

ESTIMATING THE SCALE OF SPACE RESOURCE UTILISATION (SRU) OPERATIONS TO SATISFY LUNAR OXYGEN DEMAND

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ABSTRACT

The production of oxygen from lunar regolith is analogous to metal production from ore in a terrestrial mine. The process flowsheets both include excavation, haulage and beneficiation of the regolith or ore to provide the feedstock for the chemical extraction of oxygen or metal. The production rate of oxygen depends on the mass rate of regolith treated and the efficiency of converting the regolith to oxygen. To date, the development of Space Resource Utilisation (SRU) has been concerned with the technological development of the process, particularly the excavation and oxygen extraction. However, the required operating mass rates of the mine operation and the oxygen extraction stage have not been considered in any great detail.

Previous estimates of mining scale for lunar oxygen production are reviewed, and converted to a comparable regolith mining rate of kg/h. Beneficiation of the regolith before oxygen extraction is considered, and the effects of pre-sizing and removal of a specific component, agglutinates, are considered. The oxygen yield and operational availability are also included. It is estimated that the minimum mining rate to produce 1000 kg of oxygen per annum is at least five times higher than previous estimates, 30 kg/h, for equivalent efficiency assumptions.

23 Monte-Carlo simulations were performed for statistical confidence in the estimates of the mining mass
24 rate and the required oxygen extraction feedstock rate. To be 95% confident that the 1000 kg/y O₂ will
25 be met, the designed mining rate should be at least 65 kg/h, and the beneficiated feedstock rate 16
26 kg/h.

27 This study has revised and increased the estimate of the lunar regolith mining scale required for the
28 production of a given amount of oxygen. It has also estimated the mass rate of feedstock required for
29 oxygen extraction, if the regolith is first beneficiated.

30 The findings have a significant impact on the practical implementation of lunar mining and oxygen
31 extraction, particularly the process design and whether the operation will be by batch- or continuous
32 processing. The mass scale and beneficiation approaches bring terrestrial mining and processing
33 concepts to SRU, and for the first time estimates the effect that regolith beneficiation and uncertainty
34 have on the estimated scale of both the mining and extraction operations.

35

36 *Keywords: Space resources, space mining, SRU, ISRU, lunar mining scale, Moon feedstock tonnage*

37

38 1 INTRODUCTION

39 Lunar SRU was proposed by Carr (1963) during the Apollo era as a way to reduce launch mass and the
40 dependency on Earth. Research in this field peaked first in the early 1970s at the height of the Apollo
41 programme, and then again in the early 2000s following the announcement of the (now defunct)
42 Constellation Program. With the recent advent of space privatisation and investment, and NASA's
43 commitment to return astronauts to the Moon by 2024 (NASA, 2019), SRU is once again coming to
44 the forefront of space sector research and development.

45 The commodities of interest are oxygen and water, both of which have numerous economic and
46 technological applications. In a recent and comprehensive report, Kornuta et al. (2019) estimated that
47 the demand for lunar water could be as much as 2450 t/y, including water required on the lunar surface
48 (150 t/y), for fuelling missions to Mars (180 t/y) and for transporting propellant to LEO (1880 t/y).
49 Furthermore, it is reported that 1000 kg O₂ will be required per year for life-support purposes on the
50 Moon (Lee, Oryshchyn et al. 2013). In order to design suitable technologies and processes for
51 producing a given target quantity of oxygen, the scale of the regolith mining and oxygen extraction
52 operations must be established.

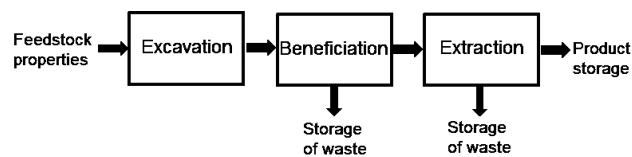
53 While the use- and business cases for lunar SRU have been well-considered, the scale of mining
54 operations required to produce the projected amount of oxygen (1000 kg/y, following Lee, Oryshchyn
55 et al., 2013) has received less attention. The scale of operations is critical as this defines the technology
56 and scheduling requirements of the whole process, for example whether to operate on a continuous
57 basis or a batch basis.

58 To estimate the scale to produce a specified target of 1000 kg/y of O₂, the efficiency of the reduction
59 reactor must be taken into account, in addition to possible upstream beneficiation of the regolith into a
60 more efficient oxygen extraction feedstock. These mass rate estimations are standard mass balancing
61 procedures carried out in terrestrial mining operations.

62 1.1 Mine Operations

63 The extraction of space resources follows the same basic flowsheet as that of terrestrial mineral
64 extraction. This is shown in Figure 1. In terrestrial mining, the operations are mining, beneficiation
65 (mineral processing) and smelting. Beneficiation is the process in which particles in a feedstock (ore or
66 regolith) are separated based on some material property, including size, density, conductivity,
67 magnetism and surface chemistry.

68



69

70 **Figure 1: Universal flowsheet for SRU processes, from Hadler et al. (2019)**

71 There are several, critical, differences between terrestrial operations and those for space resources,
72 namely the environmental differences (low gravity, vacuum and no available water for transport and
73 separation of particles), fault tolerance levels and system design. On Earth, processing operations are
74 overseen by teams of operators, to resolve frequent interruptions in operation caused by the handling
75 of large streams of slurry with variable feed properties. The flowsheets are typically complex, to enable
76 one, or several, products to be produced to a target specification. For space resources, reliability and
77 process simplicity must be taken into account, in addition to process optimisation.

