_4 and H_2 / CO fuel blends 96 are varied from $\alpha = X_{H2}/X_F = 1.0, 0.9, 0.8, 0.7, 0.6, 0.5, 0.4, 0.2$ and 0 and $97 \quad 1.0, 0.9, 0.8, 0.7, 0.6, 0.5, 0.4$ and 0.3, respectively, where X is the fuel mole

Figure 1: Schematic of the back-to-burnt opposed jet burner configuration. CFG – cross fractal grid, DSI – density segregation iso-contour, $SP -$ stagnation plane, (x,y) – coordinate system, HCP – hot combustion products, PP – perforated plate.

106 The turbulent Reynolds number $(Re_t = L_I \cdot u_{rms}/\nu_r = 314 \pm 19)$ is modestly 107 affected by fuel composition changes, primarily due to the kinematic viscos-98 fraction in the blend and F the blending component (i.e. CH_4 or CO). The ⁹⁹ mixture compositions are listed in the supplementary material. The premixed ¹⁰⁰ fuel / air mixtures are injected through the upper nozzle with a constant bulk ¹⁰¹ velocity of $U_b = 9.0 \text{ ms}^{-1}$ and a reactant temperature of $T_r = 298 \text{ K}$. A cross ¹⁰² fractal grid (CFG) is used to provide a well developed multi-scale turbulent ¹⁰³ flow with enhanced intensity [22]. The flow is maintained constant with an 104 integral length scale of turbulence of $L_I = 3.9 \pm 0.2$ mm [30]. Velocity fluctuations at the nozzle exit of $u_{rms} = 1.4 \pm 0.1$ ms⁻¹ are measured using PIV. 108 ity of the reactants (ν_r) . The turbulent kinetic energy dissipation rate in the reactants is determined as $\epsilon_r = u_{rms}^3 L_I^{-1} \simeq 700 \text{ m}^2 \text{s}^{-3}$.

¹¹⁰ The lower nozzle hot combustion products (HCP) are generated from lean ¹¹¹ ($\Phi = 0.60$) premixed 50% H₂ / 50% CH₄ flames stabilised on a perforated ¹¹² plate. The HCP are in close-to thermochemical equilibrium with a nozzle exit 113 temperature of 1640 ± 7.1 K measured using a 50 μ m R-type thermocouple. The HCP composition, including the residual oxygen concentration, does not exert a strong impact on the self-sustained flames [28, 30, 31] of primary interest here.

2.2. Diagnostic setup

 The flow field is measured by PIV using a double pulse Nd:YAG laser. 119 The laser sheet thickness is estimated to ~ 0.3 mm using burn marks with ¹²⁰ an optimum time delay between pulses of $\Delta t = 25 \mu s$. The interrogation 121 region of 40×30 mm is resolved by 1376 \times 1040 pixels using a CCD- camera. A 100 mm Nikon lens is equipped with a 99% neutral density and a 3 nm bandpass filter centred at 532 nm to minimise noise. The PIV vectors are calculated via a multi-pass cross-correlation with decreasing window size resulting in a spatial vector resolution of 0.90 mm and vector spacing of 0.45 mm. For each set of conditions, 1000 double frame images are recorded 127 to assure statistically independent data. The relaxation time (τ_p) of the ¹²⁸ 3 μ m Al₂O₃ seeding particles is estimated to $\tau_p = 107 \mu s$ which is similar to 129 the smallest PIV timescale of 102 μ s. The Stokes number ($St = \tau_p/\tau_\eta$) is 130 determined to $St \simeq 0.75$ based on a Kolmogorov timescale of $\tau_{\eta} \simeq 156 \,\mu s$. Out-of-plane, out-of-pattern, beam steering and uncertainties due to thermal gradients are negligible [24, 29].

3. Data analysis

 The following section introduces the utilised scaling relations, evaluated turbulent flame properties and uncertainty analysis.

¹³⁶ 3.1. Reaction progress and scalar transport

¹³⁷ A conventional PIV particle based density segregation technique is used to ¹³⁸ detect the instantaneous flame front [23, 33]. The accuracy of the algorithm ¹³⁹ was evaluated by Hampp and Lindstedt [24] and found to deviate less than ¹⁴⁰ two Kolmogorov length scales from the 600 K iso-contour (e.g. Schlieren ¹⁴¹ contour). The turbulent flame brush and the reaction progress variable (\bar{c}) ¹⁴² are determined from the statistical location of the instantaneous flame front ¹⁴³ (c_i), while the measured turbulent burning velocity $(S_{T,l})$ is determined based ¹⁴⁴ on the leading edge ($\bar{c} = 0.02$) [21]. The scalar flux is evaluated as $\bar{c'}\bar{u'} =$ $\overline{c} \cdot (1 - \overline{c}) \cdot \overline{U_s}$, where the slip velocity is defined as the difference between 146 product and reactant fluid velocities, i.e. $\overline{U_s} = \overline{U_p} - \overline{U_r}$ [34].

¹⁴⁷ 3.2. Conditional rate of strain

¹⁴⁸ The normal rate of strain conditioned on the instantaneous flame front ¹⁴⁹ is calculated using the methodology of Hampp and Lindstedt [29, 30]. The 150 instantaneous planar strain rate tensor $(e_{ij} = 0.5(\partial u_i/\partial x_j + \partial u_j/\partial x_i))$ is ¹⁵¹ calculated from the PIV data. The normal $(a_n = f_{11})$ strain component is 152 determined by $f = \mathbf{R}(\Theta) \cdot e$, where **R** is the rotation matrix and Θ is the ¹⁵³ angle between the iso-contour normal and the theoretical stagnation point ¹⁵⁴ streamline (SPS). The analysis includes the stagnation point movement in ¹⁵⁵ the radial limit \pm 0.5 L_I away from the SPS [26]. Correlations with mean ¹⁵⁶ values of the conditional normal compressive strain $(\overline{a_n})$ are discussed below.