78 1.2 Oxygen Production from Lunar Regolith

79 While over 20 technologies have been proposed for this step (Taylor and Carrier, 1993), the four main
80 contenders being considered by NASA and ESA are (in order of increasing oxygen yield): Hydrogen
81 reduction of ilmenite (<2% yield; Lee, 2013); carbothermal reduction of silicates and iron oxides
82 (~10% yield; Muscatello and Gustafson, 2010); molten regolith electrolysis (estimated to be 20-40%
83 yield, Schreiner, 2015); and, molten salt electrolysis, also known as the FFC Cambridge process (up to

84 100% recovery; Schwandt et al., 2012). Of these technologies, hydrogen reduction has received the
85 most attention, and its working principles are well-characterised.

86 The hydrogen reduction of regolith is primarily dependent on the ilmenite content (Schwandt et al.,
87 2012), according to the following reactions:



88 While the FeO component of ilmenite is preferentially reduced, oxygen can also be extracted (with
89 varying efficiencies) by the reduction of TiO₂, plagioclase, olivine and pyroclastic glasses (Chambers et
90 al., 1995).

91 Each hydrogen reduction reactor has been designed to operate with an ideal feed. There is, however,
92 comparatively little research on how these feeds would actually be produced in situ, and how this would
93 feed into the reactor. For example, a design patented by Gibson and Knudson (1996) requires a feed
94 comprising 80-90 wt% ilmenite and ferrous agglutinates, sized to 20-200 μm, in order to operate at the
95 highest oxygen production efficiency. While they do discuss the needs and requirements for magnetic,
96 electrostatic and size separation steps, such technology has yet to be developed for lunar applications.

97 **1.3 Regolith Variability**

98 Lunar regolith contains around 40% oxygen, however this is bound in minerals. Of the minerals on the
99 Moon, ilmenite (FeTiO₃) is one of the most attractive for SRU applications, as it is relatively abundant,
100 easily processed and has a high oxygen yield (Chambers et al., 1995). The maximum mass of O₂ that
101 can be produced for 1 kg of ilmenite is 105 g by stoichiometry; a theoretical maximum yield of 10.5%.

102 Two of the most significant uncertainties in the SRU flowsheet are the material properties and the
103 spatial variability of the regolith. According to Schreiner, Dominguez et al. (2016), this has impeded the
104 rigorous engineering design of reactors, transport mechanisms and excavation technologies. They

105 collated thermo-physical properties of regolith, reporting average contents of TiO_2 of 8.5% for high-Ti
106 Mare, 2.9% TiO_2 for low-Ti Mare and 0.5% for Highlands regolith. The Highlands regolith dominates
107 the lunar surface, accounting for 84% of the nearside and 99% of the farside (Heiken et al., 1991). The
108 bulk density of regolith lies between 1500 kg/m^3 for the top 15 cm to 1790 kg/m^3 at around 60 cm
109 depth (Schreiner, Dominguez et al., 2016). It should be noted that the TiO_2 exists typically in the
110 mineral form of ilmenite (FeTiO_3), however titanium dioxide content is often cited in the literature as
111 an analogue.

112 Lunar simulant JSC-1A has a TiO_2 content of approximately 1-2% (Schwandt et al., 2012). Clark et al.
113 (2009) describe a single field test that treated a JSC-1A sample in the PILOT hydrogen reduction
114 system. The single successful trial produced close to 1% by mass water. This translates into a
115 conversion of regolith to oxygen of 0.88%, assuming a 100% conversion and gas recovery during the
116 subsequent water electrolysis. It is of interest that the experimental system was not straightforward and
117 suffered numerous practical complications. It appears that at least some of these issues can be
118 attributed to the large proportion of very fine particles in the feedstock.

119 The variability in the TiO_2 content of 17 lunar maria was reported by Sato et al., (2017), where the
120 average was 3.9% TiO_2 . The highest TiO_2 content was reported for Mare Tranquillitatis at 12.6%, with
121 an average of 6.4% TiO_2 . The authors highlighted that many of the maria had TiO_2 levels below the
122 detection limit ($\sim 2\%$), and also that there was great variability. It is suggested that older maria have
123 greater variability.

124 A comprehensive petrological and geochemical study of 12 particles from the Apollo 12 soil samples is
125 presented in Alexander et al. (2014). The samples are all of low-Ti basalts (i.e. 1-6% TiO_2), with an
126 ilmenite content ranging from 0.2% to 7.3%.

127 The variability of the feedstock, and the subsequent effect on processing, is a feature of both terrestrial
128 and space resource extraction. The estimation of mining and extraction rates can therefore be
129 performed in two ways. First, by using the average properties of the regolith, a single estimate is

130 obtained, and which can be altered by changing the process variables, such as particle sizing and
131 decontamination to assess the range. Second, this can be repeated by taking also the regolith property
132 variance into account and performing numerous Monte-Carlo simulations, where many estimates are
133 made by repeatedly drawing from the property and variable distributions. This yields a confidence
134 estimate for the predictions, which is useful for mission-critical production of oxygen. Both these
135 approaches will be used here.

136 **1.4 Objectives**

137 While a great deal of research effort has been allocated to the technologies that produce oxygen in situ,
138 the feed preparation stage has generally been oversimplified or overlooked entirely. The preparation of
139 a suitable feedstock for a given reduction process, hydrogen or otherwise, will directly impact the
140 design of the reduction technology, and will govern the scale of the mining operation itself.

141 Here, we consider two lunar oxygen production facilities, each capable of generating 1000 kg/y of
142 oxygen via a hydrogen reduction process. The first uses a regolith run-of-mine feedstock that has been
143 well characterised, and the second uses a variable feedstock with regolith from various locations. For
144 each of these facilities, the impact of feed preparation, namely sizing and decontamination, on the
145 overall scale of the mine are evaluated.

146 **2 METHODOLOGY**

147 The estimation of the scale of a mine depends on a number of variables. For lunar applications, the
148 estimation equations are most appropriately scaled to the mass rate of oxygen required.

149 The fraction of time that the complete oxygen production operation is in production is called the
150 *operational availability* and must be taken into account when estimating the regolith mining mass rate
151 required and the feedstock mass rate required to the oxygen extraction process. Operational availability
152 will depend on, for example, the availability of energy, light, temperature limitations and shift working
153 practices.

154 The oxygen extraction process treats a feedstock. This feedstock is the product from the beneficiation
155 stage (Figure 1). If there is no beneficiation stage, the oxygen extraction feedstock is the same as the
156 mined regolith.

157 The mass rate of feedstock required (kg feedstock/h) for oxygen extraction can be estimated from the
158 mass rate of oxygen required (kg oxygen/h), divided by the yield of oxygen from the feedstock (kg
159 oxygen/kg feedstock) and the operational availability:

$$\text{Mass rate of feedstock} = \frac{\text{Mass rate of Oxygen}}{(\text{Oxygen Yield} * \text{Operational Availability})} \quad (3)$$

160 The mined regolith may be beneficiated, for example, to remove first particles that are too fine and too
161 coarse, and subsequently, specific deleterious components such as agglutinates. The removed material is
162 the regolith waste. The mass rate of mined regolith required to meet the required feedstock mass rate is
163 determined by the fraction of the mined regolith that is not removed by beneficiation.

$$\text{Mass rate of mined regolith} = \frac{\text{Mass rate of feedstock}}{(1 - \text{Mass fraction of regolith waste})} \quad (4)$$

164 These two equations can be extended to include any beneficiation processes, for example that may be
165 mineral specific, or using different operational availabilities for the mining and beneficiation and
166 oxygen extraction stages.

167 In order to estimate the mining scale for lunar oxygen production from regolith, the efficiency of the
168 reactor and the resultant yield of oxygen must be considered, but also the effect of beneficiation on the
169 mining rate required, the oxygen yield, and the operational availability.

170 Previous studies have addressed several of these aspects in their mine scale estimations (for example
171 Linne et al., 2015 and Williams et al., 1977). Linne et al. (2015) consider that 1000 kg of oxygen will be
172 required annually in an early outpost for life support closure only (as did Lee, Oryshchyn et al. 2013),
173 and as much as 10 000 kg oxygen per year to supply a full scale, permanent site.

174 Linne et al. (2015) estimate the mining scale required based on 1000 kg/y of oxygen and assume that
175 the full mined regolith can be used in a hydrogen reduction process, as Clark et al. (2009) tested
176 experimentally on JSC-1A. As noted, this may cause significant operational problems and system
177 reliability and efficiency are likely to improve with a feedstock from which has removed both the very
178 fine and coarse particles. Further, Linne et al. (2015) assume that the oxygen yield from maria is 4%.
179 The O₂ yield is a most important variable in determining the mine scale, as shown in Equation 3.
180 Considering the average mare TiO₂ content was found to be 3.9% (Sato et al., 2017), 4% O₂ yield is a
181 relatively high estimate (Chambers et al., 1995). Linne et al. (2015) also assume a 70% available
182 operation time, based on a solar-power-driven outpost near the southern poles with up to 90%
183 daylight. Using 4% oxygen yield and 70% operational availability, this yields a required feedstock mass
184 rate of 4.1 kg/h (Equation 3). Since un-beneficiated regolith is used as the feedstock, the mined regolith
185 mass rate is also 4.1 kg/h (Equation 4).

186 Williams et al. (1977) quantify the efficiencies required of theoretical beneficiation stages by specifying
187 the mineralogical content (% TiO₂) of both the mined regolith (13% TiO₂) and that of the oxygen
188 extraction feedstock (48.1% TiO₂). Williams et al. (1977) estimate a working year of 3700 hours (42.2%
189 operational availability). They assume an optimistic 8% oxygen yield, therefore requiring a feedstock
190 mass rate of only 3.4 kg/h to produce 1000 kg O₂ annually (Equation 3). A mass balance shows that to
191 increase the TiO₂ concentration 3.7 times from mined regolith to feedstock requires 73% of the mined
192 regolith tonnage to be rejected as waste. Their predicted required mined regolith rate is therefore 12.6
193 kg/h (Equation 4).

194 As noted previously, the mined regolith and oxygen extraction feedstock mass rate estimates here are
195 all scaled to the same required oxygen mass rate of 1000 kg/y or 0.114 kg oxygen/h.

196 The estimated 12.6 kg/h regolith mined mass rate of Williams et al. (1977) is almost double the 6.8
197 kg/h of Linne et al. (2015), at an equivalent 42.2% operational availability. However, as a result of the
198 increase in oxygen yield from regolith beneficiation, the Williams et al. (1977) oxygen extraction

199 feedstock mass rate is only 3.4 kg/h. This has tremendous implications for the design of the complete
200 oxygen production flowsheet, from mine to oxygen product. In particular, halving the required furnace
201 capacity has significant energy and mass benefits, but these may be offset by the large increase in the
202 required mine scale.

203 It is evident that the impact of regolith beneficiation on oxygen production must be examined in
204 greater detail.

205 This study estimates the oxygen extraction and regolith mining scales required to produce 1000 kg/y of
206 oxygen, considering:

- 207 • Regolith particle sizing, and the effect of removing the fine and coarse materials that are likely
208 to cause blockages, material handling issues and dust.
- 209 • Regolith decontamination.
- 210 • Oxygen yield from regolith (conversion efficiency).
- 211 • Operational availability.