157 3.3. Scaling relations for mixtures with high H_2 content

 158 The wide range of H_2 concentrations results in significant differences in the 159 mixture reactivity as well as reactant and burning properties. Li et al. $[5, 7]$

 $_{160}$ presented a scaling factor for fuel lean blends of H_2 with CH_4 and CO defined ¹⁶¹ based on the amount of air required to fully oxidise each fuel component (0 $162 \leq \beta \leq 1$. The β factor improved the scaling of explosion over-pressures and 163 related flow velocities compared to α (= $X_{H2}/(X_{H2} + X_F)$) for a wide range $_{164}$ of binary and ternary H₂ / CH₄ / CO mixtures. The scaling of auto-ignition 165 in a turbulent shear layer was also improved [8] and β (see Eq. (1), where ¹⁶⁶ $(X_k/X_A)_{st}$ is the stoichiometric fuel air ratio) is used here.

$$
\beta = \left(\frac{X_{H2}}{(X_{H2}/X_A)_{st}}\right) / \left(\frac{X_{H2}}{(X_{H2}/X_A)_{st}} + \frac{X_F}{(X_F/X_A)_{st}}\right)
$$
(1)

¹⁶⁷ 3.4. Turbulent burning velocity scaling relationships

 Classical theories for turbulent combustion resulting in eddy breakup based models for the reaction rate source term (e.g. [35]) only provide a scaling of the turbulent burning velocity based on the (isotropic) velocity 171 fluctuations (e.g. u') as shown via Kolmogorov, Petrovskii and Piskunov (KPP) [36] or eigenvalue [37] analyses. The velocity fluctuations are kept constant in the current study to isolate the impact of the mixture reactivity. The classical form for expressing the turbulent burning velocity is:

$$
\frac{S_T}{S_L} = 1 + c \left(\frac{u'_l}{S_L}\right)^m \text{ with } 0.5 \le m \le 1 \tag{2}
$$

175 Examples of correlations (e.g. [10, 11, 13, 14]) for S_T from Zhang *et al.* [14] $(S_T^{\#})$ $g_T^{(\#})$, Bradley *et al.* [38] (S_T^{\ddagger}) $T(T)$ and Peters [39] (S_T^{\dagger}) ¹⁷⁶ $(S_T^{\#})$, Bradley *et al.* [38] $(S_T^{\{1\}})$ and Peters [39] $(S_T^{\{1\}})$ are given in Eq. (3),

$$
\frac{S_T^{\#}}{S_L} = 2.92 \left(\frac{u'_l}{S_L}\right)^{0.55} \n\frac{S_T^{\ddagger}}{S_L} = \frac{0.88}{(KLe_r)^{0.3}} \left(\frac{u'_l}{S_L}\right) \text{ where } K = 0.157 \left(\frac{u'_l}{S_L}\right)^2 Re_t^{-1/2} \n\frac{S_T^{\dagger}}{S_L} = 1 - \frac{\psi L_I}{4\delta_f} + \left[\left(\frac{\psi L_I}{4\delta_f}\right)^2 + \psi \frac{u'_l L_I}{S_L \delta_f}\right]^{\frac{1}{2}}
$$
\n(3)

¹⁷⁷ In Eq. (3), $\psi = 0.78$, u'_{l} is the axial velocity fluctuation at the leading edge, ¹⁷⁸ δ_f the laminar flame thickness, K the Karlovitz stretch factor and Le_r the ¹⁷⁹ Lewis number (ratio of Schmidt and Prandtl numbers) of the reactants. The 180 expression for Le_r is given in Eq. (4) and assumes that the deficient reactant ¹⁸¹ is the dominant fuel component for the current fuel lean mixtures.

$$
Le_r = \frac{X_{H_2}Le_{H_2} + X_FLe_F}{X_{H_2} + X_F} \tag{4}
$$

¹⁸² A fractal flame front analysis has been used to show that the rate of 183 reaction depends on the ratio of the laminar burning (S_L) and Kolmogorov ¹⁸⁴ $(V_{\kappa} = (\nu_r \epsilon_r)^{1/4}$) velocities [40]. Such forms are evidently more applicable ¹⁸⁵ in the current context where fuel reactivity comes to the fore. The applied ¹⁸⁶ KPP analysis [41] provides the estimate of the turbulent burning velocity ¹⁸⁷ (S_T^*) shown in Eq. (5). The form includes the KPP limit eigenvalue ($\Lambda = 2$), ¹⁸⁸ the customary turbulent eddy viscosity $C_{\mu} = 0.09$ and reaction rate $C_R \simeq 4$ ¹⁸⁹ constants [42]. The Le_r correction by Aluri *et al.* [43] gives $C_R = 4.0/e^{Le_r-1}$.

$$
\frac{S_T^*}{S_L} = 1 + \Lambda \cdot \sqrt{\frac{3}{2} C_R \frac{C_\mu}{S_c} \frac{S_L}{V_\kappa}} \cdot \left(\frac{u'_l}{S_L}\right) \tag{5}
$$

190 The turbulent Schmidt number (Sc_t) is also required and depends greatly on

191 the flow configuration. It is typically in the range $0.3 \leq Sc_t \leq 1$ [44] and ¹⁹² a common value $Sc_t = 0.7$ is used here. As discussed below, the gradient ¹⁹³ diffusion hypothesis used to derive Eq. (5) is subject to significant uncertain-¹⁹⁴ ties. Laizet and Vassilicos [45] have further shown that scalar transport can ¹⁹⁵ be significantly enhanced for fractal grids due to the multiscale nature of the 196 generated turbulence. Nevertheless, Goh *et al.* [22, 23] have shown that the ¹⁹⁷ above expression can provide reasonable agreement and it was further noted 198 that a modest reduction in Sc_t would bring improved agreement.