212

213 **3 MINE SCALE ESTIMATION: SPECIFIC REGOLITH COMPOSITION**

214 The first estimations establish the mine scale for a given specified feedstock that takes into account
215 variation in the beneficiation stage and oxygen yields.

216 As an exemplar of a mined lunar material that could be used as a feedstock, consider sample 71061,1,
217 which is regarded as a typical Apollo 17 mare soil (Heiken, 1975). The soil property data were collated
218 into a number of size classes and component classes to highlight the key groups that will be considered.

219 It is highly likely that some regolith pre-treatment will be required to reduce operational problems
220 primarily associated with dust and abrasiveness. These pre-treatment (beneficiation) stages will reduce
221 the mass of feedstock available for oxygen production, and therefore increase the required mine scale.

222

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Table 1: Collated properties of lunar sample 71061,1, data from Heiken (1975), converted from volume % to weight %. Note that

224

material coarser than 10 mm constitutes 9.4% of the sample mass, all of which is assumed to be basaltic.

		SIZE (µm)	<45	45-90	90-1000	1000-10000
Total Mass Fraction			30.2%	11.4%	26.2%	22.8%
Mass Fraction (in size range)	Agglutinates		12.0%	9.9%	6.9%	0.0%
	Glass		0.0%	23.8%	13.3%	0.0%
	Basalts		0.0%	13.0%	41.5%	100.0%
	Breccia		0.0%	3.5%	6.9%	0.0%
	Other minerals		0.0%	39.6%	26.4%	0.0%
	Ilmenite		0.0%	8.6%	4.7%	0.0%
	Other material		88.0%	1.6%	0.3%	0.0%

225

3.1 Particle Size

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It will be necessary to remove material that is either too coarse or too fine, to reduce possible

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blockages, equipment damage and dust issues during materials handling and oxygen extraction. The

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regolith material greater than 1 mm constitutes 32.2% (22.8% is 1 mm to 10 mm and 9.4% greater than

229

10 mm) of the total mass (Table 1). It is considered difficult to size efficiently dry particles below 90 µm

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(Kelley, 2007), and this is considered the minimum size that can be processed. The material finer than

231

90 µm constitutes 41.6% of the total mass.

232

From Table 1, it can be seen that removing both the coarse (>1 mm) and fine (<90 µm) fractions

233

eliminates 73.8% of the mined regolith as waste and leaves only 26.2% of the mass for further

234

processing. The process step of sizing alone will increase the mine scale almost fourfold from mine

235

scale estimates that propose utilising the full regolith size distribution.

236

3.2 Decontamination

237

Agglutinates are considered the most abrasive part of the feedstock and should not be handled

238

extensively, as it will rapidly cause extensive equipment wear. Furthermore, the glass fraction is likely to

239 cause issues in the high-temperature furnace due to sintering (Allen et al., 1994). It is considered
 240 possible theoretically to remove both the agglutinates and glass particle types selectively by using shape-
 241 selective techniques, however these are yet to be developed.

242 Using the data in Table 1, agglutinate decontamination will reduce by a further 6.9% the sized regolith
 243 mass available for oxygen extraction. Pre-sizing the mined material and then removing the agglutinates
 244 reduces the proportion of mined regolith available for oxygen production to 24.4% of the total mined
 245 mass. Removing also the glass fraction (13.3%) will reduce the useable part of the regolith to 20.9% of
 246 the total mined regolith.

247 An additional benefit of sizing and removing both glass and agglutinates is that the ilmenite content is
 248 increased from 2.2% in the full size distribution up to 5.9% in the remaining sized and decontaminated
 249 fraction. The effect on the mass fraction remaining and the ilmenite proportion can be estimated by
 250 assuming a perfect separator that removes only non-ilmenite minerals, as shown in Table 2.

251 **Table 2: The effect of increasing the ilmenite content from 4% to higher values on the mass fraction of mined regolith available**
 252 **for oxygen extraction by removing additional constituents from the 90-1000 µm fraction.**

Components Removed	Ilmenite content	Remaining Mass %
Agglutinates	5.0%	26.0%
Glass + Agglutinates	5.9%	20.9%
Glass, Agglutinates + Basalts	12.2%	10.0%
Glass, Agglutinates, Basalts + Breccias	14.9%	8.2%

253 It can be noted that the effect of selective mineral upgrading on the mine scale is significant, and that
 254 even a modest doubling of the proportion of ilmenite in the feedstock for chemical extraction (from
 255 5.9% to 12.2%, for example) will halve the mass remaining as feedstock (from 20.9% to 10.0%) and
 256 therefore double the mine regolith mass rate required. The impact of the ilmenite content on the
 257 oxygen yield must therefore be very significant to offset the increase in the amount of regolith that will
 258 be required.

259 We will not consider further in this paper the effect on the mine scale of selectively upgrading specific
260 minerals.

261 **3.3 Operational Availability**

262 As mentioned, Linne et al. (2015) estimate and use a 70% operational availability of the mining and
263 extraction operations. They consider the use of solar energy that is supplied from an outpost located
264 near the southern pole with up to 90% daylight. This places a significant constraint on the mine
265 location. Williams et al. (1977) estimate a 3700 h work year based on the lunar day/night cycle, a loss
266 of two shifts in each lunar day, and 7.5 production hours per 8 h work period. This corresponds to
267 42.2% operational availability and has been considered as the baseline in this study.

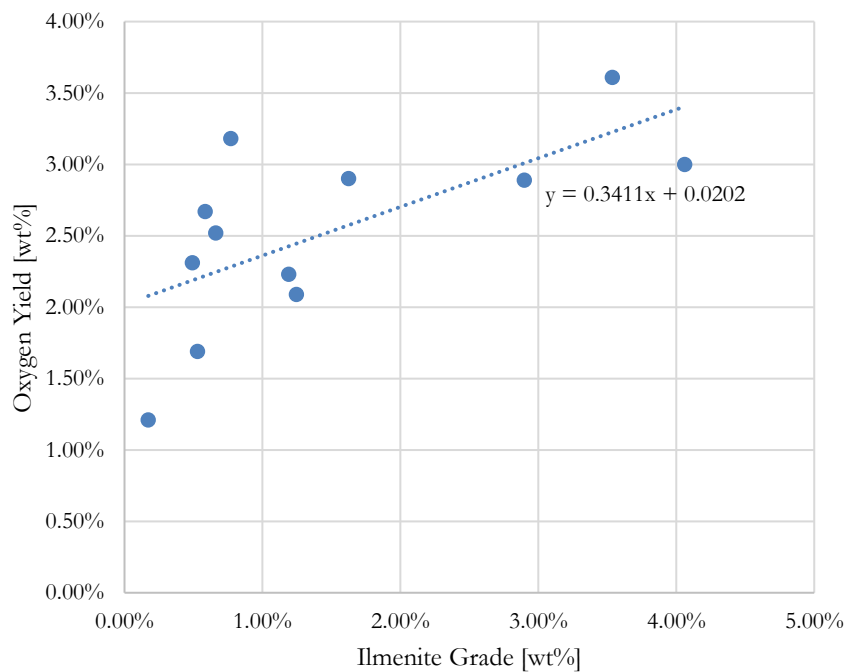
268 **3.4 Mine Scale**

269 NASA's ROxygen demonstration reactor, for example, employs hydrogen reduction. Experiments on
270 small-scale reduction systems showed the yield of O₂ produced from lunar regolith simulant to be 1-2%
271 (Lee, Oryshchyn et al., 2013). The authors noted that at temperatures of over 800 °C, clumping of
272 regolith particles was observed, which may result in improper mixing and sintering, fouling permanently
273 the reactor. In larger-scale experiments, the O₂ yield was 0.2-0.5% by mass; the decrease in yield was
274 attributed to issues associated with scaling the benchtop design, in particular with the heat transfer to
275 the reactor.

276 As noted by Linne et al. (2015), maximum yields of 2% to 4% oxygen mass to regolith mass have been
277 demonstrated for high-iron mare soils (McKay and Allen, 1996). The O₂ yield given a particular grade
278 of ilmenite in the feed is arguably the greatest unknown in the reduction process, thus a range of values
279 are considered here.

280 The experimental hydrogen reduction results of Allen et al. (1994) for various lunar soil samples were
281 published originally as the weight-% of Fe²⁺ and the oxygen yield. The results include the allocation of

282 Fe²⁺ to different feed constituents, including ilmenite. Figure 2 provides an interpretation of those data
283 in terms of ilmenite grade versus oxygen yield.
284 It has been assumed that the data follow a linear trend for total oxygen production versus grade, as was
285 used by Chambers et al. (1995), and is shown on Figure 2. There is significant scatter, however
286 additional, comparable data of oxygen yield versus ilmenite grade for the hydrogen reduction of lunar
287 regolith (or simulants) are not readily available.



288

289 **Figure 2: Oxygen yield from a given grade of ilmenite in a hydrogen reduction cell. Adapted from Allen et al. (1996).**

290 Table 3 presents the expected mine scale based on the data and equation derived from the experimental
291 work of Allen et al. (1996, Figure 2) and the degree of beneficiation as presented in Table 2. While this
292 method is approximate, this analysis does highlight the appreciable impact of O₂ yield on the capacity
293 and throughput of the reduction technology, and the scale of the required mining operation.

294

295 **Table 3: Analysis of mine scale for an annual output of 1000 kg O₂ as a function of ilmenite grade and oxygen yield, showing**
 296 **also the feedstock rate to the reduction process. Note that following the ‘As Mined’ data, the remaining data assumes the feed**
 297 **has also been sized to the 90-1000 µm fraction.**

			Linear Approximation from Allen et al. (1996)			
Components Removed	Ilmenite Grade [wt%]	% of Mined Material Remaining after Beneficiation	Oxygen Yield [wt%]	Total Regolith Required [t/year]	Mining rate required [kg/h]	Reduction process rate [kg/h]
As Mined (no sizing)	2.2%	100.0%	2.77%	36	9.8	9.8
Agglutinates	5.0%	26.0%	3.73%	103	27.9	7.3
Glass + Agglutinates	5.9%	20.9%	4.03%	119	32.1	6.7
Glass, Agglutinates + Basalts	12.2%	10.0%	6.18%	162	43.7	4.4
Glass, Agglutinates, Basalts + Breccias	14.9%	8.2%	7.10%	172	46.4	3.8

298

299 It is notable from Table 3 that beneficiation of the regolith to increase the ilmenite content increases
 300 significantly the mined tonnage rate required, but reduces also the feedstock mass rate for the reduction
 301 process. This is an important trade-off that has not been considered in detail previously. An oxygen-
 302 producing reduction furnace treating less than half a kilogram per hour of high-grade, sized feedstock
 303 will be smaller, lighter and less energy intensive than one treating a kilogram per hour of unsized
 304 regolith. The associated mining and beneficiation processes must handle a five time greater tonnage
 305 rate, although probably at a lower energy intensity and technical complexity. This is a complex technical
 306 and economic trade-off.

307 The mining rates estimated in Table 3 can be compared with previous studies. Linne et al. (2015)
 308 estimate a mining rate of 6.8 kg/h at 42.2% operational availability and 4% oxygen yield. The mine

309 scale required when first removing the glass and agglutinates is almost five times higher (32.1 kg/h),
310 while the oxygen extraction feedstock rate is, of course, the same.