¹⁹⁹ 3.5. Laminar flame properties required for scaling

200 The laminar flame data, i.e. unstrained burning velocity $(S_{L,0})$, flame ²⁰¹ thickness (δ_f) and adiabatic flame temperature (T_{ad}) , required for the eval-²⁰² uation of the used scaling relationship, are obtained from freely propagating ²⁰³ flame calculations using detailed chemistry with differential diffusion effects $_{204}$ included [46]. Li *et al.* [5] validated the chemical mechanism for a wide range ²⁰⁵ of binary fuel mixtures. The inlet conditions of the laminar flame calcula-²⁰⁶ tions matched the experiment, i.e. $T_r = 298 \text{ K}$, $P_0 = 1 \text{ atm}$ and species mole ²⁰⁷ fractions. The computational domain is resolved by 550 nodes featuring a 208 mesh size of ~ 10 μm. Reactant mixture properties such as kinematic vis-²⁰⁹ cosity (ν_r) , density (ρ_r) and Lewis number (Le_r) are also inferred from the 210 calculations. The laminar extinction strain (a_q) is determined by means of ²¹¹ counterflow flame calculations. The computational domain of the latter is 212 resolved by 340 nodes with a local resolution of $\sim 20 \ \mu m$ in the reaction zone. $_{213}$ The rate of strain is increased in steps of 100 s⁻¹ until the laminar flames ²¹⁴ extinguished. The extinction strain is defined as the highest rate of strain 215 prior extinction for each mixture. The strained laminar burning velocity (S_L)

 used for the theoretical scaling relations is determined at the bulk strain rate $_{217}$ ($a_b = 600 \text{ s}^{-1}$) matching the experiment. This accounts for the differences $_{218}$ in stretch effects on S_L for the wide range of mixture reactivity investigated and improves the predictability of the scaling relations [12]. The calculation results are summarised in Table 1. The data are further used to determine ²²¹ the Damköhler number $(Da = (L_I \cdot S_L)/(u_{rms} \cdot \delta_f))$ as listed in Table 2.

3.6. Uncertainty analysis and data rejection

 The flames of interest in the current investigation are within or close-to the thin reaction zone regime. Moreover, the heat release parameter is only weakly dependent on the fuel blending fractions with similar adiabatic flame $_{226}$ temperature for a given Φ . Consequently, the effect of the fuel mixture com- position on the detected flame front and thus reaction progress variable is 228 negligible for a given stoichiometry. In the range from $\bar{c} = 0.5 \pm 0.1$, i.e. the conditioning variable, the reactant and product fluid velocity changes are < 2.5%. Consequently, a translation of the detected instantaneous iso-contour within the laminar flame thickness has a negligible impact on the measured scalar flux. A translation of the detected instantaneous iso-contour also pro- vides the largest uncertainties in the measured turbulent burning velocities and the turbulence intensity required for the determination of $S_T^{\#}$ $T^{\#}, S_T^{\ddagger}$ T^{\ddagger} , S_{T}^{\dagger} T ²³⁵ and S_T^* . Therefore, a sensitivity analysis of the leading edge position (x_l) on ²³⁶ S_T and u'_l is conducted in the range from $x_l \pm 0.25L_I$. The range corresponds to pure mixing dominated reaction progress that provides the lowest gradi- ent in the reaction progress variable and thus a conservative estimate. The turbulent intensity is approximately constant in the proximity of the leading edge with a variation below 3% for the above x_l variation. The effect of x_l

²⁴¹ on the measured S_T^* is below 10%. Thus, a shift in the reaction progress variable iso-contour, e.g. due to uncertainties in detection, fuel composition and changes in equivalence ratio, has a modest impact on the determined turbulent burning velocities.

²⁴⁵ Flame propagation in the BTB configuration can be influenced or gov-²⁴⁶ erned by the counterflow hot combustion products for cases with insufficient $_{247}$ flame detachment from the stagnation plane [30]. This can result in ques- $_{248}$ tionable S_T values as the reaction progress may become influenced by auto-²⁴⁹ ignition events. Accordingly, the mean velocity at the trailing edge $(\overline{u_t})$ of ²⁵⁰ the flame brush (i.e. at $\bar{c} = 0.95$) is evaluated and cases where $\overline{u_t}$ is negative ²⁵¹ (the flow direction of the HCP) are removed. For such cases the thermal ²⁵² support provided by the HCP can have a significant impact.

²⁵³ 4. Results and discussion

²⁵⁴ 4.1. Flow field statistics

²⁵⁵ An example of unconditional velocity statistics along the theoretical stag-256 nation point streamline is shown in Fig. 2 for the 60% $\rm H_2$ / 40% CH₄ mix-257 ture at $\Phi = 0.50$ and 0.70 and 60% H₂ / 40% CO mixture at $\Phi = 0.45$. An ²⁵⁸ increase in mixture reactivity (e.g. H_2 content or Φ) results in a faster prop-²⁵⁹ agating flame that stabilises further upstream. An increase in equivalence ²⁶⁰ ratio further leads to a stronger dilatation. The latter results in an elevated ₂₆₁ mean axial velocity (\overline{U}/U_b) and an earlier and more pronounced peak in the axial velocity fluctuations $(\sqrt{u'u'}/U_b)$ [23]. The stronger inhibiting ef- $_{263}$ fect of methane on the H_2 reaction chemistry compared to carbon monoxide, ²⁶⁴ that was also observed in turbulent explosions [5, 7] and auto-ignition re²⁶⁵ lated flame stabilisation [8], is readily evident in the unconditional velocity ²⁶⁶ statistics. This can be inferred from Fig. 2, where the unconditional velocity 267 statistics of the 60% H₂ / 40% CO mixture at $\Phi = 0.45$ is distinctly closer ²⁶⁸ to the 60% H₂ / 40% CH₄ mixture at $\Phi = 0.70$ than the $\Phi = 0.50$ flame.

Figure 2: Unconditional velocity statistics (\overline{U} mean axial velocity, u' axial fluctuation, v' radial fluctuation) along the stagnation point streamline, normalised by the bulk velocity (U_b) , for the 60% H₂ / 40% CH₄ and 60% H₂ / 40% CO mixtures at different Φ .

²⁶⁹ The corresponding conditional reactant and product fluid velocity statis-²⁷⁰ tics are depicted in Figs. 3 and 4, respectively. The advanced reaction onset $_{271}$ for the case with $\Phi = 0.70$ yields an elevated mean axial reactant veloc-²⁷² ity $(\overline{U_r}/U_b)$ in the proximity of $x/D = 0$. As the turbulent flow field is ²⁷³ maintained constant, the conditional axial and radial reactant velocity fluc-²⁷⁴ tuations are very similar. The latter also highlights the good experimental ²⁷⁵ repeatability and accurate control of boundary conditions that is essential ²⁷⁶ for the comparisons with numerical investigations. The enhanced dilation ²⁷⁷ of the $\Phi = 0.70$ case leads to the higher mean axial product fluid velocity ₂₇₈ and reduced fluctuations close to the reaction onset (i.e. $x/D \simeq 0$). The

Figure 3: Conditional reactant fluid velocity statistics (\overline{U} mean axial velocity, u' axial fluctuation, v' radial fluctuation) along the stagnation point streamline, normalised by the bulk velocity (U_b) , for the 60% H₂ / 40% CH₄ and 60% H₂ / 40% CO mixtures at different equivalence ratios.

²⁷⁹ 60% H₂ / 40% CO at $\Phi = 0.45$ shows conditional reactant and product fluid 280 velocities and fluctuations similar to the $\Phi = 0.70$ mixture with CH₄. This ²⁸¹ can be attributed to the advanced reaction onset and flame anchoring in low ²⁸² strain regions where dilatation is more effective.

²⁸³ 4.2. Turbulent burning velocity

²⁸⁴ The measured leading edge ($\bar{c} = 0.02$ iso-contour) [21] turbulent burning 285 velocity $(S_{T,l})$ is depicted for all H_2 / CH_4 mixtures in the top panel of Fig. 5 286 as a function of β . As the amount of CH₄ is increased, a higher equivalence ²⁸⁷ ratio is required to stabilise a self-propagating flame that detaches from the ²⁸⁸ stagnation plane. For example, the pure H₂ / air flame is stabilised at $\Phi =$ $_{289}$ 0.35 (i.e. upper limit to avoid flashback). A CH₄ blending of 20% and 50% $_{290}$ results in a significantly decreased reactivity, a strong reduction of $S_{T,l}$ and ²⁹¹ increased upper Φ limits of 0.50 and 0.80, respectively. Scaling of $S_{T,l}$ with

Figure 4: Conditional product fluid velocity statistics (\overline{U} mean axial velocity, u' axial fluctuation, v' radial fluctuation) along the stagnation point streamline, normalised by the bulk velocity (U_b) , for the 60% H₂ / 40% CH₄ and 60% H₂ / 40% CO mixtures at different equivalence ratios.

the axial velocity fluctuations collapses the data to $S_{T,l}/\sqrt{u_l'u_l'} = 2.68 \pm 0.19$. ²⁹³ By omitting cases that are influenced by the HCP (i.e. grey data points in ²⁹⁴ Fig. 5) the predictive capability of the KPP type velocity in Eq. (5) over the current range is reasonable with $S_{T,l}/S_T^* = 1.16 \pm 0.07$. Scaling the 296 data with the correlations from Peters [39], Bradley *et al.* [38] and Zhang *et* ²⁹⁷ al. [14] result in $S_{T,l}/S_T^{\dagger} = 2.83 \pm 0.65$, $S_{T,l}/S_T^{\ddagger} = 1.64 \pm 0.11$, $S_{T,l}/S_T^{\#} =$ $_{298}$ 1.95 \pm 0.11, respectively. The KPP derived expression also shows reasonable agreement with $S_{T,l}/S_T^* = 1.24 \pm 0.09$ over the entire H₂ / CO data range. ³⁰⁰ Scaling by $\sqrt{u'_l u'_l}$ results in 2.66 \pm 0.20 and the correlations from Peters [39], 301 Bradley *et al.* [38] and Zhang *et al.* [14] result in $S_{T,l}/S_T^{\dagger} = 3.16 \pm 0.57$, 302 $S_{T,l}/S_T^{\ddagger} = 1.76 \pm 0.16$, $S_{T,l}/S_T^{\#} = 2.02 \pm 0.21$, respectively.

³⁰³ The level of disagreement obtained with some of the scalings is not sur-³⁰⁴ prising. The opposed jet configuration features an imposed pressure gradient

Figure 5: Leading edge turbulent burning velocity $(S_{T,l})$ for H₂ / CH₄ mixtures as a function of β . Top: Measured $S_{T,l}$; Bottom: $S_{T,l}/S_T^*$. The grey symbols indicate thermally supported cases.

 and a complex transition between gradient and non-gradient transport. In addition, it has been shown that pressure transport, dilation and scram- bling terms play a significant role [47] in the flamelet regime. Furthermore, the observed trend is in agreement with the observations of Laizet and Vas- silicos [45] that scalar transport is significantly enhanced by fractal grids. 310 Accordingly, the systematic deviation $\simeq 21\%$ obtained with the KPP de- rived scaling expression obtained with $Sc_t = 0.7$ appears modest and can be 312 further reduced to $\simeq 10\%$ for $Sc_t = 0.5$.