311 Williams et al. (1977) estimate a mining rate of 12.6 kg/h at 42.2% operational availability and 8%
312 oxygen yield. To upgrade the TiO₂ fraction, they reject 73% of the mined material as waste, an amount
313 comparable to sizing the regolith and removing the agglutinate fraction. They assumed, however, an
314 8% yield from the hydrogen reduction step. With a lower O₂ yield (3.73%), their mining rate is less than
315 half the estimate (27.9 kg/h) in Table 3.

316 It is of interest that the required mining mass rate increases linearly, while the required oxygen
317 production feedstock mass rate decreases linearly (Figure 3), since the O₂ yield has been linearly
318 correlated to the feedstock TiO₂ content. This implies that reducing the oxygen production feedstock
319 mass rate by increasing the feedstock TiO₂ content will also increase linearly the required mining rate.
320 This allows optimisation of the power, mass and design of the oxygen extraction flowsheet.

321 It is evident from these estimates that the required lunar mining mass rate, for even limited oxygen
322 production, may be significant. The mine mass rate depends on the combination of assumptions made
323 about the regolith size range used, any decontamination or selective upgrading and the operational
324 availability. There is also a balance between the oxygen extraction feedstock mass rate and the mining
325 mass rate that allows optimisation.

326 In the following section, a statistical estimation will be used to estimate the confidence levels in the
327 required feedstock rate and mining rate to produce 1000 kg/y oxygen.

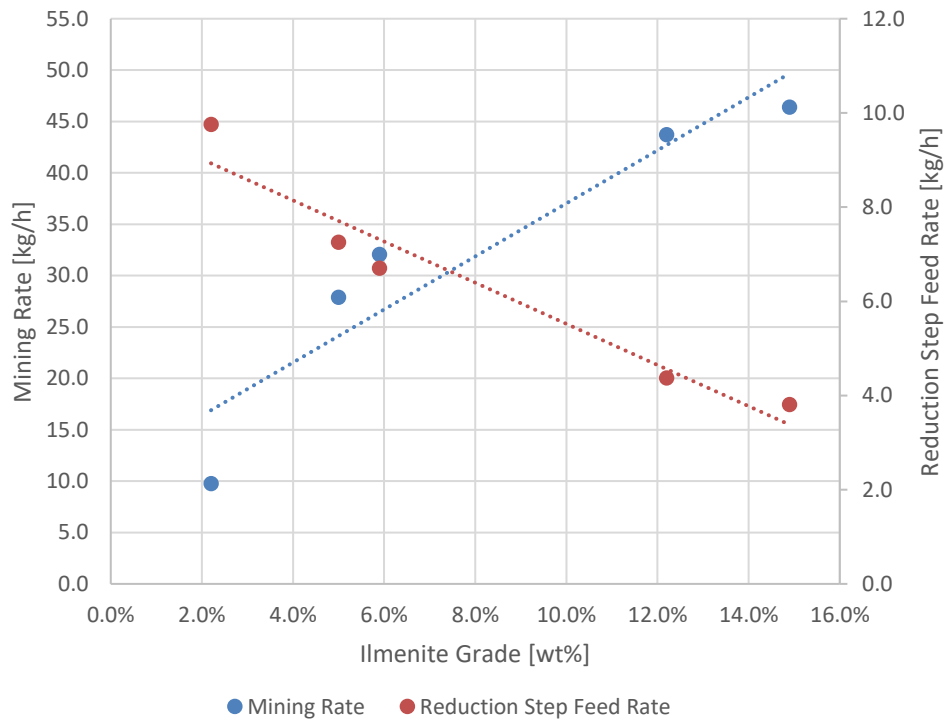


Figure 3: Regolith mining mass rate and reduction step feed rate versus ilmenite grade.

4 MINE SCALE ESTIMATION: VARIABLE REGOLITH COMPOSITION AND OPERATING CONDITIONS

In the calculations above, the very specific properties of lunar regolith sample 71061,1 were used to estimate the mine scale. The size distribution and proportion of agglutinates varies across the Moon, however. In order to estimate more comprehensively the range of mine scales that are likely, Monte-Carlo simulations were performed, using a range of values and their distributions.

Monte-Carlo simulations make repeated, random draws from known distributions of variables and combine them to make a prediction. This is repeated a large number of times to generate a probability distribution of the likely outcome. The results can be interpreted as the confidence that can be placed in the predicted value.

340 **4.1 Particle Size**

341 The size range between 90 μm and 1 mm is well characterised for many lunar samples. It is also
342 considered to be a range likely to reduce operational problems caused by very fine or coarse regolith.

343 The mass fraction of regolith between 90 μm and 1 mm was estimated from distributions compiled by
344 Phinney et al. (1977), and Heiken et al. (1991) for the Apollo samples. They note that some statistical
345 descriptions can be made, and that most mean grain sizes fall between 45 and 100 μm .

346 Interpolating from the graphical lunar soil size distributions reported (Phinney et al., 1977), the fraction
347 of regolith in the 90 μm to 1 mm size range varies from approximately 36% to 48%, or an average of
348 42% \pm 6%. For the Monte-Carlo simulations, it is assumed that the distribution of the mass fraction of
349 regolith between 90 μm and 1 mm is normal and that the 6% mass range describes \pm 2 standard
350 deviations (encompassing 95% of the sample means).

351 **4.2 Agglutinate Decontamination**

352 Heiken et al. (1991, Table 7.2) report the volume percent of agglutinates for the 90 μm to 1 mm
353 material for fourteen lunar samples, representative of each Apollo mission. The data was collated and
354 the average and standard deviation were calculated. Since the density of each component reported was
355 not known, the mass and volume percentages were considered equivalent.