³¹³ The influence of CO blending on the mixture reactivity is strongly re- $_{314}$ duced compared to CH₄, in particular for lean cases as illustrated in Fig. 6. 315 For example, a mixture with $\Phi = 0.35$ can accommodate a 30% CO sub-316 stitution while maintaining a self-sustained flame. By contrast, a $20\% \text{ CH}_4$ ³¹⁷ substitution results in thermally supported burning. The turbulent burning $_{318}$ velocity reduces by $\sim 0.02 \text{ ms}^{-1}$ per % of CO substitution and approximately 319 twice this value for CH₄ substitution in the range $80 \leq$ H₂ (%) ≤ 100 content.

Figure 6: Leading edge turbulent burning velocity $(S_{T,l})$ for H₂ / CO mixtures as a function of β . Top: Measured $S_{T,l}$; Bottom: $S_{T,l}/S_T^*$.

320 This strong difference is also evident at $\Phi = 0.50$, where $S_{T,l}$ reduces by 0.01 $_{221}$ and 0.02 ms⁻¹ per percent CO or CH₄ substitution corresponding to a varia-³²² tion from $0.3 \le \beta \le 0.5$ for both blends. The strong inhibiting effect of CH₄ 323 on the H_2 reaction chemistry compared to CO was also evident in laminar 324 flames, explosion over pressures and fast turbulent deflagrations [5, 7] as well ³²⁵ as auto-ignition in a turbulent shear layer [8].

³²⁶ 4.3. Reaction progress conditions

327 The conditional reactant $(\overline{U_r})$, product $(\overline{U_p})$ and slip velocity $(\overline{U_s}/U_b)$ are α depicted along with the axial scalar flux $\left(\overline{c'u'}\right)$ across the turbulent flame 329 brush in Fig. 7 for the 60% H₂ mixture at $\Phi = 0.50$ and 0.70 for CH₄ and $\Phi = 0.45$ for CO. The $\Phi = 0.70$ mixture exhibits a higher $S_{T,l}$, an earlier reaction onset and thus a higher reactant velocity throughout the entire flame brush compared to the leaner mixtures. This leads to a higher product fluid velocity, advanced dilatation and consequently a less negative slip velocity and reduced gradient scalar flux. The reduced inhibiting effect of CO on the

Figure 7: Normalised slip velocity $(\overline{U_s})$ and scalar flux (\overline{cu}) as a function of \overline{c} for the mixture 60% H₂ / 40% CH₄ or CO. \Box CH₄ $\Phi = 0.50$; \circ CH₄ $\Phi = 0.70$; \times CO $\Phi = 0.45$.

335 H₂ reactivity leads to a reduced gradient scalar flux for the CO ($\Phi = 0.45$) 336 compared to the CH₄ ($\Phi = 0.50$) mixture.

337 Tian and Lindstedt [47] showed that the product τS_L , where $\tau = (T_{ad} -$ 338 T_r / T_r is the heat release parameter, scales the pressure transport, dilatation 339 and scrambling terms in the flamelet regime of combustion, while Li et al. [5] 340 showed that a normalised scaling based on $(\tau+1)S_L$ can successfully correlate $_{341}$ syngas explosion data. The influence of τS_L on the scalar flux conditioned on ³⁴² the $\bar{c} = 0.50$ iso-contour is shown in Fig. 8 for varying H₂ concentration and ³⁴³ Φ while maintaining the other parameters constant to quantify their impact. 344 The Φ variation is performed for the 60% H₂ fuel blends in the range 0.50 $345 \leq \Phi \leq 0.70$ for CH₄ and $0.40 \leq \Phi \leq 0.45$ for CO due to the number of 346 available points. For a given lean mixture, an increase in Φ exhibits a two-³⁴⁷ fold effect that influences the scalar flux: (i) a higher burning velocity due ³⁴⁸ to the elevated mixture reactivity and (ii) an enhanced dilatation due to the higher heat release parameter. For CH₄ / H₂, $S_{T,l}$ increases from 4.7 to

350 6.2 ms⁻¹ and τS_L from 1.4 to 3.0 ms⁻¹ with Φ . For CO / H₂, $S_{T,l}$ increases σ_{351} from 5.5 to 5.9 ms⁻¹ and τS_L from 1.5 to 2.0 ms⁻¹. The scalar flux in ³⁵² the centre of the turbulent flame brush decreases accordingly (i.e. towards counter-gradient transport) from –0.41 to –0.27 ms⁻¹ and –0.48 to –0.37 ms⁻¹ 353 354 with increasing Φ for CH₄ and CO blends, respectively. The H₂ variation for ³⁵⁵ the CH₄ and CO blends is performed at $\Phi = 0.80$ and 0.35, representing the 356 highest and lowest heat release limits. The addition of H_2 to CH_4 / (CO) 357 yields a pronounced / (modest) increase of τS_L of $25 \times 10^{-3} \text{ ms}^{-1}$ / (8.7 10^{-3} ms⁻¹) per percent H₂ substitution for the given Φ . Moreover, the $_{359}$ measured $S_{T,l}$ increases from 4.4 to $5.5~\mathrm{ms^{-1}}$ by moving from $100\%~\mathrm{CH_4}$ to the 360 50% H₂ / 50% CH₄ fuel blend at $\Phi = 0.80$. Similarly, $S_{T,l}$ increases from 4.8 ³⁶¹ to 5.5 ms⁻¹ by increasing the H₂ content from 70% to 100% in the CO blend 362 at $\Phi = 0.35$. By contrast, \overline{cu} $(\overline{c} = 0.50)$ decreases significantly with increasing $_{363}$ H₂ content for both fuel blends but at a different rate. The transition towards 364 counter-gradient transport is more rapid in CO mixtures where $\overline{cu}|(\overline{c} = 0.50)$ 365 reduces by 0.57 ms⁻¹ per unit increase of τS_L (5.7 × 10⁻³ ms⁻¹ per % H₂ ³⁶⁶ increase) in CO blends compared to $0.30 \text{ ms}^{-1} (8.7 \times 10^{-3} \text{ ms}^{-1})$ in CH₄. The ³⁶⁷ trends are consistent for all conditions with the acceleration across the flame $\frac{1}{100}$ front, as expected, stronger for mixtures with a higher $S_{T,l}$. With increasing 369 reactivity, $\overline{cu}|(\overline{c} = 0.50)$ moves towards counter-gradient transport (i.e. less ³⁷⁰ negative values), yet the transition is suppressed at the current turbulence 371 levels. The stronger inhibiting effect of CH₄ prevails for $\overline{cu}|(\overline{c} = 0.50)$.

 372 The turbulent flow field (i.e. U_b , u_{rms} and Re_t) is maintained constant, yet ³⁷³ the rate of the normal compressive strain increases towards the stagnation $_{374}$ plane. Selected data are plotted in terms of $S_{T,l}$ as function of the mean

Figure 8: Scalar flux (\overline{cu}) conditioned on the $\overline{c} = 0.5$ iso-contour as a function of τS_L for selected data. Left: H_2/CH_4 with $0\% - 50\% H_2$ at $\Phi = 0.80$; $0.50 \le \Phi \le 0.70$ at 60% $\rm H_2/40\% \rm CH_4; \, Right: \, H_2/CO \, with \, 70\% - 100\% \, H_2 \, at \, \Phi = 0.35; \, 0.40 \leq \Phi \leq 0.45 \, at \, 60\%$ $H₂/40\%$ CO.

375 normal compressive strain conditioned on the instantaneous flame front $(\overline{a_n},$ 376 see Sec. 3.2) in Fig. 9 to delineate the effect of H_2 addition and Φ separately. 377 An increase in H₂ content from 0 to 50% in CH₄ with $\Phi = 0.80$ (2.0 $\leq \tau S_L$) $\text{378} \quad (\text{ms}^{-1}) \leq 3.5$ results in a reduction of $\overline{a_n}$ acting on the flame front from -582 379 to -39.4 s⁻¹. This is attributed to the increasing detachment of the flame 380 front from the stagnation plane with $S_{T,l}$, i.e. from 4.4 to 5.5 ms⁻¹. The 381 corresponding scalar flux conditioned on the $\bar{c} = 0.50$ iso-contour reduces 382 from -0.53 to -0.09 ms^{-1} .

383 An increase in Φ (0.50 to 0.70) of the 60% H₂ / CH₄ mixture results in ³⁸⁴ a similar increase of $S_{T,l}$ (4.7 to 6.2 ms⁻¹) and τS_L (1.4 to 3.0 ms⁻¹) and 385 a weaker reduction in $\overline{a_n}$ (–493 to –232 s⁻¹) as well as $\overline{cu}|(\overline{c} = 0.5)$ (–0.41) ³⁸⁶ to -0.28 ms^{-1}). Hydrogen blending to CO (70 to 100% H₂) at $\Phi = 0.35$ 387 causes an increase of $S_{T,l}$ from 4.8 to 5.5 ms⁻¹ and a reduced mean normal 388 compressive strain from -473 to -231 s⁻¹. The scalar flux reduces towards 389 counter-gradient transport from -0.47 to -0.34 ms⁻¹. The increase in Φ ³⁹⁰ leads to an increase in $S_{T,l}$ from 5.5 to 5.9 ms⁻¹, $\overline{a_n}$ is reduced from -422 to ³⁹¹ -291 s⁻¹) and \overline{cu} $(\overline{c} = 0.5)$ from -0.48 to -0.37 ms⁻¹.

Figure 9: Turbulent burning velocity $(S_{T,l})$ vs the mean normal compressive strain acting on the flame front $(\overline{a_n})$. Left: H₂/CH₄ with $0\% \leq H_2 \leq 50\%$ at $\Phi = 0.80$; $0.50 \leq \Phi \leq 0.70$ at 40% CH₄; Right: H₂/CO with $70\% \leq H_2 \leq 100\%$ in CO at $\Phi = 0.35$; $0.40 \leq \Phi \leq 0.45$ at 40% CO. The symbol colour blue to red indicates $\overline{cu}|(\overline{c} = 0.50)$, i.e. gradient towards counter-gradient transport.

Figure 10: Scalar flux conditioned at the centre of the turbulent flame brush $(\overline{cu}|(\overline{c} = 0.5))$ as a function of mean normal compressive strain acting on the flame front $(\overline{a_n})$. Top left: $H₂$ / CH₄; Right: H₂ / CO.

³⁹² The correlation between the mean normal compressive strain and scalar $\frac{393}{100}$ flux conditioned on ($\bar{c} = 0.50$) is depicted in Fig. 10 for all data points. $_{394}$ The data are very well correlated with linear regression coefficients $(R²)$ of 395 0.93 and 0.95 for $\rm H_2$ / CH₄ and $\rm H_2$ / CO, respectively. Consequently, scalar ³⁹⁶ transport is strongly influenced by the total rate of strain strain.