356 An average of 66.6% of the total mass of the 90 μm to 1 mm fraction remained after agglutinate
357 decontamination, with a standard deviation of 12.6%. The distribution of values is assumed to be
358 lognormal to eliminate negative values.

359 **4.3 Operational Availability**

360 The operational availability estimate of 42.2% by Williams et al. (1977) was used as the mean of a
361 lognormal distribution. Without any further knowledge of the lunar operating conditions, a standard
362 deviation of 2.5% was assumed.

363 4.4 Oxygen Yield

364 The data from Allen et al. (1996) were used as the basis for the oxygen yield. The data in Figure 2 and
365 the linear regression indicate a relatively weak dependence of oxygen yield on ilmenite content. For the
366 Monte-Carlo simulations the ilmenite content is not considered as a variable, and the average and
367 standard deviation of the oxygen yield are from the experimental values reported by Allen et al. (1996).
368 The mass of oxygen per mass of regolith into the reduction process was therefore assumed to have a
369 mean of 2.5% and a standard deviation of 0.7%, lognormally distributed to eliminate negative yields.
370 Table 4 summarises the assumptions made for all the variables considered in the Monte-Carlo
371 simulations.

372 **Table 4: Summary of Monte-Carlo variables and distributions.**

Variable	Average	Standard Deviation	Distribution	Reference
Size mass fraction (90-1000 μm)	42.0%	3.0%	Normal	Clark et al. (2009)
Fraction post agglutinate removal	66.6%	12.6%	Lognormal	Heiken et al. (1991, Table 7.2)
Oxygen yield	2.5%	0.7%	Lognormal	Allen et al. (1996)
Operational availability	42.2%	2.5% (approx.)	Lognormal	Williams et al. (1977)

373 4.5 Mine Scale

374 The mine scale estimated directly from the mean values of all the variables in Table 4 yields an estimate
375 of approximately 40 kg/h of regolith to be mined to produce 1000 kg/y O₂. This corresponds well with
376 the regolith mass rate estimated for lunar sample 71061,1 above, which yielded an estimate of 57.5
377 kg/h, for a slightly lower (2%) oxygen yield.

378 The Monte-Carlo simulations were run 10000 times. First, each variable was varied individually, to
379 assess the magnitude of its effect and to determine which variable had the largest influence on the
380 range of estimates.

381 A simulation output is shown in Figure 4Figure 5 and Figure 5, where all the four variables were
 382 randomly sampled simultaneously. The figures show the results in two ways, as fractional and
 383 cumulative probabilities. The fractional probability shows the shape of the probability distribution of
 384 the predicted regolith mining rate required (kg/h). The most likely rate is 40 kg/h, which corresponds
 385 to using the means of all the variables. There is a long tail of predictions of high mining rates. Figure 5
 386 shows the confidence that can be placed in the predictions. There is only a 47% confidence that a 40
 387 kg/h mining rate will produce 1000 kg per year of oxygen, but a 90% confidence that a regolith mining
 388 rate of 63.9 kg/h will deliver the required O₂ rate.

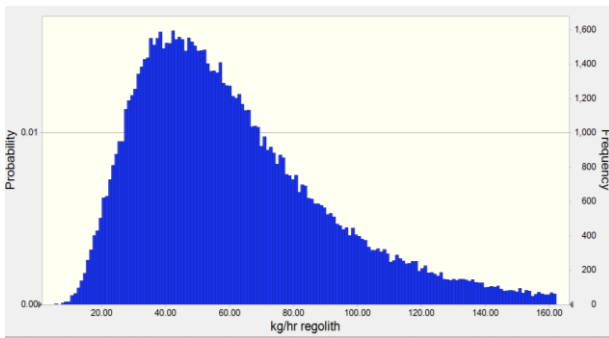


Figure 4: The range of estimates of mining scale for varied regolith properties, operational availability and O₂ yields

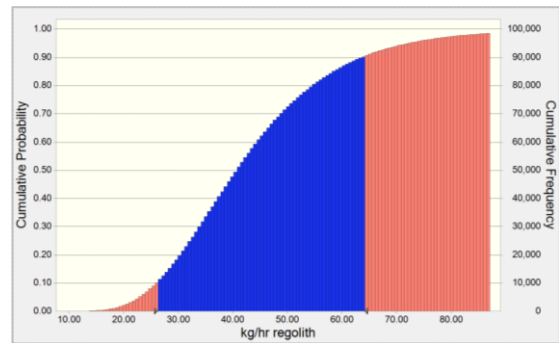


Figure 5: The mining scale for varied regolith properties, operational availability and O₂ yields as a cumulative frequency of estimates

389 The results taken from the simulation of each variable individually, and all variables simultaneously are
 390 summarised in Table 5.

391 Table 5: Summary of Monte-Carlo results for the variables in Table 4 individually and combined, showing the cumulative
 392 fraction of estimates.

Variable	Confidence in Estimation of Mining Rate (kg/h)			
	10%	25%	75%	90%
Size mass fraction (90-1000 μm)	30.9	34.7	44.7	50.0
Fraction post agglutinate removal	35.5	36.9	40.6	42.6
Oxygen yield	28.3	33.4	48.4	57.1
Operational availability	35.9	37.2	40.4	41.8

All variables simultaneously	26.5	32.6	51.8	63.9
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393 It can be observed that the interquartile range (25% to 75%) is widest for oxygen yield, and shows that
 394 gaining confidence in technology to produce, consistently, high yields of oxygen will significantly lower
 395 and narrow the mine scale estimates. Sizing, agglutinate removal and operational availability produce
 396 similar interquartile estimates. This highlights the importance of quantifying the performance of the
 397 hydrogen reduction process, and its performance variability to varying feedstock properties.