³⁹⁷ 5. Conclusions

³⁹⁸ The present study has quantified the impact of the reactivity of hydro-³⁹⁹ gen enriched fuel blends, e.g. syngas, on turbulence–chemistry interactions ⁴⁰⁰ and scalar transport utilising a back-to-burnt opposed jet configuration for 401 aerodynamic flame stabilisation. The turbulent flow field $(Re_t = 314 \pm 19)$ μ_{102} and the burnt gas state temperature $(T_{HCP} = 1640 \pm 7 \text{ K})$ are maintained $\frac{403}{403}$ constant to delineate chemistry effects of binary H₂ / CH₄ and H₂ / CO fuel $_{404}$ blends. The H₂ content is varied from pure CH₄ to 100\% H₂ and from 30 ⁴⁰⁵ to 100% H² in CO blends. The equivalence ratio is adjusted between the ⁴⁰⁶ mixture specific lower limit of local flame extinction and the upper limit of ⁴⁰⁷ flashback. An increase in mixture reactivity results in a higher turbulent ⁴⁰⁸ burning velocity. Methane exhibits a stronger pronounced inhibiting effect $\frac{409}{409}$ on the hydrogen chemistry compared to carbon monoxide with the β scaling ⁴¹⁰ attenuating the differences. The KPP-type approach predicts the measured ⁴¹¹ $S_{T,l}$ within 21 \pm 9% for all self-sustained flames. A distinct transition to- $_{412}$ wards counter-gradient transport with increasing $S_{T,l}$ and thermal expansion ⁴¹³ ratio is established and influenced by the rate of compressive strain. Thus, $_{414}$ mixtures with a high $S_{T,l}$ detach further from the stagnation plane and an-⁴¹⁵ chor in low compressive strain regions where dilatation is more effective and ⁴¹⁶ lead to a transition towards counter-gradient transport. The transition is ⁴¹⁷ suppressed by the high turbulence levels. This is supported by a correlation ⁴¹⁸ coefficient $(R^2) > 0.93$ for the scalar flux conditioned on the centre of the 419 turbulent flame brush as a function of mean normal compressive strain $(\overline{a_n})$ 420 acting on the flame front. The correlation $S_{T,l} - \overline{a_n}$ also shows a reason-⁴²¹ able correlation. The results provide guidance to engine manufacturers and

 inform risk assessments as to the use of hydrogen enriched mixtures. The wide range of conditions further presents an excellent challenge for turbulent combustion models that aim to delineate the influence of fuel reactivity.

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Table 1: Data summary of H₂ / CH₄ and H₂/CO mixtures, where β is defined in Eq. (1), Φ is the equivalence ratio, ρ_r the density of the reactants (r) , ν_r the kinematic viscosity, Le_r the Lewis number, $\tau = (T_{ad} - T_r)/T_r$ the heat release parameter, $S_{L,0}$ and S_L the unstrained and strained laminar burning velocity, δ_f the laminar flame thickness and a_q the laminar rate of strain at extinction.

Mixture	β	Φ	ρ_r	$\nu_r \times 10^{-5}$	Le_r	τ	$S_{L,0}$	S_{L}	δ_f	\boldsymbol{a}_q
H_2 / CH_4	$\lceil - \rceil$	$[\hbox{-}]$	$\left[\mathrm{kg}/\mathrm{m}^3\right]$	$\rm [m^2/s]$	$\lceil - \rceil$	$\lceil - \rceil$	[m/s]	[m/s]	mm	$[s^{-1}]$
100%/0%	1.0	0.35	1.04	1.78	0.374	3.33	0.076	0.44	1.19	3600
90%/10%	0.69	0.35	1.08	1.71	0.438	3.22	0.032	0.33	2.59	2000
80%/20%	0.50	0.35	1.10	1.68	0.500	3.13	0.016	0.26	4.42	800
80%/20%	0.50	0.50	1.07	1.72	0.527	4.10	0.164	0.52	0.703	4000
70%/30%	0.37	0.50	1.09	1.68	0.585	4.04	0.121	0.39	0.821	2400
70%/30%	0.37	0.60	1.07	1.70	0.600	4.55	0.240	0.55	0.515	4300
60%/40%	0.27	0.50	1.10	1.66	0.641	3.99	0.096	0.35	0.971	1400
60%/40%	0.27	0.60	1.09	1.67	0.654	4.52	0.194	0.46	0.588	2900
60%/40%	0.27	0.70	1.08	1.69	0.666	4.93	0.308	0.61	0.436	4300
50%/50%	0.20	0.60	1.11	1.65	0.707	4.49	0.164	0.39	0.651	1900
$50\%/50\%$	0.20	0.70	1.09	1.66	0.717	4.91	0.263	0.52	0.480	3100
50%/50%	0.20	0.80	1.08	1.67	0.727	5.24	0.364	0.66	0.393	4000
40%/60%	0.14	0.70	1.11	1.64	0.768	4.89	0.231	0.46	0.517	2300
40%/60%	0.14	0.80	1.10	1.65	0.775	5.22	0.322	0.57	0.420	3000
20%/80%	0.06	0.80	1.12	1.61	0.869	5.19	0.267	0.46	0.468	1900
$0\% / 100\%$	0.00	0.80	1.14	1.58	0.962	5.09	0.228	0.39	0.511	1300
H_2 / CO										
100%/0%	1.0	0.35	1.04	1.78	0.374	3.34	0.077	0.44	1.19	3600
90%/10%	0.90	0.35	1.05	1.75	0.467	3.37	0.072	0.41	1.25	3500
80%/20%	0.80	0.35	1.07	1.73	0.554	3.40	0.069	0.38	1.31	2700
80%/20%	0.80	0.40	1.05	1.75	0.567	3.76	0.142	0.47	0.786	4200
70%/30%	0.70	0.35	1.08	1.71	0.635	3.44	0.065	0.34	1.37	1900
70%/30%	0.70	0.40	1.07	1.73	0.648	3.80	0.130	0.43	0.839	3200
60%/40%	0.60	0.40	1.08	1.70	0.722	3.83	0.120	0.39	0.895	2300
60%/40%	0.60	0.45	1.07	1.72	0.734	4.15	0.191	0.48	0.672	3500
50%/50%	0.50	0.40	1.10	1.68	0.788	3.87	0.110	0.35	0.955	1600
50%/50%	0.50	0.45	1.09	1.69	0.800	4.19	0.173	0.43	0.724	2500
50%/50%	0.50	0.50	1.08	1.70	0.811	4.48	0.246	0.52	0.596	3500
40%/60%	0.40	0.50	1.10	1.67	0.866	4.51	0.219	0.46	0.646	2500
40%/60%	0.40	0.60	1.09	1.69	0.883	4.97	0.366	0.63	0.508	4200
30%/70%	0.30	0.50	1.12	1.65	0.911	4.54	0.193	0.39	0.708	1700
30%/70%	0.30	0.60	1.11	1.66	0.924	5.00	0.319	0.56	0.556	2900

Table 2: Result summary for H₂ / CH₄ and H₂ / CO fuel mixtures, where Φ_{UN} is the upper nozzle equivalence ratio, Re_t the turbulent Reynolds number, Da the Damköhler number, u'_r the axial reactant velocity fluctuation at the nozzle exit, u'_l the axial velocity fluctuation at the leading edge ($\bar{c} = 0.02$ iso-contour), \hat{u}' the maximum axial velocity fluctuation in
the flame brush \bar{u} the mean velocity at the trailing edge ($\bar{c} = 0.05$ iso contour). $S_{\rm int}$ the the flame brush, $\overline{u_t}$ the mean velocity at the trailing edge ($\overline{c} = 0.95$ iso-contour), $S_{T,l}$ the measured turbulent burning velocity at the leading edge ($\bar{c} = 0.02$ iso-contour) and S_T^* is the KPP-type turbulent burning velocity).

Mixture	Φ_{UN}	Re_t	$_{Da}$	u_r'	u'_l	$\overline{\widehat{u}'}$	$\overline{u_t}$	$S_{T,l}$	S_T^*
H_2 / CH_4	$\left[-\right]$	$\lceil - \rceil$	$\lceil - \rceil$	[m/s]	[m/s]	[m/s]	[m/s]	[m/s]	$\left[\text{m/s}\right]$
$100\overline{\%}/0\%$	0.35	307	1.03	1.60	1.87	1.96	0.846	5.51	4.65
90%/10%	0.35	294	0.384	1.47	1.88	1.96	0.154	4.64	3.90
80%/20%	0.35	339	0.157	1.68	2.01	2.19	-0.0880	4.76	3.55
80%/20%	0.50	314	2.09	1.61	1.98	1.99	0.669	5.20	5.03
$70\%/30\%$	0.50	299	1.44	1.48	1.87	1.97	0.169	4.66	3.97
70%/30%	0.60	302	3.13	1.55	1.94	1.98	1.11	5.74	4.95
60%/40%	0.50	300	1.11	1.47	1.91	2.04	0.0056	4.71	3.74
60%/40%	0.60	320	2.22	1.63	1.96	1.98	0.338	5.06	4.41
60%/40%	0.70	334	3.74	1.71	2.07	2.13	0.545	6.22	5.37
50\%/50\%	0.60	315	1.76	1.49	1.93	2.01	-0.0114	4.56	3.89
$50\%/50\%$	0.70	310	3.20	1.59	2.01	2.04	0.585	5.25	4.69
$50\%/50\%$	0.80	302	5.11	1.48	1.98	1.98	1.58	5.47	5.27
$40\%/60\%$	0.70	321	2.56	1.58	2.00	2.02	0.124	5.52	4.28
40\%/60\%	0.80	317	3.96	1.52	1.93	2.00	0.722	5.28	4.66
20%/80%	0.80	327	2.83	1.56	2.03	2.09	0.164	5.04	4.16
0% /100%	0.80	324	2.27	1.51	$2.03\,$	2.17	-0.0145	4.41	3.64
CO H ₂									
$100\%/0\%$	0.35	307	1.03	1.60	1.87	1.96	0.846	5.51	4.85
90%/10%	0.35	289	0.985	1.48	1.93	2.01	0.413	5.53	4.42
80%/20%	0.35	306	0.833	1.54	1.93	1.96	0.266	5.23	4.08
80%/20%	0.40	303	1.70	1.52	1.93	1.97	0.713	5.43	4.56
70%/30%	0.35	307	0.719	1.50	1.96	1.99	0.0462	4.81	3.77
70%/30%	0.40	314	1.44	1.51	1.98	2.06	0.435	5.48	4.29
60%/40%	0.40	323	1.20	1.56	1.99	2.10	0.193	5.48	3.96
60\%/40\%	0.45	313	2.03	1.54	1.99	2.08	0.544	5.92	4.41
$50\%/50\%$	0.40	381	0.876	1.95	2.23	2.24	0.0958	5.03	4.02
$50\%/50\%$	0.50	358	2.16	1.88	2.11	2.13	1.67	5.07	4.68
40\%/60\%	0.50	347	1.85	1.81	1.97	2.08	1.56	5.27	4.04
$40\%/60\%$	0.60	338	3.29	1.75	2.03	2.04	3.33	5.64	4.89
30%/70%	0.50	342	1.48	1.72	2.03	2.05	0.465	4.88	3.71
30%/70%	0.60	332	2.76	1.66	1.89	1.97	1.47	5.08	4.24