398 When designing a lunar oxygen production facility on which will depend the survival and transport of
 399 humans, a high confidence level of adequate scale is essential. For a 90% confidence that the mine rate
 400 will be adequate, a mine must be designed to produce at least 63.9 kg/h of regolith. This is almost
 401 double the estimate not taking the variance into account. For a 95% confidence, this increases to 72.6
 402 kg/h. These Monte-Carlo simulation estimates are orders of magnitude greater than the bare minimum
 403 that is obtained from single point calculations without taking onto account the variance.

404 **4.6 Hydrogen Reduction Scale**

405 The scale of the reduction operation is also required to design the complete oxygen production
 406 flowsheet from mine to product. The average size distribution and agglutinate content for a range of
 407 lunar samples was used to calculate the mine rate. Comparable data for ilmenite content was not
 408 available to relate it to the oxygen yield and the latter was treated as an independent variable in the
 409 Monte-Carlo simulations to estimate the scale required of the reduction stage.

410 The average scale of the reduction operation to produce 1000 kg/y O₂ can be estimated from the mean
 411 values of the oxygen yield and the operational availability as 10.8 kg/h. This is in the range of values
 412 estimated for lunar sample 71061 in Table 3, but here not taking into account the effect of ilmenite
 413 concentration on the oxygen yield.

414 The Monte-Carlo simulation was run 10000 times. First, the operational availability and oxygen yield
 415 were varied individually to assess the magnitude of their effects and to determine which of these two

416 variables had the largest influence on the range of estimates. Thereafter, they were varied
417 simultaneously.

418 The results taken from the simulation of each variable individually and both simultaneously are
419 summarised in Table 6.

420 **Table 6: Summary of Monte-Carlo results for the variables in Table 4 individually and combined, showing the cumulative**
421 **fraction of estimates.**

Variable	Confidence in Estimate of Reduction Rate (kg/h)			
	10%	25%	75%	90%
Oxygen yield	7.9	9.3	13.5	16.0
Operational availability	10.1	10.4	11.3	11.7
Both variables	7.8	9.3	13.6	16.1

422 It can be observed that the interquartile range (25% to 75%) is widest for oxygen yield, and shows that
423 gaining confidence in technology to produce, consistently, high yields of oxygen will significantly lower
424 and narrow the reduction process scale estimates.

425 From the Monte-Carlo simulations under the assumptions made, there is a 90% confidence that a
426 reduction rate of 16.1 kg/hr of feedstock, produced from sized and decontaminated regolith, will
427 deliver the annual requirement of 1000 kg of O₂. This increases to 17.9 kg/h for a 95% confidence
428 level.

429 **5 IMPACT ON SRU FLOWSHEET**

430 The mining rate estimates are significant. The estimated minimum lunar mining mass rate of 30 kg/h
431 will require substantial infrastructure and logistics. If the range of possible values is considered, this rate
432 is likely to be more than double to have sufficient confidence in the estimate, at least 65 kg/h.

433 The required mining mass rate determines whether the mineral sizing and decontamination can be
434 performed in batches, or as a continuous process. This has very significant design implications. A batch
435 process design may allow integration of mining and sizing equipment, so that the majority of the

436 unwanted material is rejected at the mining stage. This will reduce significantly the tonnages that must
437 be transported from the mine to the centralised beneficiation and extraction points. Continuous
438 operation may require satellite operations to prepare the feedstock, which is then transported, possibly
439 to a single extraction point. In both cases, operation may have to pause during the lunar night.

440 The specific details of the process engineering clearly depends significantly on the regolith tonnage rate
441 that has to be mined, sized, beneficiated and transported.

442 Considering the reduction stage to produce oxygen, there is evidence that the oxygen yield is enhanced
443 when the ilmenite concentration in the feedstock is increased. This is an effect of sizing and agglutinate
444 decontamination of the mined regolith, but can also be further enhanced by mineral beneficiation. This
445 reduces significantly the mass scale of the reduction process, but increase the mined tonnage required.
446 This aspect of the overall oxygen production flowsheet will require significant further investigation, as
447 the relative scales will require optimisation to determine the most efficient combination.

448 **6 CONCLUSIONS**

449 This study has produced a range of possible values for the lunar regolith mining rate required.
450 Estimates were based on data from lunar samples, and reasonable assumptions from terrestrial mining
451 and the hydrogen reduction extraction process.

452 The estimated minimum mining rate required is approximately 30 kg/h, which is as much as five-fold
453 higher than previous estimates. Monte-Carlo simulations were performed to take into account
454 uncertainty in the values. These indicated that to have 95% confidence that adequate O₂ will be
455 produced, the required mining mass rate is likely to be as high as 65 kg/h. The mining mass rate is
456 important for process design, as at some point the beneficiation process may be more efficient if
457 designed as a continuous rather than batch operation.

458 The impact of ilmenite content on oxygen yield is significant, as it determines both the increase in the
459 mine tonnage required and the decrease in tonnage to be treated during the reduction stage. More

460 accurate estimates can be made once there is greater confidence in the oxygen yields from various
461 beneficiated feedstocks, as well as the requirements of alternative reduction processes. This is probably
462 the source of the greatest uncertainty, and which can only be reduced by further experimentation.
463 In the mining scale estimate above, no allowance has been made for activities such as preparing the
464 mine area by removing boulders, smoothing out small craters and preparing roadways. It is estimated
465 that such activity could add a significant burden to the mining tonnages that must be planned for.

466

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